Recent progress on advanced transmission electron microscopy characterization for halide perovskite semiconductors

Xiaomei Wu, Xiaoxing Ke⁺, and Manling Sui⁺

Faculty of Materials and Manufacturing, Beijing University of Technology, Beijing 100124, China

Abstract: Halide perovskites are strategically important in the field of energy materials. Along with the rapid development of the materials and related devices, there is an urgent need to understand the structure–property relationship from nanoscale to atomic scale. Much effort has been made in the past few years to overcome the difficulty of imaging limited by electron dose, and to further extend the investigation towards operando conditions. This review is dedicated to recent studies of advanced transmission electron microscopy (TEM) characterizations for halide perovskites. The irradiation damage caused by the interaction of electron beams and perovskites under conventional imaging conditions are first summarized and discussed. Low-dose TEM is then discussed, including electron diffraction and emerging techniques for high-resolution TEM (HRTEM) imaging. Atom-ic-resolution imaging, defects identification and chemical mapping on halide perovskites are reviewed. Cryo-TEM for halide perovskites is discussed, since it can readily suppress irradiation damage and has been rapidly developed in the past few years. Finally, the applications of *in-situ* TEM in the degradation study of perovskites under environmental conditions such as heating, biasing, light illumination and humidity are reviewed. More applications of emerging TEM characterizations are foreseen in the coming future, unveiling the structural origin of halide perovskite's unique properties and degradation mechanism under oper-ando conditions, so to assist the design of a more efficient and robust energy material.

Key words: organic-inorganic hybrid perovskite solar cell materials; energy materials; scanning electron microscopy; transmission electron microscopy; irradiation damage

Citation: X M Wu, X X Ke, and M L Sui, Recent progress on advanced transmission electron microscopy characterization for halide perovskite semiconductors[J]. J. Semicond., 2022, 43(4), 041106. https://doi.org/10.1088/1674-4926/43/4/041106

1. Introduction

The fast rising of halide perovskites has drawn significant attention in the field of energy materials in the past decade. Owning to their unique structure, high degree of tunability, and excellent photoelectric properties, halide perovskites have been widely used in photovoltaic and optoelectronic devices, including solar cells, LED, and photodetectors^[1-5]. Taking CH₃NH₃Pbl₃ (referred as MAPbl₃) as an example, it is a prototypical organic-inorganic hybrid perovskite used in emerging photovoltaic material, and has a bandgap of about 1.55 eV, large absorption coefficient similar to Si, better tolerance to defect, high carrier mobility, long length of electron-hole transport and thus high conversion efficiency^[6, 7]. In addition, it can be prepared as thin films for flexible devices, easy to synthesize and allows scaling-up for production^[8, 9]. In just a few years, the conversion efficiency of the perovskite solar cell (PSC) using MAPbl₃ as the light-absorption layer has increased from 3.8% to 25.7% as reported in a most recently published paper^[10]. By replacing the A-site cation with other organic moieties and doping at the B-site/X-site, not only the bandgap can be tuned, but also its stability under environmental conditions can be improved, thereby providing more possibilities for the developments and applications of halide perovskites^[11–13]. In addition, two-dimensional (2D) nanosheets, one-dimensional (1D) nanowires^[14, 15], all-inorganic quantum dots such as CsPbX₃^[16, 17], Pb-free double perovskites nanostructures^[18–23] have also demonstrated remarkable performance and unlimited applications for light-emitting devices and photodetectors etc.

The unusual optoelectronic properties and performance of halide perovskites are closely related to their unique crystal structure and microstructure. The name of the perovskite is originated from the calcium titanate (CaTiO₃) compound in the perovskite mineral, and is referred to as ABX₃ compound (where A and B are cations and X is an anion). In perovskite structure, [BX₆] octahedra are corner-sharing to be ordered in a cubically, where the cations at A-site are located in the center of cubic cage^[24]. Halide perovskites have similar structures following the general form of ABX₃, where the A site is a monovalent cation, such as $CH_3NH_3^+$ ($R_A = 0.18$ nm), $CH_3CH_2NH_3^+$ ($R_A = 0.23$ nm), $NH_2CH=NH_2^+$ ($R_A \approx 0.19-$ 0.22 nm), or inorganic cations such as Cs⁺ ($R_A = 0.167$ nm), Rb⁺ ($R_A = 0.152 \text{ nm}$), K⁺ ($R_A = 0.138 \text{ nm}$); the B site is mostly divalent cations such as Pb²⁺ ($R_B = 0.119$ nm), Sn²⁺ ($R_B =$ 0.112 nm); the X site is a halogen anion such as Cl⁻ (R_X = 0.181 nm), Br⁻ (R_{χ} = 0.196 nm), I⁻ (R_{χ} = 0.220 nm). Additionally, when the B site Pb2+ ion is replaced by a monovalent cation and a trivalent cation, an inorganic Pb-free double perovskite is formed, written as A₂BB'X₆, such as Cs₂AgInCl₆, Cs₂Ag-BiBr₆ and Cs₂AgBiCl₆. The unique perovskite structure, e.g. the crystal symmetry, the vibration and ordered arrangement of the organic groups, and the tilt of the [Pbl₆] octahedra have shown to be related to their photovoltaic proper-

Correspondence to: X X Ke, kexiaoxing@bjut.edu.cn; M L Sui, mlsui@bjut.edu.cn Received 28 FEBRUARY 2022; Revised 25 MARCH 2022. ©2022 Chinese Institute of Electronics

ties^[25]. For all-inorganic halide perovskite such as CsPbX₃, it is known to own multiple phases and is therefore possible to form nano-domain structure at the coexistence of multiphases, which further influences its photoelectric properties^[26]. Chemical composition inhomogeneity, strain distribution, polarization, defects such as domain walls, grain boundaries and surface defects are also known to influence the materials' properties and performances^[27–30]. It is hence of great importance to study the microstructures and their degradation under operando conditions and to understand the relationship between properties and microstructures^[31–33].

Transmission electron microscopy (TEM) is one of the most straightforward characterization techniques to investigate the microstructure from nanometer to the atomic scale^[34–37]. Along with the rapid developments of halide perovskites, much progress has been achieved in the past few years. Rothmann et al. were one of the first research groups to report twinning structure in the as-grown MAPbl₃ single crystal sample using electron diffraction^[38]. The atomic-resolution structure of MAPbl₃ was further revealed using direct electron detector and cryo-TEM, respectively^[39, 40]. Yu et al. reported a nanodomain structure in CsPbBr₃ 2D nanocrystals where tetragonal and cubic phases coexist^[41]. On the other hand, in-situ TEM and guasi-in-situ TEM has been developed on cross-sectional samples from MAPbl₃ PSCs, where ion migration and degradation of the perovskite thin film under environmental conditions have been studied in nanoscale^[42-44]. However, the microstructural investigation of halide perovskites using TEM has been significantly constrained by its extreme sensitivity to electron beams. The inevitable knock-on damage and radiolysis damage on halide perovskites accelerate its degradation in TEM, complicating the interpretation of as-observed phenomena: when irradiated by electron beams, the halogen ions rapidly desorb from the structure and the organic moieties easily break down due to radiolysis, resulting in structure collapse^[3, 45]. As reported by Rothmann *et al.* and confirmed by other TEM study, a total dose of <100 e/Å² can destruct halide perovskite^[46], which is far below the imaging condition for conventional TEM study.

Driven by the urgent need to understand the microstructure-properties relationship for halide perovskites which is strategically important, much effort has been made in the past few years to overcome the difficulty of imaging limited by electron dose, and to further extend the investigation towards operando conditions. This review is therefore dedicated to recent studies of advanced TEM characterizations for halide perovskites. Firstly, the irradiation damage and "artefacts" caused by the interaction of electron beams and perovskites under conventional imaging conditions are summarized and discussed, calling attention to the interpretation of TEM results for this particular materials. Low-dose TEM is then discussed, including conventional technique of electron diffraction and emerging techniques for high-resolution TEM (HRTEM) imaging. Atomic-resolution imaging and defects identification on halide perovskites are reviewed. Composition analysis using energy dispersive X-ray microscopy (EDX) and electron energy loss spectroscopy (EELS) mapping are also discussed. Cryo-TEM for halide perovskites is then given special attention in this review, since it can readily suppress irradiation damage and has been rapidly developed for imaging beam-sensitive energy materials in the past few years. Finally, the applications of *in-situ* TEM in the degradation study of per-



Fig. 1. (Color online) Schematic illustration of advanced TEM characterization for halide perovskites.

ovskites under environmental conditions such as heating, biasing, light illumination and humidity are reviewed. An illustration of advanced TEM characterizations for halide perovskites is presented in Fig. 1.

2. Irradiation damage on halide perovskites by conventional TEM imaging

The irradiation damage induced by energetic electrons during the imaging of halide perovskites has significantly limited the TEM characterization, and is considered as the cornerstone for the interpretation of as-observed microstructures.

When organic-inorganic hybrid perovskites are imaged using conventional TEM, the structure of the halide perovskites can be destroyed in seconds, resulting in the irradiation damage such as generation of "bubbles" and precipitation of nanoparticles, as shown in Fig. 2. Chen et al. noticed severe damage in the MAPbl₃ perovskite polycrystalline film when imaged at a relatively high electron dose of ~9870 e/(Å²·s) (i.e. the dose for obtaining conventional HRTEM images), where nanoparticles precipitated quickly within the irradiated area (Fig. 2(a)). Such irradiation damage was attributed to the electric field generated by the electron beam within the perovskite, and remained by reducing electron dose to 500 e/(Å²·s)^[47, 48]. On the other hand, Kim et al. revealed the generation and expansion of "bubbles" by obtaining a series of TEM images through continuous irradiation on MAPbl₃ perovskite single crystals (Fig. 2(b))^[49]. In general, the degradation of organic-inorganic hybrid halide perovskites in vacuum or in air has been frequently noticed, and the precipitated nanofragmentaions are mostly identified mostly as PbI₂ or Pb clustering^[50, 51]. In addition, 2D organic-inorganic hybrid halide perovskites are more susceptible to electron beam damage due to the presence of interlayer organic ligands, and very few TEM studies on 2D perovskites were reported. Dou et al. attempted to study (C₄H₉NH₃)₂PbBr₄ (i.e. BA₂PbBr₄) by TEM, which degraded rapidly into Pb nanoparticles under a very low dose^[52]. Nie et al. reported the same phenomenon in the study of BA₂PbBr₄ nanosheets, where the nanoparticles



Fig. 2. (Color online) (a) Electron beam irradiation damage observed in free-standing MAPbl₃ films. (i, ii). TEM images recorded initially and after the irradiation (9870 e/(Å²·s) for ~2 min), respectively^[47]. (b) Time-series of TEM images on MAPbl₃ single crystal showing the electron beam damage from 0 to 50 s, where bubble-like morphology (colored arrows) emerged and grew^[49]. (c) Time-series of TEM images obtained on BA₂PbBr₄ nanosheets^[53]. (d) TEM images of the (i) CsPbCl₃, (ii) CsPbBr₃ and (iii) CsPbl₃ QDs where "dark spots" present at the QD corners indicated irradiation damage^[54]. (e) Schematic illustration of CsPbBr₃ degradation pathway^[56].

quickly precipitated and the nanosheets were amorphized even at low magnification, as shown by time series TEM in Fig. $2(c)^{[53]}$.

Irradiation damage is also frequently noticed in all-inorganic halide perovskites. Taking CsPbX₃ for example, Li et al. found that even short time of exposure could cause "dark spots" to emerge at the corners of the quantum dots (QDs) when imaging CsPbCl₃, CsPbBr₃ and CsPbl₃ QDs as shown in Fig. 2(d)^[54]. The emerged "dark spots" were confirmed as PbX₂ nanoparticles, as reported in other related works as well^[55-58]. Further analysis inferred that the irradiation-induced precipitation of PbX₂ was due to the surface defects on nanoparticles, and could be effectively suppressed by ZnX₂/hexane treatment which "repaired" the halogen ion defects^[54]. Similarly, Wang et al. reported a layer-by-layer decomposition of CsPbBr₃ which also started from the surface, confirming the impact of surface passivation^[58]. Continuous irradiation may further cause phase separation and lead to amorphization in nanocrystals as revealed by Funk et al.^[59]. Similarly, Dang et al. studied CsPbX₃ 2D nanosheets and 1D nanowires by HAADF-STEM imaging, and also reported the precipitation of nanoparticles, at low dose of 400 e/Å² under both 80

and 200 kV^[56]. The degradation mechanism of CsPbBr₃ under electron beam irradiation was proposed to be dominated by decomposition coupled with knock-on damage: the incident electrons first led to the desorption of halogen ions and reduced Pb²⁺ to Pb⁰ partially; Pb⁰ then migrated and aggregated to form precipitation of Pb; finally, the CsPbBr₃ collapsed and became amorphized, as illustrated in Fig. 2(e).

More recently, Pb-free halide perovskites with a double perovskite structure are being developed, such as $Cs_2AgBiBr_6$, $Cs_2CuSbCl_6$, $Cs_2AgInCl_6$. Zhou *et al.* synthesized $Cs_2CuSbCl_6$ nanoparticles, and Cruetz *et al.* synthesized $Cs_2AgBiCl_6$, $Cs_2Ag-BiBr_6$ and $CsAgBil_6$ nanoparticles, and they were all characterized by TEM^[60, 61]. However, irradiation damage was not alleviated in these structures, if not worse. Quick precipitation of nanoparticles and decomposition was also noticed in these materials even at extremely low dose. It was proposed that the degradation was initially induced by the Knotek-Fabel effect at the surface, and was then followed by decomposition into $Cs_3Bi_2Br_9$, $BiBr_3$, $CsAgBr_2$ and AgBr. When the electron beam dose rate was less than 2 $e/(Å^2 \cdot s)$, the structure could be maintained for about 4 min as confirmed from electron diffraction. Alternatively, $Cs_2AgBiBr_6$ lattice image could be ob-



Fig. 3. (Color online) (a) Degradation of MAPbl₃ studied using SAED taken from a near-<110> t-oriented grain: i) the initial, pristine phase and ii) after 1 min (total dose per area of $\approx 1 \times 10^2$ e/Å²), iii) 2 min (total dose per area of $\approx 2 \times 10^2$ e/Å²), iv) 18 min (total dose per area of $\approx 2 \times 10^3$ e/Å²) of weak electron beam exposure (≈ 2 e/Å²)^[46]. (b) Degradation in MAPbX₃ by forming superstructured intermediate phase: i) atomistic structure of tetragonal MAPbl₃; ii) electron diffraction (ED) pattern along the [001]_c direction; iii) the observed ED of superstructure phase; iv) the simulated ED of superstructure phase MAPbl_{2.5}; v) the corresponding atomistic structure; vi) atomistic structure of MAPbBr₃; vii) ED pattern along the [001] direction; viii) the observed ED pattern with additional reflections; ix) the simulated ED of superstructure phase MAPbl_{2.5}; x) the corresponding atomistic structure; vi) atomistic structure phase MAPbBr_{2.5}; x) the corresponding atomistic structure; vii) atomistic structure phase MAPbBr_{2.5}; x) the corresponding atomistic structure; vii) atomistic structure phase MAPbBr_{2.5}; x) the corresponding atomistic structure; vii) atomistic structure phase MAPbBr_{2.5}; x) the corresponding atomistic structure; viii) the observed ED pattern with additional reflections; ix) the simulated ED of superstructure phase MAPbBr_{2.5}; x) the corresponding atomistic structure; viii) atomistic structure phase MAPbBr_{2.5}; x) the corresponding atomistic structure with ordered bromine vacancies^[69]. (c) TEM images and [110] oriented-SAED patterns taken from grain highlighted in yellow circles from FAPbl₃ films with (i, ii) 10% MA, (iii, iv) 20% MA, (v, vi) 30% MA, (vii, viii) 40% MA^[78]. (d) Stabilization of photoactive perovskites against degradation by tilted octahedral, as illustrated by structural model (i–vi), calculated energy difference (vii), AFM-IR characterization (viii–x), and TEM imaging (xi) with corresponding SAED (xii–I)^[79].

tained under a single exposure with a dose rate of less than 281 e/(Å²·s) using conventional HRTEM^[62].

It is generally accepted that the electron dose and dose rate have the most influence on the degradation of halide perovskites under electron irradiation, yet it is interesting to discuss the impact of accelerating voltage, i.e. beam energy. In the damage caused by electron beam radiation, it is generally believed that there are three types of radiation damage mechanisms, that is, knock-on damage, radiolysis, and rise of local temperature caused by phonons excited by electron beam radiation^[63, 64]. As the latter two factors are more related to electron dose, the first factor is closely related beam energy. Recently, Cai et al. calculated the knock-on damage on hybrid halide perovskites using first principle, and showed that iodine is only knocked-out when accelerating voltage is higher than ~250 kV^[65]. This is in contrast to experimental results, where low kV has been performed to study the hybrid perovskites where degradation is not noticeably reduced^[56, 66, 67]. Therefore, it is a strong evidence that radiolysis dominates the degradation of halide perovskites under electron beam irradiation, and low-dose is vital for TEM characterization.

3. Degradation mechanism and defects identification of halide perovskite using electron diffraction

Due to the beam-sensitivity of halide perovskites, selec-

ted area electron diffraction (SAED) with the advantage of low dose, has become a powerful tool to study the degradation mechanism and intrinsic microstructures of halide perovskites.

Rothmann *et al.* acquired SAED time-series through continuous irradiation to study the degradation of MAPbl₃ single crystal as shown in Fig. 3(a)^[46]: (1) the disappearance of intrinsic twinning structures; (2) the generation of $\sqrt{2} \times \sqrt{2}$ superlattices; (3) the disappearance of the superlattices and the resize of diffraction patterns; (4) the decrease of I : Pb from 3 : 1 to 2 : 1, suggesting the formation of Pbl₂. Degradation mechanism was proposed as: CH₃NH₃Pbl₃ $\stackrel{e}{\rightarrow}$ ($-CH_2-$) + NH₃ \uparrow + HI \uparrow + Pbl₂, where the volatility of NH₃ and HI first led to the twisting of [Pbl₆] octahedra, thereby resulting in intermediate phase as $\sqrt{2} \times \sqrt{2}$ superlattices. It is interesting to notice that the degradation of MAPbl₃ was related to the total dose rather than dose rate. Under the accelerating voltage of 200 kV, a dose of <100 e/Å² can cause the degradation of MAPbl₃ to Pbl₂ through the above-mentioned process.

Similar to Rothmann's work, Chen *et al.* also studied the degradation process of MAPbl₃ through SAED time series^[68]. Whilst degradation from MAPbl₃ to Pbl₂ was also observed and confirmed, the generation of superstructure diffraction patterns was interpreted due to the formation of ordered halide vacancies, thereby resulting in MAPbl_{2.5} intermediate phase. Moreover, the effects of electron dose and dose rate under the accelerating voltage of 300 kV were compared: at a dose rate of 0.5 e/(Å²·s), MAPbl_{2.5} mesophase appeared in

303 s, and at a dose rate of 1 $e/(Å^2 \cdot s)$, MAPbl_{2.5} mesophase appeared in 73 s, indicating that the degradation of MAPbl₃ under electron beam irradiation was not only related to the total dose, but also to the dose rate, and thus a higher dose rate might accelerate the degradation of MAPbl₃. The same group of authors continued to study the degradation process of MAPbBr₃ by SAED and reported a similar degradation process, proposing a general degradation mechanism for MA-PbX₃ as halogen ion desorption \rightarrow formation of ordered vacan- $\mathsf{cies} \to \mathsf{formation} \text{ of } \mathsf{MAPbX}_{\mathsf{3-x}} \text{ mesophase} \to \mathsf{precipitation} \text{ of }$ PbX₂", as shown in Fig. 3(b)^[69]. Similar degradation pathways had been generally recognized in oxide perovskites, but it was first proposed in halide perovskites^[70, 71]. More recently, similar degradation mechanism was reported in all-inorganic perovskite of γ -CsPbIBr₂ as well, where an intermediate phase $[e.g., CsPb_{(1-x)}(IBr)_{(3-v)}]$ with a superstructure of ordered vacancies was formed, followed by reduction from Pb²⁺ to Pb⁰ and thus precipitation of Pb nanoparticles^[72].

SAED is also used to study the degradation mechanism of 2D nanosheets due to their extreme beam-sensitivity. Yuan *et al.* studied the structural changes of $BA_2FAPb_2I_7$ and $BA_2MAPb_2I_7$ under electron beam irradiation by recording SAED time series, and proposed the degradation mechanism as following: organic ligands between the nanosheets were firstly destroyed by the electron beam; the organic moieties and [PbBr₆] octahedra were then damaged, resulting in the rapid collapse of the structure^[73]. The proposed degradation mechanism was confirmed by Nie *et al.*: by investigating the morphology changes and SAED patterns of BA_2PbBr_4 nanosheets exposed to ultraviolet and white light, the authors found that the organic ligands between nanosheets were most sensitive to light irradiation; further exposure resulted in the tilt of [PbBr₆] octahedra and then amorphization.

In addition to the study of degradation mechanism, SAED has been employed to characterize the microstructures of halide perovskites in a damage-free manner. Rothmann et al. discovered the existence of nano-twinning in MAPbl₃ single crystal by SAED^[38]. Li et al. studied the intergrain planar defects in $MA_{1-x}FA_xPbI_3$ perovskites by low dose electron diffraction, and found that the as-fabricated solar cell performance correlated empirically with the density and structure of {111}_c planar defects and {112}_t twin boundaries^[74]. Gao et al. obtained SAED diffraction patterns of BA₂Pbl₄ nanosheets with different structures, and obtained the lattice image of 2D nanosheets through filtering and inverse fast Fourier transform^[75]. For all-inorganic perovskites, Pan et al. revealed the B-site cations disorder in Cs₂AgBiBr₆ using electron diffraction^[76]. Luo et al. combined SAED and nano-diffraction to reveal a partial ordering of Ag/Na cations in Cs₂Ag(Na)InCl₆ single crystal^[77]. Using a nano-beam with a diameter of ~5 nm (equal to current density of ~1.6 pA/cm²) and a dwell time of 0.1 s, the authors demonstrated damagefree scanning over the sample for several minutes. Pham et al. unraveled the influence of CsCl/MACl on stabilizing cubic α -FAPbl₃ phase and minimizing twinning/stacking faults through electron diffraction as shown in Fig. 3(c)^[78]. More recently, Doherty *et al.* revealed that α -FAPbI₃ stabilized by Cs and MA cations alloying had octahedral tilting of ~2°, which imparted the phase stability by frustrating transitions from photoactive to hexagaonal phases. Such minor tilting was neglected by bulk characterization such as XRD, but was resolved by electron diffraction, as shown in Fig. $3(d)^{[79]}$.

4. Emerging TEM approaches for halide perovskites at atomic resolution

Driven by the urgent need to understand the structureproperties relationship of halide perovskites, novel approaches on both hardware and methodology are being developed to overcome the limits opposed by electron dose, and to further push the spatial resolution down to atomic scale.

With regard to the degradation mechanism of halide perovskites under electron irradiation, it is generally accepted that: (1) the electron dose and dose rate are dominating the irradiation process, and therefore low-dose imaging is essential; (2) the disassociation of organic ligands and halide ions is the first to trigger decomposition of halide perovskites. In light of these understandings, various methods have been proposed to push HRTEM imaging of halide perovskites towards atomic scale, including: (1) deposition or coating of carbon film to alleviate the radiolysis damage; (2) improvement on detectors such as direct electron detector to enable ultra-lowdose imaging; (3) developments of low-dose imaging techniques such as exit-wave reconstruction, ptychography, and iDPC etc.

Above-mentioned techniques have demonstrated to be successful in resolving pristine structures of halide perovskites at atomic scale. Yu et al. revealed the microstructure of CsPbBr₃ by defocus series at a low dose rate of ~100 e/(Å²·s) under accelerating voltage of 80 kV^[41]. As shown in Fig. 4(a), nano-domain with the co-existence of cubic phase and orthogonal phase was demonstrated. Brennan et al. further applied the same approach onto CsPbBr₃ nanocubes and reported a size dependency on the crystal phases^[80]. Alternatively, Zhang et al. employed a direct-detection electron-counting camera to obtain the first atomic resolution image of MAPbBr₃ at a total dose of only 11 e/Å², revealing nanodomain structure where the MA⁺ cations were ordered locally in the single crystal, as shown in Fig. 4(b)^[39, 81]. Chen et al. also used direct-detection electron-counting camera to investigate the decomposition pathway of MAPbl₃ at atomic scale, where the intermediate phase of MA_{0.5}Pbl₃ with locally ordered vacancies was successfully imaged^[82]. Alternatively, iD-PC has been demonstrated to be helpful in revealing the atomic structures of halide perovskites: an early iDPC-STEM attempt for the investigation of MAPbBr₃ was performed by Song et al., enabling low-dose imaging at atomic scale^[81]; more recently, Qiao et al. demonstrated the magnified iDPC-STEM image could clearly distinguish the atomic distributions of Cs, Pb, and Br from monodispersed CsPbBr₃ nanocrystals, as encapsulated and stabilized in metal-organic framework^[83]. Recently, ptychography has been applied to image halide perovskites due to its advantage of low dose. As shown in Fig. 4(c), the presence of Br was revealed at atomic scale^[84]. It is interesting to mention that pulsed-beam TEM has been demonstrated on studying MAPbl₃, where the dose could be controlled as low as 0.001 e/Å2[85]. Although the spatial resolution is rather low compared to HRTEM, it offers a possibility for damage-free study of halide perovskites.

Thanks to the development of low-dose imaging tech-



Fig. 4. (Color online) (a) HRTEM of CsPbBr₃ nanocrystals (i) where the coexistence of cubic and orthorhombic phases were demonstrated by FFT patterns (ii, iii), simulated diffraction patterns (iv, v), and illustrated structure (vi, vii)^[41]. (b) CTF-corrected denoised HRTEM image (i) of CH₃NH₃Pb-Br₃ with different CH₃NH₃ orientations, where (ii, iii) the structural model (left) and the simulated projected potential map (right) corresponding to region 1 and 2 in (i), respectively^[39]. (c) Ptychography reconstructed image of CsPbBr₃, with the scale bar of 5Å^[84]. (d) Atomic-scale structures of intragrain stacking-fault (i) and twinning interfaces (ii) obtained on orthorhombic FA_{0.5}Cs_{0.5}Pbl₃ grains along the [100] projection direction^[86]. (e) Atomically resolved interface at the (2T)₂ Pbl₄–(2T)₂ PbBr₄ heterostructure^[87]. (f) Butterworth-filtered LAADF-STEM images of grain boundaries (i), triple junctions (ii), grain boundary (iii) and aligned vacancy defects indicated by red circle (iv), obtained from a 30-nm-thick film of FAPbl₃^[67].

nique, defects in halide perovskites have been imaged in detail as never before. Cai et al. recently succeeded in imaging grain boundaries of MAPbl₃ PSCs at atomic scale by developing a coating approach onto FIB-prepared cross section, as shown in Fig. 4(d)^[86]. The carbon coating can significantly inhibit the loss of volatile substances such as CH₃NH₂ and halogens. It was suggested by Chen et al. that depositing carbon film with a thickness of ~6-10 nm on MAPbl₃ could increase its tolerance against electron irradiation by nearly 10 times^[69]. On the other hand, Shi et al. successfully resolved the atomic structure on (2T)₂ PbI₄-(2T)₂ PbI₄-(2T)₂ PbBr₄ heterojunction nanosheets, through continuous exposure at extremely low dose and subsequent superposition of time series^[87]. In this manner, HRTEM images with a high signal-tonoise ratio were obtained under a total dose of only 9.2 e/Å², and defects at the heterojunction interface were revealed at

atomic scale as shown in Fig. 4(e). Recently, Jung et al. revealed the growth of polytypic Pbl₂ on the surface of 2D Ruddlesden-Popper halide perovskites BA2MA2Pb3I10, where a sequential layer-by-layer degradation from perovskites to Pbl₂ was imaged at atomic scale^[88]. CsPbBr₃ nanocrystals were also studied by HAADF-STEM, where Ruddlesden-Popper stacking fault was revealed by a beam current of 50 pA at an accelerating voltage of 300 kV^[89, 90]. The author mentioned that a short dwell time of 3–6 μ s can suppress irradiation damage. Additionally, Rothmann et al. used low-angle annular dark field STEM (LAADF-STEM) to study FAPbl₃ thin films, and successfully imaged grain boundaries at atomic scale as shown Fig. 4(f)^[67]. It was suggested by the authors that LAADF-STEM could effectively lower the beam damage on halide perovskites, and therefore might find more applications in resolving atomic structure of beam-sensitive materials.

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Fig. 5. (Color online) (a, b) Atomically resolved HAADF-STEM images and corresponding EDX-mappings of CsPbBr₃ nanoplates^[92]. (c) STEM-EELS from a CsPbBr₃ nanosheet to determine bandgap, where (i) demonstrates measured data and (ii) shows as-calculated bandgap value^[93].

5. Chemical mapping of halide perovskites using STEM-EDX or STEM-EELS

Due to the beam sensitivity, elemental mapping of halide perovskites has been a standing issue for microstructure investigations. Large dose is often required to obtain enough counts for either EDX-mapping or EELS-mapping, which exceeds the degradation threshold of halide perovskites. Although chemical mapping has been reported in many halideperovskites study, it is also known that the instability of halide perovskites during beam irradiation thus leads to the reliability of elemental mapping, questioning whether the mapped results reflect the pristine structure. Therefore, novel approaches have been proposed to enable the elemental mapping down to atomic scale.

Kosasih *et al.* performed a systematic study on the evaluation of STEM-EDX quality demonstrated on state-of-art triple-cation, double-halide perovskite with a nominal stoichiometry of $Cs_{0.06}FA_{0.79}MA_{0.15}Pb(I_{0.85}Br_{0.15})_3^{[91]}$. The acquisition parameters were systematically varied to evaluate the relationship between electro dose, data quality and irradiation damage. It was found that the iodine content decreased exponentially with electron dose, and thus called attention to the interpretation of STEM-EDX mapping results.

In order to minimize electron damage, Liu *et al.* recently fabricated a highly-beam-stable CsPbBr₃ nanoplates through an interface-assisted regrowth method using cyanamide^[92].

By defect-healing self-assembly/regrowth processes and surface passivation by cyanamide molecules, the as-prepared CsPbBr₃ was demonstrated to be highly stable against electron beam irradiation and thus achieved elemental mapping at atomic resolution, as shown in Figs. 5(a) and 5(b). The stability improvement as shown in this work demonstrates that the beam damage starts from the surface, where surface passivation and defect-healing can effectively suppress the radiolysis damage, although further experiments are needed to check the universality of this method.

On the other hand, STEM-EELS has been demonstrated on orthorhombic CsPbBr₃ to determine bandgap as shown in Fig. $5(c)^{[93]}$. Care was taken in spreading the electron dose for the EELS experiments so to minimize irradiation damage, and the beam current was set to be <5 pA. Meanwhile, the spectra were collected by summing up individual datapoints from spectrum images acquired over a large sample area. By this mean, the spectra with a high S/N ratio could be collected for accurate bandgap determination. Interestingly, bandgap increase was found at nanoplates with thickness below 10 nm, and thus confirming 1D confinement in nanosheets. More studies using EELS or EDX are foreseen to contribute to the studies of halide perovskites.

6. Fast-developing cryogenic TEM for halide perovskites

Cryogenic TEM (cryo-TEM) has been developed rapidly in



Fig. 6. (Color online) (a) Stacking faults observed in a MAPbl₃ with corresponding FFT patterns as inset (i), and corresponding magnified HRTEM (ii, v, vi) with structural model (iii) and (iv) the simulated HRTEM image^[96]. (b) Atomically resolved-cryo-TEM image of aged MAPbl₃ collected at a low dose condition (electron dose, ~5.96 $e/Å^2$), with corresponding enlargement (ii, iv), structural model (iii) and polarization map (v)^[40].

recent years. Beyond life science, its development has also offered opportunities to materials science, particularly to those too sensitive for conventional HRTEM and related spectroscopy^[94]. Growing applications are found for energy-related materials in past few years, and halide perovskites are benefitting from cryo-TEM as well^[95].

Li et al. combined the cryogenic temperature with direct electron detector dedicated for low-dose imaging, and successfully obtained HRTEM of MAPbl₃ and MAPbBr₃ under the dose of 12 e/Å² and 46 e/Å^{2[48]}. Zhu et al. investigated the MAPbl₃ single crystal at atomic scale using cryo-TEM, where electron dose rate was kept at ~47 e/(Å²·s) for about 2 s^[96]. Defects of surface steps and high density of stacking faults were revealed and correlated to the optoelectronic behavior of charge separation and carrier recombination (Fig. 6(a)). The same group further studied the formation of halide perovskites in the initial nucleation stage using low-dose imaging by cryo-TEM, as demonstrated on both MAPbl₃ and MAPbBr₃ respectively (Fig. 6(b)). A growth pattern from initially stacked slices towards a perfect single crystal structure was revealed, proposing a so-called twist-to-untwist structural evolution^[40, 97]. Low-temperature imaging was also applied to all-inorganic perovskites such as CsPbBr₃^[98]. The degradation mechanism was proposed as following: the migration and aggregation of Pb were suppressed at low temperature, and degradation was then dominated by the desorption of halogen ions induced by incident electrons, which led to decomposition products of CsBr, CsPb and PbBr₂ nanoparticles; further irradiation caused the formation and growth of pores and thus destroyed the structure. Additionally, it is interesting to mention that cryo-FIB has been applied to prepare specimen for atom probe tomography^[99]. More recently Zhou et al. reported cryo-FIB preparation of a MAPbl₃ single crystal film solar cell cross-section, enabling high quality microstructural investigation by electron diffraction^[100]. Damage-minimized sample preparation by cryo-FIB is thus expected to find more applications in the cross-section preparation of PSCs for more reliable microstructural investigation.

Nevertheless, it should be noted that the use of cryogenic temperature for halide perovskites imaging remains an open issue. It was reported by Rothmann *et al.* that low temperature could not alleviate irradiation damage but led to rapid amorphization of MAPbl₃^[46]. Similar results were reported on FAPbl₃ thin films, where both room-temperature and cryogenic conditions were performed for imaging, but beam sensitivity was not noticeably reduced in the cryogenic conditions^[67]. Therefore, detailed studies on the imaging conditions and degradation pathways in the cryogenic conditions are still needed towards further developments.

In-situ TEM dedicated to reveal degradation of halide perovskites under environmental conditions

Although much progress has been achieved in resolving the microstrure of as-grown halide perovskites or as-fabricated photovoltaic devices from nanoscale down to atomic scale, there is an urgent need to study the structural evolution of halide perovskites based devices under external stimuli such as light, heat and moisture so to understand the degradation of perovskites under environmental factors^[101–104]. *In-situ* TEM is thus considered as a powerful tool to this end, and has achieved significant progress in the past few years^[3, 105, 106].

The thermal stability of halide perovskites can be studied by *in-situ* heating in TEM. As demonstrated by Divitini *et al.*, the morphology and chemical composition change of MAPbl₃-based PSC during elavating temperature (up to



Fig. 7. (Color online) (a) *In-situ* heating of MAPbI₃ based PSCs up to 250 °C, where the temperature evolution of morphology change and elemental migration was monitored by HAADF-STEM images and EDX mappings. The same scale bar applies to all panels^[42]. (b) *In-situ* electrical biasing on MAPbI₃, where morphology and structure change was monitored by HAADF-STEM, TEM and SAED, respectively^[109]. (c) *In-situ* TEM showing the impact of controlled humidity on the conversion of MAPbI₃ into MAPbI₃·H₂O and finally PbI₂, using liquid cell^[112].

250 °C) was monitored using a FIB prepared cross-section (Fig. 7(a))^[42]. Migration of I and Pb was revealed and correlated to the synthesis condition. Seo *et al.* further compared the thermal responses of $Cs_x(FA_yMA_{(1-y)})_{(1-x)}Pb(I_zBr_{(1-z)})_3$ to MAPbI₃ PSCs using *in-situ* heating, and found that MAPbI₃ degraded rapidly by forming particles and voids, whereas FA-based PSCs is more heat-resistant. Similarly, the thermal stability of inorganic perovskites has been demonstrated on CsPb-Br₃, and found a superb stability up to 690 K^[107].

The impact of curent-voltage stimulus on perovskites degradation can be investigated by in-situ biasing. MAPbl₃based PSC was prepared into cross section and subjected to biasing in TEM by Jengros et al.[108]. Degradation was identified particularly at the positively biased MAPbl₃ interface, where iodide migrated into the transport layer and volatilized, resulting in the nucleation of Pbl₂ nanoparticles and voids. MAPbl₃ interface with TiO₂ as electron transport layer was further studied using in-situ biasing combined with HRTEM and STEM-EELS (Fig. 7(b))^[109]. It was found that positive bias led to oxygen diffusion into the MAPbl₃ layer accompanied with performance degradation, wheras a negative bias could remove oxygen ions from MAPbI₃ and partially restored the performance. Kim et al. observed amorphization of perovskite layer by in-situ biasing as induced by ion migration, whereas the bias-introduced degradation could be healed by heating at 50 °C^[110]. More recently, an *in-situ* electrical probing and light illumination through an optical fiber was applied to study the optoelectronic and photovoltaic properties on different crystal facets of CsPbBr₃^[111]. This work demonstrated a possibility of studying light-introduced degradation of perovskites.

Last but not least, the moisture-induced degradation was recently studied using in-operando liquid cell TEM (Fig. 7(c))^[112]. A live visualization from meso-to-nanoscale transfomration between prototypical MAPbl₃ and water molecules was presented: intermediate phase of MAPbl₃·H₂O was first formed upon exposure to humidity, and was subsequently decomposed into Pbl₂ crystals through dissolution/recrystallization. In addition, *in-situ* liquid cell TEM has been used to study the nucleation of pervoskites as demonstrated on MAPbl₃ as well, providing useful reference for understanding the intriguing chemical and physical properties of halide perovskites ^[113].

8. Conclusions and perspective

This review summarizes the recent progress of advanced TEM characterization for halide perovskites. The main conclusions are as following.

(1) Due to the presence of halogen ions and organic moieties, halide pervoskites are extremely sensitive to electron beam irradiation. Conventional TEM causes degradation in almost all halide perovskite structures, showing "artifacts" and damage such as the generation and expansion of holes, grain boundary cracking, and nanoparticle precipitation. Although the degradation mechanism under electron beam interaction remains unclear, it generally follows the pathway of "electron beam incidence \rightarrow organic moieties and/or halogen ion desorption \rightarrow defect generation and expansion \rightarrow lattice collapse and amorphization \rightarrow Pbl₂ or Pb precipitation". Both electron dose and dose rate have impact on the degradation. The understanding of degradation is essential, as it calls attention to the interpretation of TEM results and ultra-care in per-

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Table 1. Summary of TEM characterization details for MAPbl₃ and related structures. No damage or little damage was reported unless specified in the table.

| Perovskite chemistry | Specimen morphology | TEM method | Special characterization tools | Accelerating voltage | Dose (e/Ų) | Dose rate (e/ (Å ² ·s)) | Damage evaluation and additional note | Corresponding figure in this review | Year and reference |
|--|---------------------------------|-------------------------------------|--------------------------------------|----------------------|--------------------------|--|--|---|----------------------------|
| MAPbl ₃ | Freestanding thin film | SAED | | 200 | | 1 | | | 2017, [38] |
| MAPbl ₃ | Single crystal | SAED | | 300 | 151 73 4 | 0.5 1 2 | Upper dose limit at mentioned dose rate before damage | | 2018, [68] |
| MAPbl ₃ | Freestanding thin film | SAED | | 200 | <100 | 2 | Exposure > 60 s at the mentioned dose rate led to damage | 3(a) | 2018, [46] |
| MAPbl ₃ | Single crystal | SAED | | 300 | <99 | 1 | Ordered Vacancy was noticed at 145 e/(Å ² ·s) | 3(b) | 2020, [<mark>69</mark>] |
| | | | Cryogenic temperature | 300 | <30 | | Descendentific | | |
| $MAPbl_3$ | Single crystal | SAED | 25 | 300 | 30–41 | | Damaged at the dose of 475 e/(Å ² ·s) | | 2020, [<mark>7</mark> 1] |
| | | | 90 | 300 80 | 38 12 16 | | dose of 523 e/(Å ² ·s) | | |
| MA/FAPbl ₃ | Freestanding | SAED | | 200 | 15-10 | 2 | | | 2021, [74] |
| MAPbl | Single crystal | TFM | | 200 | 4-8 | 1 | | - | 2019 [50] |
| MAPbl ₃ | Freestanding thin film | TEM | | 200 | 9870 | | Damaged | 2(a) | 2019, [47] |
| MAPbl ₃ | Nanoparticles | TEM | Pulsed-beam TEM | 200 | 10 | 0.001 | | | 2020, [85] |
| | | | | | 66 | | Undamaged | 4(f) | |
| MAPbl₃/ FAPbl₃ | Freestanding thin film | LAADF- STEM | | 200 | 200 | | Change in structure was noticed | | 2020, [67] |
| | ••••• | 0.12.00 | | | 600 | | Damaged | | |
| MAPbl ₃ | Nanowire | HRTEM | Cryo-TEM | 300 | 12 | | 5 | | 2019, [48] |
| MAPbl ₃ | Nanoparticles | HRTEM | Cryo-TEM | 300 | <100 | | | 6(a) | 2020, [96] |
| MAPbl ₃ | Nanoplatelets | HRTEM | Cryo-TEM | 300 | 5.96 | 2 | Lattice changed noticed at accumulated dose of 23.64 e/Å ² | 6(b) | 2021, [40] |
| MAPbl ₃ | _ | HRTEM | Direct electron detector | 300 | 5.95 | | Superstructure was | | 2021, [81] |
| | | | | 300 | 1 | | MA loss at the mentioned dose | | |
| MAPbl ₃ | Single crystal | HRTEM | Direct electron detector | 300 | <28 | | MA loss reached a balanced state between ~10.5 and 28.0 e/Å ² | | 2021, [82] |
| MAPbl ₃ | FIB-milled PSC cross section | HAADF- STEM and STEM-EDX | <i>In-situ</i> heating ~150°C | 200 | | | | 7(a) | 2016, [42] |
| MAPbl ₃ | FIB-milled PSC cross section | HAADF- STEM, STEM-EDX | <i>In-situ</i> biasing | 300 | | | Beam current of 50 pA was used | | 2016, [<mark>108</mark>] |
| MAPbl ₃ | Nanoparticles | TEM | Liquid cell | 200 | | <90 | Dose rate of >125 e/(Å ² ·s) causes damage | | 2016, [113] |
| MAPbl ₃ | FIB-milled PSC cross section | TEM | In-situ biasing | 300 | 2–3 | 1–10 | | 7(b) | 2018, [<mark>109</mark>] |
| MAPbl ₃ | Nanoparticles | TEM | In-situ humidity | 200 | | 1.3 | | 7(c) | 2021, [112] |
| FA _{1-x} Cs _x Pbl ₃ FA _{0.5} Cs _{0.5} Pbl ₃ | FIB-milled PSC cross section | HAADF- STEM | | 300 | 1.3 × 10 ⁴ | | A thin conformal coating of amorphous carbon about 10 nm thick was deposited to avoid beam damage | 4(d) | 2022, [86] |
| Cs _{0.05} FA _{0.78} MA _{0.17} Pb (I _{0.83} Br _{0.17}) ₃ | Nanoparticles | Scanning electron diffraction | | 200 | 10 | | | 3(d) | 2021, [79] |
| $\begin{array}{c} Cs_{0.06}FA_{0.79} \\ MA_{0.15}Pb \\ (I_{0.85}Br_{0.15})_3 \end{array}$ | FIB-milled PSC cross section | STEM-EDX | | NA | >400 | | Damaged | | 2021, [91] |

Table 2. Summary of TEM characterization details for MAPbBr₃. No damage or little damage was reported unless specified in the table.

| Perovskite chemistry | Specimen morphology | TEM method | Special characteri- zation tool | Accelerating voltage | Dose (e/Ų) | Dose rate (e/(Å ² ·s)) | Damage evaluation and additional note | Corresponding figure in this review | Year and reference |
|-------------------------|------------------------|---------------|---------------------------------------|----------------------|---------------|---|---|---|--|
| MAPbBr ₃ | Single crystal | HRTEM | Direct electron detector | 300 | 6–12 | | | 4(b) | 2018, [<mark>39</mark>] |
| MAPbBr ₃ | _ | HRTEM | iDPC | 300 | 280 | | | | 2020, [81] |
| | | | | 300 | 163 | <1 | Formation of ordered vacancy | 3(b) | 2020, [<mark>69</mark>] |
| MAPbBr ₃ | Single crystal | SAED | | 300 | 63–113 | <1 | | | |
| | | | Cryogenic temperature | 300 | 81 | <1 | | | ⁻ 2020, [<mark>7</mark> 1] |
| MAPbBr ₃ | Nanowire | HRTEM | Cryo-TEM | | 46 | | | | 2019, [<mark>48</mark>] |
| MAPbBr ₃ | Nanoplatelets | HRTEM | Cryo-TEM | 300 | 4–8 | 4 | | | 2021, [<mark>97</mark>] |

Table 3. Summary of TEM characterization details for all-inorganic CsPbBr₃ and related structures. No damage or little damage was reported unless specified in the table.

| Perovskite chemistry | Specimen morphology | TEM method | Special characteri- zation tool | Accelerating voltage | Dose (e/Ų) | Dose rate (e/(Ų·s)) | Damage evaluation and additional note | Correspondi ng figure in this review | Year and reference |
|--|------------------------|------------------|---|----------------------|-------------------|---------------------------|--|--|--|
| CsPbBr ₃ | Nanosheets | HRTEM | In-line holography, imaging series | 80 | ~8000 | 100 | | 4(a) | 2016, [41] |
| | | | acquisition | 80 | | 3800 | | | |
| | | HRTEM | <-40 °C | 200 | 500 | | | | |
| CsPbBr ₃ | Nanosheets | HAADF- STEM | <-40 °C | 200 | <1800 | | | | [_] 2017, [<mark>98</mark>] |
| CsPbBr ₃ | Nanosheets | HRTEM | · | 200 | 1140 | | Pb formation | | 2017, [56] |
| CsPbBr ₃ | Nanosheets | CBED | Cryogenic temperature | 200 | | | | 4(c) | _ |
| | | Ptycho graphy | Direct electron detector | 300 | 1000 | | | | 2018, [<mark>84</mark>] |
| CsPbBr ₃ | Nanoparticles | TEM | | 200 | 100-1000 | | | | 2020, [111] |
| CsPbBr ₃ | Nanosheets | EELS | | 60 | | | The beam current was set to <5 pA | | 2020, [<mark>93</mark>] |
| CsPbBr ₃ | Nanoparticles | STEM-EDX | | 300 | 9×10 ⁴ | | Cyanamide passivation to protect surface from damage | 5(a)/5(b) | 2020, [81] 2021, [92] |
| CsPb (Br _{0.8} I _{0.2}) ₃ | Nanoparticles | HRTEM | | 0 | | 500 | Damaged | | 2020, [59] |
| γ-CsPblBr ₂ | Nanoparticles | HAADF- STEM | | 300 | 6000-8000 | | | | 2021, [72] |
| CsPbl ₃ | Freestanding thin film | TEM | | 200 | 7800 | | Damaged | | 2019, [47] |

forming TEM characterization.

(2) In order to reduce the irradiation damage, low-dose is vital for halide perovskites' characterization. Electron diffraction has shown to be powerful in resolving the crystal structure and microstructure at low dose. In Table 1–4, we have summarized the imaging conditions as reported in this review according to perovskite chemistry, i.e. MAPbl₃, MAPbBr₃, all-inorganic CsPbBr₃/CsPbl₃, and 2D halide perovskites, as an imaging guide for researchers in this field. More importantly, safe dose and damage dose for different perovskite materials are summarized in Fig. 8, where dose limit for imaging is better represented. 2D perovskites are most sensitive to elec-

tron dose, whereas MAPbI₃ and MAPbBr₃ is slightly more robust, and CsPbBr₃ has the best tolerance to beam irradiation. It is important to note that dose rate and accelerating voltage also has influence on imaging, and surface treatment to protect the sample can drastically increase the critical dose. Readers are encouraged to refer to the tables and corresponding references for specific notes. Undoubtedly, halide perovskites is one of the most beam-sensitive materials, and researchers are encouraged to carefully design and prepare their TEM experiments in order to achieve meaningful results.

(3) Emerging techniques have been developed to en-

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Table 4. Summary of TEM characterization details for 2D halide perovskites. No damage or little damage was reported unless specified in the table.

| Perovskite chemistry | Specimen morphology | TEM method | Special characteri- zation tool | Accelerating voltage | Dose (e/Ų) | Dose rate (e/(Ų·s)) | Damage evaluation and additional note | Correspondi ng figure in this review | Year and reference |
|--|------------------------|---------------|---------------------------------------|----------------------|---------------|------------------------|---|--|---------------------------|
| (2T) ₂ PbI ₄ – (2T) ₂ PbBr ₄ | Nanosheets | HRTEM | | 80 | 9.2 | 65 | Imaging series of 40 ms/frame. Damaged within 1 s | 4(e) | 2020, [87] |
| $BA_2MA_2Pb_3I_{10}$ (n = 3) | Nanosheets | HRTEM | Direct electron detector | 300 | >104 | | Surface damaged with reconstruction | | 2019, [<mark>88</mark>] |
| BA ₂ PbBr ₄ | Nanosheets | SAED | | 300 | | <1 | <1 (spot size 6) | 2(c) | 2019, [53] |
| BA ₂ FAPb ₂ I ₇ / BA ₂ MAPb ₂ I ₇ | Nanosheets | HRTEM | | 200 | <50 | | | | 2020, [73] |



Fig. 8. (Color online) Illustrated summary of safe dose and damage dose for different perovskite materials, plotted in coloured columns. Numbers in the figure correspond to the reference numbers as listed in the tables and references. Shade in each column indicates relatively-safe dose range versus damage-prone dose range. Generally speaking, 2D pvsk is suggested to be imaged below the dose of 50 e/(Å²·s), MAPbl₃/ MAPbBr₃ below ~100 e/(Å²·s), whereas CsPbBr₃ can tolerate dose up to more than 1000 e/(Å²·s).

able atomic-resolution imaging and chemical mapping of halide pervoskites. Low-dose imaging techniques such as exitwave reconstruction, LAADF-STEM and ptychography has successfully imaged perovskites at atomic scale. The development of hardware such as direct electron detectors have also succeeded in providing atomically resolved structure of halide perovskites. Additionally, the rapid development of cryo-TEM has emerged as a powerful tool to study the halide perovskites and reveal defects such as grain boundaries and surface steps.

(4) *In-situ* TEM under heating, biasing and liquid cell is being developed and applied to study the degradation of halide perovskites under operando conditions. Microstructural change has been revealed with unprecedented details.

Thanks to the rapid developments of emerging TEM techniques, halide perovskites have been resolved down to atomic scale with unprecedented details. Perspectives of TEM characterizations for halide perovskites are proposed as following:

(1) Sample preparation. Imaging techniques such as iD-PC, ptychography, and the developments of detectors have paved ways for low-dose imaging, but the bottleneck is sample preparation, particularly for PSC thin films. Conventionally, FIB is used to prepare cross sections, but beam damage of Ga⁺ ions can induce significant damage to the sample prior to imaging. The use of cryo-FIB and surface passivation has shown their advantages in alleviating beam damage, and more studies are encouraged.

(2) Development of emerging techniques. Although lowdose imaging and *in-situ* TEM has been demonstrated to reveal the structure-property relationship, the studies are still limited to a few examples. More investigations on different materials and devices are expected.

(3) Interpretation of TEM data. Degradation mechanism has been discussed in this review, emphasizing the impact of electron dose/dose rate on halide perovskite structure. It is clear that only low dose imaging can reveal the halide perovskite structure, although conventional TEM imaging and EDX/EELS study involves large dose, which significantly exceeds damage threshold. Therefore, we call attentions to the interpretation of TEM data as reported in literatures, where attentions should be paid to the imaging condition before correlation to the conclusions.

Halide perovskites are strategically important in the field of energy materials. Along with the rapid development of the materials and related devices, there is an urgent need to understand the structure-property relationship from nanoscale to atomic scale. Much effort has been made in the past few years to address the beam-sensitivity of this particular materials, and more applications of emerging TEM characterizations are foreseen in the coming future, unveiling the structural origin of halide perovskite's unique properties and degradation mechanism under operando conditions, so to assist the design of a more efficient and robust energy material.

Acknowledgements

X.K. and M.S. acknowledge the Beijing Municipal High Level Innovative Team Building Program (IDHT20190503) and the National Natural Science Fund for Innovative Research Groups of China (51621003). X.W. and X.K. acknowledge the National Natural Science Foundation of China (12074017).

Chem Lett, 2017, 8, 772

- References
- Jiang Q, Zhao Y, Zhang X W, et al. Surface passivation of perovskite film for efficient solar cells. Nat Photonics, 2019, 13, 460
- [2] Lin Y W, Lin G M, Sun B Y, et al. Nanocrystalline perovskite hybrid photodetectors with high performance in almost every figure of merit. Adv Funct Mater, 2018, 28, 1705589
- [3] Ran J H, Dyck O, Wang X Z, et al. Electron-beam-related studies of halide perovskites: Challenges and opportunities. Adv Energy Mater, 2020, 10, 1903191
- [4] Chen P F, Ong W J, Shi Z H, et al. Pb-based halide perovskites: Recent advances in photo(electro)catalytic applications and looking beyond. Adv Funct Mater, 2020, 30, 1909667
- [5] Ye T, Pan L, Yang Y, et al. Synthesis of highly-oriented black CsPbl₃ microstructures for high-performance solar cells. Chem Mater, 2020, 32, 3235
- [6] Green M A, Ho-Baillie A, Snaith H J. The emergence of perovskite solar cells. Nat Photonics, 2014, 8, 506
- [7] Dong Q, Fang Y, Shao Y, et al. Electron-hole diffusion lengths > 175 μm in solution-grown CH₃NH₃PbI₃ single crystals. Science, 2015, 347, 967
- [8] Shi Z J, Guo J, Chen Y H, et al. Lead-free organic-inorganic hybrid perovskites for photovoltaic applications: Recent advances and perspectives. Adv Mater, 2017, 29, 1605005
- [9] Yang Z B, Rajagopal A, Jen A K Y. Ideal bandgap organic–inorganic hybrid perovskite solar cells. Adv Mater, 2017, 29, 1704418
- [10] Kim M, Jeong J, Lu H Z, et al. Conformal quantum dot-SnO₂ layers as electron transporters for efficient perovskite solar cells. Science, 2022, 375, 302
- [11] Liu C, Sun J, Tan W L, et al. Alkali cation doping for improving the structural stability of 2D perovskite in 3D/2D PSCs. Nano Lett, 2020, 20, 1240
- [12] Xie F X, Chen C C, Wu Y Z, et al. Vertical recrystallization for highly efficient and stable formamidinium-based inverted-structure perovskite solar cells. Energy Environ Sci, 2017, 10, 1942
- [13] Xiang W C, Wang Z W, Kubicki D J, et al. Europium-doped CsPbl₂Br for stable and highly efficient inorganic perovskite solar cells. Joule, 2019, 3, 205
- [14] Yang S, Niu W X, Wang A L, et al. Ultrathin two-dimensional organic-inorganic hybrid perovskite nanosheets with bright, tunable photoluminescence and high stability. Angew Chem Int Ed, 2017, 56, 4252
- [15] Sun Y, Yin Y, Pols M, et al. Engineering the phases and heterostructures of ultrathin hybrid perovskite nanosheets. Adv Mater, 2020, 32, 2002392
- [16] Su Y, Chen X J, Ji W Y, et al. Highly controllable and efficient synthesis of mixed-halide CsPbX₃ (X = Cl, Br, I) perovskite QDs toward the tunability of entire visible light. ACS Appl Mater Interfaces, 2017, 9, 33020
- [17] Utzat H, Sun W W, Kaplan A E K, et al. Coherent single-photon emission from colloidal lead halide perovskite quantum dots. Science, 2019, 363, 1068
- [18] Filip M R, Hillman S, Haghighirad A A, et al. Band gaps of the lead-free halide double perovskites Cs₂BiAgCl₆ and Cs₂BiAgBr₆ from theory and experiment. J Phys Chem Lett, 2016, 7, 2579
- [19] Zhong H X, Yang M, Tang G, et al. Type-II lateral heterostructures of monolayer halide double perovskites for optoelectronic applications. ACS Energy Lett, 2020, 5, 2275
- [20] McClure E T, Ball M R, Windl W, et al. Cs₂AgBiX₆ (X = Br, Cl): New visible light absorbing, lead-free halide perovskite semiconductors. Chem Mater, 2016, 28, 1348
- [21] Wu C C, Zhang Q H, Liu Y, et al. The dawn of lead-free perovskite solar cell: Highly stable double perovskite Cs₂AgBiBr₆ film. Adv Sci, 2018, 5, 1700759
- [22] Volonakis G, Haghighirad A A, Milot R L, et al. Cs₂InAgCl₆: A new lead-free halide double perovskite with direct band gap. J Phys

[23] Li Z Z, Yin W J. Recent progress in Pb-free stable inorganic double halide perovskites. J Semicond, 2018, 39, 071003

- [24] Xiao Z W, Yan Y F. Progress in theoretical study of metal halide perovskite solar cell materials. Adv Energy Mater, 2017, 7, 1701136
- [25] Yang Y, Sun Y B, Jiang Y S. Structure and photocatalytic property of perovskite and perovskite-related compounds. Mater Chem Phys, 2006, 96, 234
- [26] Zhang H, Fu X, Tang Y, et al. Phase segregation due to ion migration in all-inorganic mixed-halide perovskite nanocrystals. Nat Commun, 2019, 10, 1088
- [27] Huang B Y, Liu Z H, Wu C W, et al. Polar or nonpolar? That is not the question for perovskite solar cells. Natl Sci Rev, 2021, 8, nwab094
- [28] Lei Y T, Xu Y K, Wang M, et al. Origin, influence, and countermeasures of defects in perovskite solar cells. Small, 2021, 17, 2005495
- [29] Wu J P, Liu S C, Li Z B, et al. Strain in perovskite solar cells: Origins, impacts and regulation. Natl Sci Rev, 2021, 8, nwab047
- [30] Sin C K, Zhang J Z, Tse K, et al. A brief review of formation energies calculation of surfaces and edges in semiconductors. J Semicond, 2020, 41, 061101
- [31] Bhattacharya S, Chandra G K, Predeep P. A microstructural analysis of 2D halide perovskites: Stability and functionality. Front Nanotechnol, 2021, 3, 657948
- [32] Kim T W, Park N G. Methodologies for structural investigations of organic lead halide perovskites. Mater Today, 2020, 38, 67
- [33] Kumar V, Nisika, Kumar M. Temporal-spatial-energy resolved advance multidimensional techniques to probe photovoltaic materials from atomistic viewpoint for next-generation energy solutions. Energy Environ Sci, 2021, 14, 4760
- [34] Liu J J. Advances and applications of atomic-resolution scanning transmission electron microscopy. Microsc Microan, 2021, 27, 943
- [35] Ribet S M, Murthy A A, Roth E W, et al. Making the most of your electrons: Challenges and opportunities in characterizing hybrid interfaces with STEM. Mater Today, 2021, 50, 100
- [36] Zha F X, Zhang Q Y, Dai H G, et al. The scanning tunneling microscopy and spectroscopy of GaSb_{1-x}Bi_x films of a few-nanometer thickness grown by molecular beam epitaxy. J Semicond, 2021, 42, 092101
- [37] Yang Z, Liu S Z. Perspective on the imaging device based on perovskite materials. J Semicond, 2020, 41, 050401
- [38] Rothmann M U, Li W, Zhu Y, et al. Direct observation of intrinsic twin domains in tetragonal CH₃NH₃Pbl₃. Nat Commun, 2017, 8, 14547
- [39] Zhang D L, Zhu Y H, Liu L M, et al. Atomic-resolution transmission electron microscopy of electron beam-sensitive crystalline materials. Science, 2018, 359, 675
- [40] Zhu Y, Wang S, Li B, et al. Twist-to-untwist evolution and cation polarization behavior of hybrid halide perovskite nanoplatelets revealed by cryogenic transmission electron microscopy. J Phys Chem Lett, 2021, 12, 12187-95
- [41] Yu Y, Zhang D D, Kisielowski C, et al. Atomic resolution imaging of halide perovskites. Nano Lett, 2016, 16, 7530
- [42] Divitini G, Cacovich S, Matteocci F, et al. *In situ* observation of heat-induced degradation of perovskite solar cells. Nat Energy, 2016, 1, 15012
- [43] Seo Y H, Kim J H, Kim D H, et al. *In situ* TEM observation of the heat-induced degradation of single- and triple-cation planar perovskite solar cells. Nano Energy, 2020, 77, 105164
- [44] Ge Y, Mu X L, Lu Y, et al. Photoinduced degradation of lead halide perovskite thin films in air. Acta Phys Chim Sin, 2020, 36, 1905039
- [45] Rothmann M U, Li W, Etheridge J, et al. Microstructural characterisations of perovskite solar cells - from grains to interfaces: Tech-

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niques, features, and challenges. Adv Energy Mater, 2017, 7, 1700912

- [46] Rothmann M U, Li W, Zhu Y, et al. Structural and chemical changes to CH₃NH₃Pbl₃ induced by electron and gallium ion beams. Adv Mater, 2018, 30, 1800629
- [47] Chen X Y, Wang Z W. Investigating chemical and structural instabilities of lead halide perovskite induced by electron beam irradiation. Micron, 2019, 116, 73
- [48] Li Y B, Zhou W J, Li Y Z, et al. Unravelling degradation mechanisms and atomic structure of organic-inorganic halide perovskites by cryo-EM. Joule, 2019, 3, 2854
- [49] Kim T W, Kondo T. Direction-selective electron beam damage to CH₃NH₃Pbl₃ based on crystallographic anisotropy. Appl Phys Express, 2020, 13, 091001
- [50] Alberti A, Bongiorno C, Smecca E, et al. Pb clustering and Pbl₂ nanofragmentation during methylammonium lead iodide perovskite degradation. Nat Commun, 2019, 10, 2196
- [51] Manekkathodi A, Marzouk A, Ponraj J, et al. Observation of structural phase transitions and Pbl₂ formation during the degradation of triple-cation double-halide perovskites. ACS Appl Energy Mater, 2020, 3, 6302
- [52] Dou L T, Wong A B, Yu Y, et al. Atomically thin two-dimensional organic-inorganic hybrid perovskites. Science, 2015, 349, 1518
- [53] Nie L F, Ke X X, Sui M L. Microstructural study of two-dimensionalorganic-inorganichybrid perovskite nanosheet degradation under illumination. Nanomaterials, 2019, 9, 722
- [54] Li F, Liu Y, Wang H, et al. Postsynthetic surface trap removal of CsPbX₃ (X = Cl, Br, or I) quantum dots via a ZnX_2 /hexane solution toward an enhanced luminescence quantum yield. Chem Mater, 2018, 30, 8546
- [55] Su G D, He B L, Gong Z K, et al. Enhanced charge extraction in carbon-based all-inorganic CsPbBr₃ perovskite solar cells by dualfunction interface engineering. Electrochim Acta, 2019, 328, 135102
- [56] Dang Z Y, Shamsi J, Palazon F, et al. *In situ* transmission electron microscopy study of electron beam-induced transformations in colloidal cesium lead halide perovskite nanocrystals. ACS Nano, 2017, 11, 2124
- [57] Zou S H, Liu C P, Li R F, et al. From nonluminescent to blue-emitting Cs₄PbBr₆ nanocrystals: Tailoring the insulator bandgap of 0D perovskite through Sn cation doping. Adv Mater, 2019, 31, 1900606
- [58] Wang T, Yang Z, Yang L, et al. Atomic-scale insights into the dynamics of growth and degradation of all-inorganic perovskite nanocrystals. J Phys Chem Lett, 2020, 11, 4618
- [59] Funk H, Shargaieva O, Eljarrat A, et al. *In situ* TEM monitoring of phase-segregation in inorganic mixed halide perovskite. J Phys Chem Lett, 2020, 11, 4945
- [61] Creutz S E, Crites E N, de Siena M C, et al. Colloidal nanocrystals of lead-free double-perovskite (elpasolite) semiconductors: Synthesis and anion exchange to access new materials. Nano Lett, 2018, 18, 1118
- [62] Feng Y H, Ke X X, Sui M L. Effect of electron irradiation on inorganic double perovskite solar cell material Cs₂AgBiBr₆. J Chin Electron Microsc Soc, 2020, 39, 1
- [63] Egerton R F, Li P, Malac M. Radiation damage in the TEM and SEM. Micron, 2004, 35, 399
- [64] Gong Z L, Yang Y. The application of synchrotron X-ray techniques to the study of rechargeable batteries. J Energy Chem, 2018, 27, 1566
- [65] Cai Z H, Wu Y N, Chen S Y. Energy-dependent knock-on damage of organic-inorganic hybrid perovskites under electron beam irradiation: First-principles insights. Appl Phys Lett, 2021,

119, 123901

- [66] Chen Z X, Ke X X, Zhu L J, et al. Electron microscopy of organic-inorganic hybrid perovskite solar cell materials: degradation mechanism study and imaging condition optimization. J Chin Electron Microsc Soc, 2019, 38, 15
- [67] Rothmann M U, Kim J S, Borchert J, et al. Atomic-scale microstructure of metal halide perovskite. Science, 2020, 370, 6516
- [68] Chen S, Zhang X, Zhao J, et al. Atomic scale insights into structure instability and decomposition pathway of methylammonium lead iodide perovskite. Nat Commun, 2018, 9, 4807
- [69] Chen S L, Zhang Y, Zhang X W, et al. General decomposition pathway of organic-inorganic hybrid perovskites through an intermediate superstructure and its suppression mechanism. Adv Mater, 2020, 32, 2001107
- [70] Chen S L, Gao P. Challenges, myths, and opportunities of electron microscopy on halide perovskites. J Appl Phys, 2020, 128, 010901
- [71] Chen S L, Zhang Y, Zhao J J, et al. Transmission electron microscopy of organic-inorganic hybrid perovskites: Myths and truths. Sci Bull, 2020, 65, 1643
- [72] Zhou X G, Yang C Q, Sang X, et al. Probing the electron beam-induced structural evolution of halide perovskite thin films by scanning transmission electron microscopy. J Phys Chem C, 2021, 125, 10786
- [73] Yuan B, Shi E Z, Liang C, et al. Structural damage of two-dimensional organic–inorganic halide perovskites. Inorganics, 2020, 8, 13
- [74] Li W, Rothmann M U, Zhu Y, et al. The critical role of composition-dependent intragrain planar defects in the performance of MA_{1-x}FA_xPbl₃ perovskite solar cells. Nat Energy, 2021, 6, 624
- [75] Gao Y, Shi E, Deng S, et al. Molecular engineering of organic–inorganic hybrid perovskites quantum wells. Nat Chem, 2019, 11, 1151
- [76] Pan W, Wu H, Luo J, et al. Cs₂AgBiBr₆ single-crystal X-ray detectors with a low detection limit. Nat Photonics, 2017, 11, 726
- [77] Luo J, Wang X, Li S, et al. Efficient and stable emission of warmwhite light from lead-free halide double perovskites. Nature, 2018, 563, 541
- [78] Pham H T, Yin Y T, Andersson G, et al. Unraveling the influence of CsCl/MACI on the formation of nanotwins, stacking faults and cubic supercell structure in FA-based perovskite solar cells. Nano Energy, 2021, 87, 106226
- [79] Doherty T A S, Nagane S, Kubicki D J, et al. Stabilized tilted-octahedra halide perovskites inhibit local formation of performancelimiting phases. Science, 2021, 374, 1598
- [80] Brennan M C, Kuno M, Rouvimov S. Crystal structure of individual CsPbBr₃ perovskite nanocubes. Inorg Chem, 2019, 58, 1555
- [81] Song K P, Liu L M, Zhang D L, et al. Atomic-resolution imaging of halide perovskites using electron microscopy. Adv Energy Mater, 2020, 10, 1904006
- [82] Chen S, Wu C, Han B, et al. Atomic-scale imaging of CH₃NH₃Pbl₃ structure and its decomposition pathway. Nat Commun, 2021, 12, 5516
- [83] Qiao G Y, Guan D H, Yuan S, et al. Perovskite quantum dots encapsulated in a mesoporous metal-organic framework as synergistic photocathode materials. J Am Chem Soc, 2021, 143, 14253
- [84] dos Reis R, Yang H, Ophus C, et al. Determination of the structural phase and octahedral rotation angle in halide perovskites. Appl Phys Lett, 2018, 112, 071901
- [85] VandenBussche E J, Clark C P, Holmes R J, et al. Mitigating damage to hybrid perovskites using pulsed-beam TEM. ACS Omega, 2020, 5, 31867
- [86] Cai S H, Dai J, Shao Z P, et al. Atomically resolved electrically active intragrain interfaces in perovskite semiconductors. J Am Chem Soc, 2022, 144, 1910
- [87] Shi E, Yuan B, Shiring S B, et al. Two-dimensional halide per-

ovskite lateral epitaxial heterostructures. Nature, 2020, 580, 614

- [88] Jung H J, Stompus C C, Kanatzidis M G, et al. Self-passivation of 2D ruddlesden-popper perovskite by polytypic surface Pbl₂ encapsulation. Nano Lett, 2019, 19, 6109
- [89] Yu Y, Zhang D D, Yang P D. Ruddlesden-popper phase in two-dimensional inorganic halide perovskites: A plausible model and the supporting observations. Nano Lett, 2017, 17, 5489
- [90] Dang Z Y, Dhanabalan B, Castelli A, et al. Temperature-driven transformation of CsPbBr₃ nanoplatelets into mosaic nanotiles in solution through self-assembly. Nano Lett, 2020, 20, 1808
- [91] Kosasih F U, Cacovich S, Divitini G, et al. Nanometric chemical analysis of beam-sensitive materials: A case study of STEM-EDX on perovskite solar cells. Small Methods, 2021, 5, 2000835
- [92] Liu J, Song K, Zheng X, et al. Cyanamide passivation enables robust elemental imaging of metal halide perovskites at atomic resolution. J Phys Chem Lett, 2021, 12, 10402
- [93] Brescia R, Toso S, Ramasse Q, et al. Bandgap determination from individual orthorhombic thin cesium lead bromide nanosheets by electron energy-loss spectroscopy. Nanoscale Horiz, 2020, 5, 1610
- [94] Li Y B, Huang W, Li Y Z, et al. Opportunities for cryogenic electron microscopy in materials science and nanoscience. ACS Nano, 2020, 14, 9263
- [95] Zhang Z W, Cui Y, Vila R, et al. Cryogenic electron microscopy for energy materials. Acc Chem Res, 2021, 54, 3505
- [96] Zhu Y M, Gui Z G, Wang Q, et al. Direct atomic scale characterization of the surface structure and planar defects in the organic-inorganic hybrid CH₃NH₃Pbl₃ by Cryo-TEM. Nano Energy, 2020, 73, 104820
- [97] Zhu Y M, Zhang Q, Yang X M, et al. Probing atomic structure of beam-sensitive energy materials in their native states using cryogenic transmission electron microscopes. iScience, 2021, 24, 103385
- [98] Dang Z Y, Shamsi J, Akkerman Q A, et al. Low-temperature electron beam-induced transformations of cesium lead halide perovskite nanocrystals. ACS Omega, 2017, 2, 5660
- [99] Rivas N A, Babayigit A, Conings B, et al. Cryo-focused ion beam preparation of perovskite based solar cells for atom probe tomography. PLoS One, 2020, 15, e0227920
- [100] Zhou J F, Wei N N, Zhang D L, et al. Cryogenic focused ion beam enables atomic-resolution imaging of local structures in highly sensitive bulk crystals and devices. J Am Chem Soc, 2022, 144, 3182
- [101] Lee J W, Seo S, Nandi P, et al. Dynamic structural property of organic-inorganic metal halide perovskite. iScience, 2021, 24, 101959
- [102] Stranks S D. Multimodal microscopy characterization of halide perovskite semiconductors: Revealing a new world (dis)order. Matter, 2021, 4, 3852
- [103] Thampy S, Xu W J, Hsu J W P. Metal oxide-induced instability and its mitigation in halide perovskite solar cells. J Phys Chem Lett, 2021, 12, 8495
- [104] Zhang C C, Yuan S, Lou Y H, et al. Physical fields manipulation for high-performance perovskite photovoltaics. Small, 2022, 2107556
- [105] Kosasih F U, Ducati C. Characterising degradation of perovskite solar cells through *in situ* and operando electron microscopy. Nano Energy, 2018, 47, 243
- [106] Kundu S, Kelly T L. In situ studies of the degradation mechanisms of perovskite solar cells. EcoMat, 2020, 2, e12025
- [107] McGrath F, Ghorpade U V, Ryan K M. Synthesis and dimensional control of CsPbBr₃ perovskite nanocrystals using phosphorous based ligands. J Chem Phys, 2020, 152, 174702

- [108] Jeangros Q, Duchamp M, Werner J, et al. *In situ* TEM analysis of organic-inorganic metal-halide perovskite solar cells under electrical bias. Nano Lett, 2016, 16, 7013
- [109] Jung H J, Kim D, Kim S, et al. Stability of halide perovskite solar cell devices: *in situ* observation of oxygen diffusion under biasing. Adv Mater, 2018, 30, 1802769
- [110] Kim M C, Ahn N, Cheng D Y, et al. Imaging real-time amorphization of hybrid perovskite solar cells under electrical biasing. ACS Energy Lett, 2021, 6, 3530
- [111] Zhang C, Fernando J F S, Firestein K L, et al. Crystallography-derived optoelectronic and photovoltaic properties of CsPbBr₃ perovskite single crystals as revealed by *in situ* transmission electron microscopy. Appl Mater Today, 2020, 20, 100788
- [112] Akhavan Kazemi M A, Raval P, Cherednichekno K, et al. Molecular-level insight into correlation between surface defects and stability of methylammonium lead halide perovskite under controlled humidity. Small Methods, 2021, 5, 2000834
- [113] Qin F Y, Wang Z W, Wang Z L. Anomalous growth and coalescence dynamics of hybrid perovskite nanoparticles observed by liquid-cell transmission electron microscopy. ACS Nano, 2016, 10, 9787



Xiaomei Wu got her BS and master's degrees in 2018 and 2021 at Beijing Technology and Business University. Now she is a doctoral student at Beijing University of Technology. Her research focuses on energy materials microstructure characterization using TEM.



Xiaoxing Ke is an associate professor at Beijing University of Technology (BJUT), China. She received her Ph.D. degree from Physics Department, University of Antwerp, Belgium in 2010. After 4 years of postdoctoral fellowship in EMAT, University of Antwerp, she joined BJUT in 2014. Her research interests mainly focus on microstructure investigation on energy materials using advanced transmission electron microscopy (TEM), including aberration-corrected (S)TEM and 3D electron tomography.



Manling Sui is a Distinguished Professor at Beijing University of Technology (BJUT). She received her Ph.D. in 1991 at Institute of Metal Research, Chinese Academy of Sciences, and successively worked at Northeastern University in China, University of Wisconsin-Madison in the United States, and Institute of Metal Research, Chinese Academy of Sciences. Then she joined BJUT as Distinguished Professor of Cheung Kong Scholars Programme in 2009. Her research is focused on the structure and property relationships of advanced materials at atomic scale, especially using in situ transmission electron microscopy with external stimuli such as thermal, electrical, mechanical, optical, and liquid/gas environments.