∂ OPEN ACCESS JOURNAL OF ADVANCED DIELECTRICS Vol. 12, No. 6 (2022) 2241003 (6 pages) © The Author(s) DOI: 10.1142/S2010135X2241003X





Effect of Li₂CO₃ addition on structural and electrical properties of 0.7BiFeO₃-0.3BaTiO₃ piezoelectric ceramic

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Received 18 August 2022; Revised 4 September 2022; Accepted 8 September 2022; Published 19 October 2022

In this work, Li_2CO_3 was added into 0.7BiFeO_3-0.3BaZr_{0.02}Ti_{0.98}O_3-0.01molMnO_2 (70BFBTMn) piezoelectric ceramics to reduce their sintering temperatures. 70BFBTMn ceramics were sintered by a conventional solid reaction method, and their structural, dielectric, piezoelectric and ferroelectric properties were studied. These results indicate that 0.5% (mole) Li_2CO_3 is the optimized content and it can reduce the sintering temperature by 100°C, making the possibility of the piezoelectric ceramics cofiring with Ag electrodes at low temperatures to manufacture multilayer piezoelectric actuators.

Keywords: Piezoelectric ceramics; BiFeO3-BaTiO3; low temperature co-fired ceramic technology.

1. Introduction

High-performance piezoelectric ceramics can be applied in many fields, such as sensors and actuators due to the efficient conversion between mechanical energy and electrical energy. Pb(Zr,Ti)O₃-based ceramics are dominant commercial materials. Currently, because of the toxic Pb element and low depoling temperature (<200°C), Pb-free piezoelectric ceramics are extensively studied for replacing Pb(Zr,Ti) O₃-based ceramics and applying at high temperatures, for instance, bismuth-based,¹ sodium bismuth titanate-based,² barium titanate-based,^{3,4} potassium sodium niobate-based,⁵ etc. Although the market for piezoelectric devices is dominated by Pb(Zr,Ti_{1-x})O₃ ceramics, lead-free piezoelectric ceramics will experience a strong growth.^{6,7}

Among them, BiFeO₃-BaTiO₃ solid solution is regarded as a new generation of lead-free materials.⁸ From pure BaTiO₃ to pure BiFeO₃, the structure of BiFeO₃-BaTiO₃ varies from tetragonal, pseudocubic, and rhombohedral. The compositions of BiFeO₃-BaTiO₃ close to a mixed phase region of rhombohedral and pseudocubic have attracted great attention due to excellent piezoelectric properties. Large macrostrain was attributed to lattice distortion, domain switching, and phase transition.⁹ It is also reported that high electrostrain is related to the formation of an active pseudosymmetric structure.¹⁰ Ceramics with a rhombohedral phase show much better thermal stability and aging properties than those with mixed pseudocubic and rhombohedral phases.¹¹ Fast cooling can create core-shell microstructured BiFeO₃-BaTiO₃ with improved piezoelectric properties,¹² and the detailed nanoscale structural analysis shows the core-shell includes a pseudocubic shell enriched in Ba and Ti, an octahedrally tilted outer core richest in Bi and Fe, and an apparently pseudocubic inner-core phase richer in Ba and very low in Ti.¹³ When the content of $BiFeO_3$ is higher than 67%, the solid solution is the rhombohedral phase with the curie temperature higher than 500°C,¹¹ which makes BiFeO₃-BaTiO₃ based piezoelectric devices can operate at high temperatures. For instance, after Mn doping the Curie temperature, dielectric constant, dielectric loss, piezoelectric coefficient, and remnant polarization of 0.69BiFeO₃-0.31BaTiO₃ are 598°C, 704, 0.067, 82 pC/N, and 18.8 μ C/cm², respectively.¹⁴ In addition, since there is no other phase transition below the Curie temperature, the piezoelectric properties of BiFeO₃-BaTiO₃ solid solutions are stable at high temperatures. For instance, the piezoelectric coefficient is stable prior to 500°C after air-annealing for Mn-modified 0.75BiFeO₃-0.25BaTiO₃,¹⁵ indicating excellent thermal stability of piezoelectricity and comparable to that of bismuth titanate-based high-temperature piezoceramics.¹⁶

In the field of high-temperature piezoelectric actuators, low-temperature co-fired ceramic technology is commonly used to obtain large displacement at low voltage. For instance, the BiFeO₃-BaTiO₃-based multilayer actuator was fabricated using Pt as internal electrodes.¹⁷ Nevertheless, Pt is too expensive to be widely used in industry. The less expensive electrode like Ag is not suitable because the sintering temperatures of BiFeO₃-BaTiO₃ piezoelectric ceramics

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are commonly higher than 1000°C¹¹ which is too high for Ag paste used in low-temperature co-fired ceramic technology. Thus, reducing the sintering temperature of BiFeO₃-BaTiO₃-based ceramics is greatly expected for commercialized applications of BiFeO₃-BaTiO₃-based ceramics for multilayer actuators. By adding some oxide, the sintering temperature of ceramics can be reduced. For instance, by adding CuO, the sintering temperature of BiScO₃-PbTiO₃ piezoceramics is reduced and a multilayer actuator with Ag electrodes was successfully fabricated.¹⁸ Thus, it is possible to reduce the sintering temperature of BiFeO₃-BaTiO₃ by choosing the right sintering can be inhibited also by reducing the sintering

temperature. Since Li₂CO₃ has been proved to be an effective additive to reduce the sintering temperature of piezoceramics,^{19–23} in this work, Li₂CO₃ was added in 70BFBTMn to reduce the sintering temperature. In the composition, 1% (mole) Mn was added to improve resistivity,^{14,24–26} while 2% (mole) Zr was doped to enhance strain according to our previous works.^{27,28}

2. Experimental Procedure

70BFBTMn - xLi_2CO_3 (x = 0.005, 0.01, 0.02, 0.03) ceramics were prepared by a conventional solid-state reaction method. The raw materials used were BaCO₃, ZrO₂, Fe₂O₃,



Fig. 1. Microstructures of 70BFBTMn ceramics with different Li_2CO_3 contents sintered at 850°C and 1000°C.

 MnO_2 , TiO₂, Bi₂O₃, and Li₂CO₃. During the mixing process, 1 mol.% excess Bi was added to compensate for the volatilization of Bi. The raw materials were weighted according to the stoichiometric ratio and ball-milled for 24 h on a planetary ball mill, then the dry powders were calcined at 750°C for 4 h. Then, powders were pressed into pellets of 12 mm in diameter and 1 mm in thickness at a pressure of 150 MPa. The polyvinyl alcohol binder was burned out at 6°C for 4 h. Ceramics were sintered at 850–1000°C for 2 h.

The relative density of ceramic samples was measured by the Archimedes method. The morphology of thermal-etched ceramics was observed by field emission scanning electron microscopy (SEM, Zeiss G300). Phase structures of ceramics were analyzed by X-ray diffraction (XRD, Rigaku-D/MAX-2550) with Cu K_{a1} radiation ($\lambda = 1.54056$ Å). The dielectric properties were measured by an Agilent impedance analyzer (Agilent 4294A). Ferroelectric hysteresis loops (P-E) and electric-field-induced strain loops (S-E) of samples were measured by the ferroelectric testing system (TF1000 analyzer) at room temperature.

3. Results and Discussion

The microstructure of 70BFBTMn - xLi₂CO₃ ceramics sintered at different temperatures is shown in Fig. 1. Too many pores were observed in the 70BFBTMn ceramics sintered at 850°C and 900°C without adding Li₂CO₃, and the dense microstructure can be obtained at 1000°C. After adding Li₂CO₃, the pores decrease significantly after 900°C sintering, 950°C sintered ceramics have fine grains and clear grain boundaries, and the size of grains increases for 1000°C sintered ceramics. When the content of Li₂CO₃ is larger than 2%, the shape of grains becomes irregular. These results indicate that the sintering temperature of 0.7BiFeO₃-0.3BaZr_{0.02}Ti_{0.98}O₃ is decreased after adding Li₂CO₃, and the optimized content of Li₂CO₃ is less than 2%. The main reason for lowering the sintering temperature here is transient liquid phase assisted sintering as in the low-temperature synthesis of AlN ²⁹ and Pb-based piezoeletrics^{30,31} because the melting temperature Li₂CO₃ is 723°C.²⁹

The relative density of 70BFBTMn ceramics with different Li₂CO₃ contents is shown in Fig. 2. The ceramics sintered at 900–1000°C have high densities, and the relative density is above 94% for 70BFBTMn ceramics with Li₂CO₃ addition. For all sintered ceramics, the highest density is 96.4%, which corresponds to 0.5% Li₂CO₃ added 70BFBTMn ceramic sintered at 950°C. Overall, above 900°C is the suitable sintering temperature for Li₂CO₃ modified 70BFBTMn ceramics. Although the complete measurements were done for all sintered ceramics, the effect of doping content is dominant. The following discussions will only focus on the results of 900°C sintered ceramics.

Figure 3 shows the X-ray diffraction patterns of 70BFBTMn ceramics with different Li₂CO₃ contents sintered at 900°C. All the sintered ceramics exhibit pure perovskite



Fig. 2. The relative densities of 70BFBTMn ceramics with different Li_2CO_3 contents sintered at 850°C and 1000°C.



Fig. 3. X-ray diffraction patterns of 70BFBTMn - xLi_2CO_3 ceramics at 900°C.

structure and no impurity phases are observed within the detection limit of X-ray diffraction. Although when the content of BiFeO₃ is higher than 67% the rhombohedral structure is expected, neither pseudocubic $(111)_c$ peak nor $(200)_c$ peak splits as shown in Fig. 1, indicating the prepared ceramics are pseudocubic, which could be due to extremely small rhombohedral distortion as reported¹⁴ or more complex core–shell structure.¹³

Dielectric properties of 70BFBTMn - xLi_2CO_3 ceramics are given in Fig. 4. At room temperature for all Li₂CO₃ added ceramics, 0.5% Li₂CO₃ added ceramic has the highest dielectric permittivity from 10³ Hz to 10⁶ Hz as shown in Fig. 4(a). Since at room temperature, the ceramic without Li₂CO₃ has a dielectric permittivity of 632 (1 kHz) and a dielectric loss of 0.045 (1 kHz), the values are 577 and 0.046 for 0.5% Li₂CO added ceramic, suggesting that the dielectric permittivity does not deteriorate significantly and the dielectric loss remains small after adding 0.5% Li₂CO₃. From room temperature to 600°C for all Li₂CO₃ added ceramics, the dielectric loss of 0.5% Li₂CO added ceramic is the lowest as shown in



Fig. 4. Frequency-dependent dielectric permittivity and dielectric loss (a), and temperature-dependent dielectric permittivity and dielectric loss (b) of 70BFBTMn - *x*Li₂CO₃ ceramics sintered at 900°C.

Fig. 4(b), and the dielectric permittivity of 0.5% Li₂CO added ceramic is like to that of 70BFBTMn. These results indicate that 0.5% Li₂CO adding has the weakest influence on the high-temperature dielectric of all Li₂CO added 70BFBTMn piezoelectric ceramics. For all Li₂CO₃ added ceramics, with increasing temperature dielectric loss increases significantly. This could be related to Li replacing the A-site or B-site of the perovskite structure creating more oxygen vacancies. The mobility of oxygen vacancies increases at high temperatures, thus the conductivity of the ceramics increases. As a result, the dielectric loss rises at high temperatures.

Figure 5 shows ferroelectric hysteresis loops, bipolar, and unipolar strains of 70BFBTMn - xLi_2CO_3 ceramics at room temperature. All ceramics show typical ferroelectric hysteresis loops as shown in Fig. 5(a). At an applied electric field of 80 kV/cm (1 Hz), the polarization is 32.0 μ C/cm², 27.7 μ C/ cm², 16.8 μ C/cm², 14.8 μ C/cm², and 13.0 μ C/cm², the remanent polarization is 22.3 μ C/cm², 16.2 μ C/cm², 9.8 μ C/cm², 7.6 μ C/cm², 6.6 μ C/cm², and the coercive field is 40 kV/cm, 33 kV/cm, 37 kV/cm, 30 kV/cm, and 33 kV/cm, for x = 0, 0.5, 1, 2, and 3, respectively.

The bipolar and unipolar strain hysteresis loops of $70BFBTMn - xLi_2CO_3$ ceramics are shown in Figs. 5(b) and



Fig. 5. Ferroelectric hysteresis loops (a), bipolar strain loops (b), and unipolar strain loops (c) of 70BFBTMn - xLi_2CO_3 ceramics sintered at 900°C.

5(c). The standard butterfly curves are observed for all ceramics. The bipolar strain of 70BFBTMn is 0.235%. The bipolar strain of 0.5% Li₂CO₃ added ceramic reaches up to 0.102%, much higher than other Li₂CO₃ added ceramics (varying from 0.021 to 0.036). The unipolar strain curves are shown in Fig. 5(c). The stain at 60 kV/cm is 0.200%, 0.103%, 0.035%, 0.024%, and 0.024 for x = 0, 0.5, 1, 2, and 3, respectively. The high field strain coefficient d^*_{33} was calculated by $d^*_{33}=S_{max}/E_{max}$, where S_{max} is the largest strain, and E_{max} is the highest applied electric field. d^*_{33} is 333 pm/V, 171 pm/V, 58 pm/V, 40 pm/V, and 40 pm/V for x = 0, 0.5, 1, 2, and 3, respectively.

Above electrical properties indicate that the ferroelectric, piezoelectric, and dielectric properties of 70BFBTMn deteriorate after adding Li_2CO_3 , which could be related to the

reduced size of grains as shown in Fig. 1, grain boundaries with Lirich, and Li partly replacing A-site or B-site of the perovskite structure.

4. Conclusion

 $0.7Bi_{0.5}FeO_3-0.3BaZr_{0.02}Ti_{0.98}O_3-0.01MnO_2-xLi_2CO_3$ (x = 0.5%, 1%, 2%, 3%) piezoelectric ceramics were prepared by the conventional solid reaction method with varying the sintering temperature from 850°C to 1000°C. Dense ceramics are obtained when the sintering temperature is higher than 900°C. Thus, Li₂CO₃ doping can effectively reduce the sintering temperature of 0.7Bi_{0.5}FeO₃-0.3BaZr_{0.02}Ti_{0.98}O₃-0.01MnO₂ from 1000°C to 900°C. For 900°C sintered ceramics, pure perovskite structure can be obtained when varying the content of Li₂CO₃ from 0.5% to 3%. 0.5% Li₂CO₃ added ceramic deteriorates the electrical properties of 0.7Bi_{0.5}FeO₃-0.3BaZr_{0.02}Ti_{0.98}O₃-0.01MnO₂ slightly with a dielectric permittivity of 577 (1 kHz), a dielectric loss of 0.046 (1 kHz), a of 27.7 μ C/cm², a strain of 0.102%, and a high field strain coefficient d_{33}^* of 171 pm/V. Thus, 0.5% Li₂CO₃ added 0.7Bi_{0.5}FeO₃-0.3BaZr_{0.02}Ti_{0.98}O₃-0.01MnO₂ piezoelectric ceramic is expected to co-fire with Ag paste to prepare multilayer actuators using low temperature co-fired ceramic technology.

Acknowledgments

The work was supported by the National Natural Science Foundation of China (Grant Nos. 11704242, 51872180 and 51672169) and the Natural Science Foundation of Shanghai, China (Grant Nos. 17ZR1447200 and 18ZR1414800).

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