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Spectroscopic and Yellow Laser Features of Dy³⁺: Y₃Al₅O₁₂ Single Crystals

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Abstract: In recent years, yellow laser crystals have raised great attentions owing to their comprehensive applications in the fields such as laser display, laser medical treatment, light detection and ranging (LIDAR), Bose-Einstein condensates, and atomic cooling and trapping. With the development of commercial blue light LD, the direct pumping of Dy^{3+} doped laser crystals has realized yellow laser based on its transition ${}^{4}F_{9/2} \rightarrow {}^{6}H_{13/2}$. In this work, Dy3+: Y3Al5O12 (Dy: YAG) crystals with 0.5%, 1.0%, 2.0%, 3.0%, and 4.0% (atomic fraction) nominal concentration of Dy³⁺ were grown using Czochralski method, the reason of crystal crack was discussed. Based on Judd-Ofelt (J-O) theory, the J-O intensity parameters and utilization, and other laser parameters of Dy: YAG crystals with different doping concentrations were evaluated. The effect of the doping concentration of Dy³⁺ on the spectroscopic performances like fluorescence branching ratio, stimulated emission cross-section, quantum efficiency, were analyzed comprehensively. Among all the five crystals, 1.0% Dy: YAG has the largest stimulated emission cross-section for 582 nm yellow emission, an intense fluorescence intensity with the 447 nm excitation, and a longer decay time of 0.823 ms. The fluorescence intensity and stimulated emission cross-section of 2.0% Dy: YAG are slightly less than that of 1.0% Dy: YAG, but the former has a higher absorption coefficient. Hence, the spectroscopic analysis results show that 1.0% and 2.0% are the suitable concentrations of Dy^{3+} ion in YAG crystal for yellow laser operation by diode pumping. The continuous wave laser with peak at 582.5 nm and the maximum output power of 166.8 µW yellow laser operation were realized in 2.0% Dy: YAG crystal.

Key words: YAG crystal; Dy³⁺; crystal growth; fluorescence features; yellow laser

The interest in Dy^{3^+} doped optical materials is rapidly growing recently, and more and more researches about different Dy^{3^+} -containing materials were reported^[1-5]. Dy^{3^+} doped luminescent materials in the visible region are burgeoning due to peculiar emissions which differ from other RE (rare-earth) ions doped materials. Dy^{3^+} is also the only RE ion with yellow emission in addition to Tb^{3^+} . Because of the strong yellow emission intensity which is several times of Tb^{3^+} doping at the same doping concentration and wide emission band from 565 nm to 595 nm, Dy^{3^+} doped single crystals are promising in the field of all-solid-state laser^[6]. Generally, a yellow laser is used for the treatment of fundus diseases and dermatological disorders in medical treatments^[7-8]. 578 nm laser can be used for a Yb optical clock that matches the ${}^{1}S_{0}{}^{-3}P_{0}$ transition of the Yb atoms, and 589 nm yellow laser can be used as a sodium beacon laser^[9-10]. Dy³⁺ shows absorption bands in both blue and UV (ultraviolet) regions that match the InGaN blue LD (laser diode). The yellow laser operation using Dy³⁺ doped single crystal was reported for the first time in 2012 and gives confidence to the researchers to focus on the yellow laser based on Dy³⁺ ion-doped crystals^[11].

In recent years, all types of Dy^{3+} doped single crystals are reported, which provide a sufficient number of references for comparison of the performances among

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different host materials^[12-17]. Unfortunately, to our best knowledge, up to now yellow laser output using Dy³⁺ activated bulk crystal (not fiber laser) as a gain medium was achieved successfully only in several host materials^[11, 18-20]. $Y_3Al_5O_{12}$ (YAG) is a typical laser crystal with high mechanical strength which cannot be broken down easily during laser experiments. The higher phonon energy of YAG makes the population on ⁶H_{13/2} level could greater extent transfer to ground state ${}^{6}H_{15/2}$ by non-radiative transitions. It is relatively easy to grow YAG crystal using the Czochralski method on account of the congruent melting property. It also has a low cost due to the relatively cheaper raw materials Y₂O₃ and Al₂O₃. The spectroscopic characteristics of 3.0% Dy³⁺: YAG crystal under 384 nm excitation were investigated by Pan, et $al^{[21]}$. The spectroscopic properties of 3.0% Dy³⁺: YAG crystal grown by the micro-pulling-down method were reported by Xu^[22]. In addition, Dy³⁺ doped single crystal could be applied in the display and lighting areas on account of the strong blue and yellow emissions in visible waveband, which makes the white emitting available. Yu, et $al^{[23]}$ grew the Gd^{3+}/Dy^{3+} co-doped CaF_2 crystal by the Bridgeman method and the tunable white light were obtained by changing the concentration of Gd³⁺. The research from Xu, et al^[24] showed Dy³⁺/Eu³⁺ co-doped LiLuF₄ single crystal had good optical features and thermal stability. A white light emitting under 355 nm excitation in the Dy^{3+} : $Gd_3Sc_2Al_3O_{12}$ crystal was demonstrated by Ding, et $al^{[25]}$. The 366 nm absorption band of Dy³⁺: Y₃Al₅O₁₂ (Dy: YAG) matches the excitation wavelength of commercial GaN UV-LED chips. Compared with phosphor powders containing Dy³⁺ ion, a single crystal is free of impurity phases, has good uniformity of activated ions, which does not need epoxy resin packaging.

 Dy^{3+} concentration has a great influence on the fluorescence performance of Dy^{3+} doped crystals, and it is meaningful to choose the appropriate doping concentration of Dy: YAG crystal for luminescence and laser operation. The previous researches on Dy^{3+} doped YAG single crystals always concentrated on one crystal with a specific Dy^{3+} doping concentration, and there are only a few studies on the effect of doping concentration on the spectroscopic properties. In this work, the emission spectra of different doping concentration Dy: YAG crystals were contrasted at 447 nm excitation and the spectroscopic parameters were calculated and discussed. The laser properties of Dy: YAG single crystal as laser gain medium was presented.

1 Experimental

 $Y_{3-x}Dy_xAl_5O_{12}$ (x = 0.015, 0.03, 0.06, 0.09, and 0.12) polycrystalline powders were synthesized by hightemperature solid reaction method. Dy_2O_3 , Y_2O_3 , and Al_2O_3 were weighted and mixed according to the stoichiometric ratio and pressed into tablets for sintering. After the sintering process, they were put into an Ir crucible for crystal growth. Yttrium aluminum garnet crystals with different Dy^{3+} concentrations (0.5%, 1.0%, 2.0%, 3.0%, 4.0%, atomic fraction) were grown using the Czochralski method under a high purity N₂ atmosphere. To eliminate the defects in the crystals and release the stress, such crystals were annealed in the air. An image of as-grown crystals is presented in Fig. 1.

The Dy³⁺ concentration of the upper part of the crystals was measured by an inductively coupled plasma atomic emission spectrometer (ICP-AES, Ultima2, Jobin-Yvon). Powder X-ray diffraction (XRD) was characterized on an X-ray diffractometer (Miniflex-600, Rigaku) in step scan mode. The absorption spectra of the samples with a size of 10 mm×10 mm×1 mm were recorded by a UV-VIS-NIR spectrometer (Lamda-980, Perkin-Elmer). The emission spectra and decay curves were measured by a fluorescence spectrometer (FLS980, Edinburgh Instruments). The pump source was a diode laser with the central wavelength of 447 nm and maximum output power of 3.58 W. The laser cavity consisted of two mirror with high transmission (>96%) at 447 nm and high reflection (>99.7%). The 2.0%Dy:YAG crystal with the size of 5 mm×5 mm×10 mm was wrapped with the indium foil and put on the thermos electric cooler.

As seen in Fig. 1, there appeared bubbles and crack in 4.0%Dy: YAG crystal which was grown at first in this work. The appearance of bubbles and crack might stem from the impure polycrystalline powders. To avoid the impaction of impurity phases, we used higher purity of raw materials and extended the grinding time to make the mixture more homogeneous. High quality seed crystal was also used to inhibit the defects of crystals. Furthermore, some measurements were taken such as slowing down the rate of heating and keeping the temperature above the melting point 100 °C for several hours, to make sure the polycrystalline melt completely which eliminated gas bubbles effectively. Changing the temperature field around crucible could improve this situation efficiently too. To strengthen the insulation, the gap between zirconia cylindrical thermal insulation materials and crucible was filled with zirconia powders. At the same time, double-layer zirconia cylinder was used to prevent crystal cracking during annealing.

2 **Results and discussion**

2.1 Absorption spectra and Judd-Ofelt (J-O) analysis

Absorption spectra of Dy: YAG crystals measured at



Fig. 1 Photos of the as-grown single crystals

room temperature are depicted in Fig. 2, which indicates that the absorption coefficient (α) is proportional to the Dy³⁺ concentration. With an increase in doping concentration, the peak wavelengths of absorption bands are almost unchanged. Peaks at 353, 366, 386, 447, 479, 752, 804, 897, 1073, 1291, and 1687 nm correspond to the transitions

4.0%Dy:YAG



Fig. 2 Absorption spectra of Dy: YAG crystals

from the ground state ${}^{6}H_{15/2}$ to ${}^{6}P_{7/2} + {}^{4}I_{11/2}$, ${}^{6}P_{5/2} + {}^{4}M_{19/2}$, ${}^{4}K_{17/2} + {}^{4}M_{21/2} + {}^{4}F_{7/2} + {}^{4}I_{13/2}$, ${}^{4}I_{15/2}$, ${}^{4}F_{9/2}$, ${}^{6}F_{3/2}$, ${}^{6}F_{5/2}$, ${}^{6}F_{7/2}$, ${}^{6}H_{7/2} + {}^{6}F_{9/2}$, ${}^{6}H_{9/2} + {}^{6}F_{11/2}$, and ${}^{6}H_{11/2}$ upper-states, respectively.

Following the ICP-AES results, the segregation coefficient of Dy^{3+} in each crystal is calculated as the ratio of its concentration in the crystal and melt. Obtained results are given in Table 1. The effective segregation coefficient of Dy^{3+} in each crystal is about 0.48, which is in agreement with the previous report^[21].

Next, absorption cross-section (σ_{abs}) at 447 nm is obtained using the formula:

$$\sigma_{\rm abs} = \alpha \,/\, N_{\rm c} \tag{1}$$

Where α is absorption coefficient and N_c is the amount of Dy³⁺ ions per cm³.

The σ_{abs} of Dy³⁺ in YAG is ~1.6×10⁻²¹ cm² as presented in Table 1. The theoretical and experimental line-strengths of Dy: YAG are obtained utilizing J-O theory^[26-27]:

$$S_{\text{cal}}(J \to J') \sum_{t=2,4,6} Q_{t} | S, L, J \| U^{(t)} \| S', L', J' |^{2}$$
(2)

$$S_{\exp}(J \to J') = \frac{27hc(2J+1)n}{8\pi^3 e^2 N_c (n^2+2)^2 \bar{\lambda} L} \int \alpha(\lambda) d\lambda \qquad (3)$$

Where $\Omega_{l(t=2, 4, 6)}$ are the J-O intensity parameters, and Ω_2 is related to the symmetry of Dy³⁺ and the chemical bond between Dy³⁺ and O²⁻. Here, $\| U^{(t)} \|$ is the squared reduced matrix elements of the tensorial operator, which has been calculated by Carnall in Ref. [28]. *n* is refractive indices which are calculated by the Sellmeier equation for YAG crystal^[29]. The values of Planck constant (*h*), speed of light (*c*), and electron charge (*e*) are 6.626×10^{-27} erg·s (1 erg·s = 10^{-7} J·s), 2.998×10¹⁰ cm·s⁻¹, and 4.803×10^{-10} esu (1 A= 3×10^9 esu), respectively. $\overline{\lambda}$ is average wavelength of absorption band for $J \rightarrow J'$ transition, and *L* is length of the crystal in the light pass direction.

The value of root-mean-square (RMS) is evaluated as:

RMS =
$$\sqrt{\sum_{J'} (S_{\exp} - S_{cal})^2 / (N - 3)}$$
 (4)

Where S_{exp} and S_{cal} are the line-strengths of the experimental and theoretical data respectively, and N is the number of absorption bands used for calculation. The three J-O intensity parameters of the studied crystals are fitted and the above parameters are all given in Table 2.

Table 1Concentration, effective segregation coefficientand absorption cross-section of Dy^{3+} in YAG crystal

Crystal	c/% (in atomic)	$k_{\rm eff}$	$N_{\rm c}/{\rm cm}^{-3}$	α/cm^{-1}	$\sigma_{\rm abs}/{ m cm}^2$
0.5% Dy: YAG	0.239	0.478	3.31×10 ¹⁹	0.055	1.66×10 ⁻²¹
1.0% Dy: YAG	0.479	0.479	6.61×10 ¹⁹	0.103	1.56×10 ⁻²¹
2.0% Dy: YAG	0.970	0.485	1.33×10 ²⁰	0.215	1.61×10^{-21}
3.0% Dy: YAG	1.427	0.475	1.95×10 ²⁰	0.338	1.73×10 ⁻²¹
4.0% Dy: YAG	1.975	0.494	2.69×10 ²⁰	0.430	1.60×10^{-21}

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			0.5%Dy: YAG		1.0%Dy: YAG		2.0%Dy: YAG		3.0%Dy: YAG		4.0%Dy: YAG	
${}^{6}\mathrm{H}_{15/2} \rightarrow$	$\overline{\lambda}$ /nm	п	$S_{exp}/(\times 10^{-20}, cm^2)$	$S_{cal}/(\times 10^{-20}, cm^2)$	$S_{exp}/(\times 10^{-20}, cm^2)$	$S_{cal}/(\times 10^{-20}, cm^2)$	$S_{exp}/(\times 10^{-20}, cm^2)$	$S_{cal}/(\times 10^{-20}, cm^2)$	$S_{exp}/(\times 10^{-20}, cm^2)$	$S_{cal}/(\times 10^{-20}, cm^2)$	$S_{exp}/(\times 10^{-20}, cm^2)$	$S_{cal}/(\times 10^{-20}, cm^2)$
${}^{4}M_{17/2} {+}^{6}P_{3/2} {+}$ ${}^{4}G_{9/2} {+}^{4}I_{9/2}$	327	1.89	0.447	0.298	0.460	0.320	0.526	0.323	0.498	0.327	0.467	0.341
$^6P_{7/2} \!\!+^4I_{11/2}$	353	1.88	0.905	0.723	1.128	0.850	1.371	1.182	1.719	1.454	1.648	1.403
${}^{6}P_{5/2} + {}^{4}M_{19/2}$	367	1.87	0.733	0.474	0.806	0.511	0.907	0.525	0.898	0.540	0.918	0.560
${}^4\!K_{17/2} \!\!+ \!\!{}^4\!M_{21/2} \!\!+ \!\!\!{}^4\!I_{13/2} \!\!+ \!\!\!{}^4\!F_{7/2}$	387	1.86	0.651	0.729	0.723	0.793	0.801	0.854	0.851	0.908	0.873	0.925
⁴ I _{15/2}	451	1.85	0.257	0.178	0.253	0.191	0.187	0.191	0.213	0.192	0.227	0.199
${}^{6}F_{3/2}$	755	1.82	0.217	0.158	0.189	0.170	0.202	0.171	0.225	0.173	0.205	0.181
${}^{6}\mathrm{F}_{5/2}$	804	1.82	1.083	0.909	1.140	0.975	1.266	0.983	1.252	0.996	1.228	1.039
⁶ F _{7/2}	908	1.82	2.07	2.057	2.361	2.226	2.243	2.328	2.462	2.426	2.565	2.502
${}^{6}\mathrm{H}_{7/2} + {}^{6}\mathrm{F}_{9/2}$	1088	1.81	2.577	2.741	2.74	3.022	3.249	3.403	3.480	3.731	3.528	3.767
$RMS/(\times 10^{-20}, cm^2)$			0.180		0.230		0.237		0.246		0.226	
$\Omega_{t(t=2, 4, 6)}/(\times 10^{-1})$	²⁰ , cm ²)			$\Omega_2 = 0.793$ $\Omega_4 = 1.284$ $\Omega_6 = 2.634$	2 2 2	$ \Omega_2 = 0.747 $ $ \Omega_4 = 1.520 $ $ \Omega_6 = 2.825 $		$\begin{array}{l} \Omega_2 = 0.498 \\ \Omega_4 = 2.155 \\ \Omega_6 = 2.848 \end{array}$		$\Omega_2=0.312$ $\Omega_4=2.674$ $\Omega_6=2.887$	5. 5. 5.	$P_2=0.142$ $P_4=2.573$ $P_6=3.011$

Table 2 Experimental line strength, calculated line strength, and J-O intensity parameters of Dy: YAG crystals

Later, J-O intensity parameters Ω_t are used to obtain the radiative transition rates from the excited state ${}^4F_{9/2}$ to lower states which comprise both electric dipole (A_{ed}) and magnetic dipole (A_{md}) transitions in some cases. A_{ed} and A_{md} values are computed using formulae:

$$A = A_{\rm ed} + A_{\rm md} \tag{5}$$

$$A_{\rm ed}(J \to J') = \frac{64\pi^4 e^2 n(n^2 + 2)^2}{27h(2J+1)\lambda_{\rm a}^3} \sum_{t=2,4,6} \Omega_t |S, L, J| ||U^{(t)}|| S', L', J'|^2$$
⁽⁶⁾

$$A_{\rm md}(J \to J') = \frac{64\pi^4 e^2 n^3}{3h(2J+1)\lambda_{\rm a}^3} S_{\rm md}$$
(7)

Where A_{ed} and A_{md} are the radiative transition rates contributed by electric-dipole and magnetic-dipole transitions, respectively. Here, the emission transition matrix of Dy³⁺ ion $|| U^{(t)} ||$ and S_{md} values are quoted from Ref. [30].

The fluorescence branching ratios (β) and the radiative lifetime (τ_r) of the ${}^4F_{9/2}$ level in the investigated crystals are obtained using expressions:

$$\beta_{JJ'} = \frac{A(J \to J')}{\sum_{J'} A(J \to J')}$$
(8)

$$\tau_{\rm r} = \frac{1}{\sum_{J'} A(J \to J')} \tag{9}$$

Spontaneous emission transition rate (A) and fluorescence branching ratio (β) of ${}^{4}F_{9/2}$ to lower levels are listed in Table 3. β of yellow emission which corresponds to ${}^{4}F_{9/2} \rightarrow {}^{6}H_{13/2}$ transition is in the range of 46%–50% for Dy: YAG crystals, which indicates that the yellow emission has considerable potential value in them.

Table 3 Spontaneous emission transition rate (A) and fluorescence branching ratio (β) of Dy: YAG crystals

${}^{4}F_{9/2} \rightarrow {}^{2S+1}L_{J}$ -	0.5%Dy: YAG		1.0%Dy: YAG		2.0%Dy: YAG		3.0%Dy: YAG		4.0%Dy: YAG	
	A/s^{-1}	β /%	A/s^{-1}	β/%	A/s^{-1}	β /%	A/s^{-1}	β/%	A/s^{-1}	β/%
${}^{6}F_{1/2}$	0.08	0.01	0.10	0.01	0.14	0.01	0.17	0.02	0.16	0.02
${}^{6}F_{3/2}$	0.15	0.02	0.16	0.02	0.16	0.02	0.16	0.02	0.17	0.02
⁶ F _{5/2}	1.82	0.20	1.81	0.19	1.54	0.15	1.34	0.13	1.05	0.10
⁶ F _{7/2}	13.63	1.49	14.26	1.46	7.29	1.53	8.21	1.58	8.18	1.57
⁶ H _{5/2}	4.02	0.44	4.58	0.47	5.78	0.58	6.78	0.66	6.67	0.65
$^{6}\mathrm{H}_{7/2} + ^{6}\mathrm{F}_{9/2}$	54.73	5.96	59.10	6.04	49.39	6.71	56.13	7.19	55.44	7.10
⁶ H _{9/2}	17.78	1.94	18.84	1.93	15.42	1.98	16.34	2.02	16.13	1.99
⁶ H _{11/2}	43.30	4.72	44.21	4.52	25.93	4.26	24.99	4.06	23.05	3.85
${}^{6}\mathrm{H}_{13/2}$	456.91	49.79	483.85	49.46	482.81	48.22	486.21	47.34	479.08	46.58
${}^{6}\mathrm{H}_{15/2}$	325.23	35.44	351.44	35.92	365.79	36.54	379.80	36.98	392.16	38.13
$\tau_{\rm r}/{ m ms}$	1.090		1.022		0.999		0.974		0.972	

2.2 Fluorescent characteristics

Emission spectra of all samples under 447 nm excitation wavelength are shown in Fig. 3(a). Luminescence band peaks at 484, 582, 676, and 761 nm are corresponding to transitions from ${}^{4}F_{9/2}$ to ${}^{6}H_{15/2}$, ${}^{6}H_{13/2}$, ${}^{6}H_{11/2}$, and ${}^{6}H_{9/2}$ + ${}^{6}F_{11/2}$ lower levels, respectively. The luminescence intensity and the fitted ${}^{4}F_{9/2}$ level lifetime with different doping concentrations at 447 nm were plotted in Fig. 3(b).

Further, quantum efficiency (η) is defined by $\eta = \tau/\tau_r$, where, τ_r is ${}^{4}F_{9/2}$ level radiative lifetime, τ is ${}^{4}F_{9/2}$ level lifetime. The stimulated emission cross-section (σ_{em}) can be obtained by Füchtbauere Ladenburg formula^[31-32] as:

$$\sigma_{\rm em}(\lambda) = \frac{\lambda^5 I(\lambda) A(J \to J')}{8\pi n^2 c \int \lambda I(\lambda) d\lambda}$$
(10)

Where λ , *n*, *c*, and $A(J \rightarrow J')$ have their usual meanings. The value of $\sigma_{\rm em}\tau$ is calculated and it is inversely proportional to the laser threshold^[33]. All related spectroscopic parameters are contrasted with other Dy³⁺ doped crystals. $\sigma_{\rm em}$ at 582 nm of as-grown Dy: YAG crystals are found to be in the range from 2.36×10^{-21} to 2.71×10^{-21} cm². The obtained results in all samples are listed in Table 4.

 $\sigma_{\rm em}$ of 1.0%Dy: YAG is close to those of gallate and

scandate crystals related value at a similar concentration while it is much smaller than those of sesquioxide and fluoride crystals^[13-14, 34-36] respective values. From the $\sigma_{\rm em}\tau$ product, it is noticed that Dy: YAG possesses a smaller threshold than Dy³⁺: Gd₃Ga₃Al₂O₁₂, Dy³⁺: Gd₃Ga₅O₁₂, and Dy³⁺: GdScO₃ crystals. Fluoride crystals might have a lower threshold for yellow laser operation but due to adverse factors such as thermal effect and lower mechanical strength, it is difficult to develop high-power lasers^[18-19]. η of Dy: YAG declines rapidly with Dy³⁺ doping content increment, which is a sign of non-radiative transitions existence between Dy³⁺ ions, so it is not suitable to generate lasing action with Dy: YAG crystal at higher doping levels. Among all crystals, 1.0% Dy: YAG has the largest $\sigma_{\rm em}$ as well as $\sigma_{\rm em}\tau$ for 582 nm yellow emission. Among five samples, the 1.0% Dy: YAG crystal also shows an intense fluorescence intensity with the 447 nm excitation with a longer decay time of 0.823 ms, which is close to the previous work^[37]. The fluorescence intensity and σ_{em} of 2.0% Dy: YAG are slightly less than that of 1.0% Dy: YAG, but 2.0%Dy: YAG has a higher α . Hence, 1.0%Dy: YAG and 2.0%Dy: YAG crystals are the suitable candidates for the generation of yellow laser output.



Fig. 3 Emission spectra of YAG crystals with different Dy^{3+} concentrations excited by 447 nm (a) and variation of intensity of 582 nm and ${}^{4}F_{9/2}$ level lifetime with Dy^{3+} concentrations in Dy: YAG crystals (b)

Crystal	$\tau_{\rm r}({}^4\mathrm{F}_{9/2}\mathrm{level})/\mathrm{ms}$	$\tau({}^{4}\mathrm{F}_{9/2}\mathrm{level})/\mathrm{ms}$	$\sigma_{\rm em}$ for yellow emission/(×10 ⁻²¹ , cm ²)	$\sigma_{\rm em} \tau / (\times 10^{-21}, {\rm cm}^2 \cdot {\rm ms})$	η/%	Ref.
0.5%Dy: YAG	1.090	0.894	2.36	2.110	82.02	
1.0%Dy: YAG	1.022	0.823	2.71	2.230	80.53	
2.0%Dy: YAG	0.999	0.688	2.66	1.830	68.87	This
3.0%Dy: YAG	0.974	0.571	2.54	1.450	58.62	work
4.0%Dy: YAG	0.972	0.471	2.49	1.170	48.46	
1.0%Dy ³⁺ : Gd ₃ Ga ₃ Al ₂ O ₁₂	0.596	0.573	3.20	1.834	96.14	[13]
3.0%Dy ³⁺ : Lu ₂ O ₃	0.756	0.112	7.10	7.952	14.80	[14]
2.0%Dy ³⁺ : CeF ₃	3.747	1.530	9.259	0.1417	40.83	[17]
1.0%Dy ³⁺ : GdScO ₃	0.650	0.459	4.10	1.882	70.60	[34]
2.0%Dy ³⁺ : Gd ₃ Ga ₅ O ₁₂	1.107	0.790	2.62	2.070	71.40	[35]
2.0%Dy ³⁺ : LaF ₃	1.700	1.370	7.00	9.590	80.59	[36]



Fig. 4 Laser spectra (a) and variation of output power with absorbed pump power of 2.0%Dy: YAG crystal (b)

2.3 Laser performance

The continuous wave (CW) yellow laser output was obtained in 2.0%Dy: YAG crystal. Fig. 4(a) shows the laser spectra of Dy: YAG crystal. The central wavelength of the laser is 582.5 nm, and the full width at half maximum is 1.2 nm. Fig. 4(b) shows the relation curve between absorption pump power and output power, and the maximum output power is 166.8 µW. The slope efficiency is calculated to be 0.32%, with the maximum output power of 0.029%. The laser operation threshold is around 535 mW, which is higher than that of Dy^{3+} , Tb^{3+} : LiLuF₄ (320 mW)^[38]. When the input power exceeds 3382 mW, the output power decreases significantly which is mainly due to the degradation of rare earth ions caused by laser thermal effect. Owing to the laser experiment parameters haven't been optimized, the laser output power and efficiency are still very low, we are trying to improve the laser output power now.

3 Conclusions

The visible fluorescence features of Dy: YAG crystals with different Dy³⁺ doping concentrations were analyzed thoroughly. The optimum concentration for yellow emission under 447 nm is 1.0%, and the ${}^{4}F_{9/2}$ level lifetime is calculated to be 0.823 ms. 1.0%Dy: YAG has the largest σ_{em} at 582 nm in present work. 2.0%Dy: YAG could also be an alternative crystal for yellow lasing action on consideration of its greater yellow emission intensity and σ_{em} , the acceptable lifetime of ${}^{4}F_{9/2}$ level, and higher absorption intensity. Hence, 1.0%Dy: YAG and 2.0%Dy: YAG crystals are potential candidates for yellow laser output. Finally, the CW laser with the maximum output power of 166.8 μ W yellow laser operation is obtained based on 2.0%Dy: YAG crystal.

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Dy³⁺: Y₃Al₅O₁₂ 晶体的光谱与黄色激光性能

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摘要:近年来,黄色激光晶体在激光显示、激光医疗、激光雷达(光探测和测距)、玻色-爱因斯坦凝聚、原子冷却和俘获等领域具有广泛的应用,吸引了研究人员极大的兴趣。随着蓝光 LD 泵浦源的商用化,直接泵浦 Dy³⁺掺杂激 光晶体可输出黄色激光,对应⁴F_{9/2}→⁶H_{13/2}跃迁。本工作采用提拉法生长了 Dy³⁺掺杂浓度分别为 0.5%、1.0%、2.0%、 3.0%和 4.0%(原子分数)的 Dy³⁺: Y₃Al₅O₁₂(Dy:YAG)晶体,并分析了晶体开裂的原因。基于 Judd-Ofelt 理论计算了 J-O 强度参数,并利用其评估了不同掺杂浓度的 Dy:YAG 晶体的其它激光参数。综合讨论了 Dy³⁺掺杂浓度对荧光分支比、 受激发射截面、量子效率等光谱性能的影响。在五个晶体样品中,1.0%Dy: YAG 晶体在 447 nm 激发下实现了 582 nm 最大的受激发射截面值和最强的荧光强度值,荧光寿命较长,达到 0.823 ms。与之相比,2.0%Dy: YAG 晶体发射参 数值略低,但是其吸收系数更大。研究结果表明,激光二极管泵浦的 Dy:YAG 黄色激光晶体中 Dy³⁺离子的浓度为1.0%

关 键 词: YAG 晶体; Dy³⁺; 晶体生长; 荧光特性; 黄色激光

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