

Ti₃C₂T_x/Wood Carbon as High-areal-capacity Electrodes for Supercapacitors

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Abstract: MXenes—two-dimensional (2D) compounds generated from layered bulk materials, have attracted significant attention in energy storage fields. However, low mass loading of MXenes results in low areal capacity and impedes the practical use of MXenes electrodes. Inspired by natural basswood, an ideal architecture with natural, three-dimensionally (3D) aligned open microchannels was developed for high Ti₃C₂T_x mass loading. Compared with reported Ti₃C₂T_x electrode structure, the 3D porous carbon matrix has several advantages including low tortuosity, high conductivity and good structure stability. The Ti₃C₂T_x assembled with the wood carbon can deliver a high areal capacity of 1983 mF/cm² at 2 mV/s with a high Ti₃C₂T_x mass loading of 17.9 mg/cm² when used as electrode for supercapacitors. This work provides a new strategy to develop 3D porous electrodes for MXenes, which can achieve high areal capacity.

Key words: Ti₃C₂T_x; wood carbon; supercapacitor; high areal capacity

MXenes have gained tremendous attention in supercapacitor^[1], lithium battery^[2], electromagnetic interference shields^[3] and other energy-related fields. They were firstly reported by Michael Naguib *via* selectively etching from MAX phases^[4] (where M is a transitional metal, A is a III or IV A-group element and X is carbon or nitrogen). The resulting 2D materials, *i.e.*, MXenes, have many chemical and structural forms with high intrinsic electronic/ionic conductivities. When applied to supercapacitors, the fast surface redox makes them possess large energy. However, although much effort have been made to obtain superior gravimetric and volumetric capacity of MXenes^[1, 5], the mass loading of MXenes still needs to be further improved in order to get a high areal capacity and meet practical applications in large scale.

In 2017, Chen, *et al*^[6] reported a fascinating work to utilize natural basswood structure for an all-wood-structured asymmetric supercapacitors design. The electrodes were derived from natural wood and inherited its unique anisotropic structure which has numerous open channels along the growth direction. The wood based devices displayed high desirable thickness (up to ~1 mm) with high mass loading of 30 mg/cm². With these unique features, such as low tortuosity, high electronic conductivity, enabling high mass loading, low cost and environmentally friendly, the wood-structured electrode represents a

promising way to realize high power density and high energy density for green energy storage device^[7-9].

Herein, we reported a facile way to make a high mass loading NaOH-Ti₃C₂T_x/wood carbon electrode for supercapacitors. The resulted electrode exhibited a high areal capacity of 1983 mF/cm² at 2 mV/s with a high Ti₃C₂T_x mass loading of 17.9 mg/cm², which surpass three times than that of NaOH-Ti₃C₂T_x reported before^[10].

1 Experimental

1.1 Synthesis of Ti₃C₂T_x

Ti₃C₂T_x (T=O, OH) was synthesized according to the paper reported before^[10]. About 80 mg Ti₃AlC₂ powder (400 mesh (38 μm), purchased from 11 Technology Co. Ltd.) was added into the 35 mL NaOH solution (27.5 mol/L, the water was heated to boiling to reduce the concentration of oxygen) contained in a 50 mL autoclave in argon (Ar) atmosphere. Then the autoclave was heated at 270 °C in an electric thermostatic drying oven for 12 h. Finally, the resultant suspension was filtered with deionized water washing for several times. The obtained powder was dried at 60 °C for 12 h.

1.2 Synthesis of wood carbon

Natural basswood blocks were cut perpendicularly in

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the growth direction to afford wood slices. Then the slices were pre-carbonized in air at 250 °C for 6 h, followed by carbonization in an argon flow at 1000 °C for 6 h. The carbonized slices were carefully polished with 2000 grit sandpaper to obtain a thickness of 700 μm and residual carbon was removed with ultrasonic washing by deionized water and ethanol for three times. To further reduce the weight and activate the wood carbon, the resultant wood carbon slices were activated in a carbon dioxide (CO_2) flow at 750 °C for 1 h in a gas flow of 0.2 L/min.

1.3 Synthesis of $\text{Ti}_3\text{C}_2\text{T}_x$ @WC

$\text{Ti}_3\text{C}_2\text{T}_x$ was loaded into the channels *via* vacuum-assisted infiltration. Briefly, 160 mg amount of prepared $\text{Ti}_3\text{C}_2\text{T}_x$ powder was dispersed in 10 mL ethanol *via* sonication for 30 min. The as-prepared wood carbon was placed on a filter funnel and then the obtained homogeneous ink was dripped into the wood carbon matrix along the microchannels after vacuum filtration using a turbo pump. The wood carbon filled with $\text{Ti}_3\text{C}_2\text{T}_x$ was treated by freeze-drying for 48 h. The mass ratio of $\text{Ti}_3\text{C}_2\text{T}_x$ with wood carbon was $\sim 1 : 2.4$.

1.4 Characterization

XRD test was conducted using a Rigaku Ultima IV Powder Diffractometer (Cu $K\alpha$ radiation, sweeping speed: 5(°)/min). TEM analysis was performed on a JEM-2100F transmission electron microscope (200 kV, JEOL, Japan). SEM studies were carried out on a Mira3 scanning electron microscope (5 kV for morphology observation, Tescan, Czech).

1.5 Electrochemical measurements

All electrochemical measurements were performed using a traditional three electrode system with 1 mol/L H_2SO_4 solution as electrolyte. $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon electrode, carbon rod, and $\text{Hg}/\text{Hg}_2\text{SO}_4$ in saturated K_2SO_4 aqueous solution were used as working electrode, counter electrode, and reference electrode, respectively. Cyclic voltammograms (CVs) and galvanostatic charge-discharge (GCD) analysis were performed on an electrochemical workstation (Biologic VMP3). The scan rates for CV analysis were 1, 2, 5, 10 and 20 mV/s. The scan range ($-0.5 \sim 0$ V) was determined by series of CV scans at 10 mV/s.

2 Results and discussion

2.1 Characterizations of $\text{Ti}_3\text{C}_2\text{T}_x$

$\text{Ti}_3\text{C}_2\text{T}_x$ was synthesized from bulk material Ti_3AlC_2 *via* hydrothermal method (27.5 mol NaOH, Ar atmosphere, 270 °C). The raw material Ti_3AlC_2 was shown in Fig. 1(a). The bulk Ti_3AlC_2 material was made from Ti_2AlC and TiC by a hot pressing sintering method^[11].

There were some impurities such as Al_2O_3 (See XRD patterns of Ti_3AlC_2 , Fig. 1(c)) during the synthesis process. After alkali treatment, sheet structure emerged in obtained $\text{Ti}_3\text{C}_2\text{T}_x$ (Fig. 1(b)), which indicated the successful etching of Al atoms. It was also supported by X-ray diffraction (XRD) that the rising peak at $2\theta \approx 7.5^\circ$ corresponding to (002) plane of $\text{Ti}_3\text{C}_2\text{T}_x$. The morphology of NaOH- $\text{Ti}_3\text{C}_2\text{T}_x$ was different from accordion-like HF- $\text{Ti}_3\text{C}_2\text{T}_x$ *via* high concentration (50wt%) HF treatment but similar to low concentration (5wt%) HF treatment^[11]. It is noted that the position of (002) peak in NaOH route is similar to HF etching route followed by sodium intercalation^[12], which indicates that NaOH treatment has the same effect. Furthermore, we also performed Transmission electron microscopy (TEM) analysis to observe the morphology of $\text{Ti}_3\text{C}_2\text{T}_x$. Lamellar stripes can be observed in the edge of $\text{Ti}_3\text{C}_2\text{T}_x$ sheets and the interlayer space was ~ 1.2 nm. The space was larger than that of hydrofluoric acid (HF) method, which can be explained by sodium intercalation during the hydrothermal process.

2.2 Morphology of Wood carbon and $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon

SEM was used to observe the structure of wood carbon. Fig. 2(a) showed the top-view of wood carbon, where multi big channels (30–60 μm) were surrounded by numerous small channels (5–10 μm). From the cross-sectional view (Fig. 2(b)), we can clearly observe that almost all channels were aligned straight from top to bottom. Furthermore, there were some mesopores (high magnification image in Fig. 2(b)) existed in channels, which were beneficial to diffusion of electrolyte. Next, $\text{Ti}_3\text{C}_2\text{T}_x$ ink was made *via* sonication in ethanol for 30 min and then dripped into the wood carbon with a vacuum filtration method. SEM images showed that multilayer $\text{Ti}_3\text{C}_2\text{T}_x$ was successfully injected (Fig. 2(c, d)). However, $\text{Ti}_3\text{C}_2\text{T}_x$ was not uniformly distributed in the channels of wood carbon. Some channels were filled with $\text{Ti}_3\text{C}_2\text{T}_x$, while some were empty, which may not take full use of wood structure and every single $\text{Ti}_3\text{C}_2\text{T}_x$.

2.3 Supercapacitor performance of $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon

Since wood displayed anisotropic structure with vertical micro-channels serving as the high transport pathway of ion transport in the electrode, we thus tested the supercapacitor performance of $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon. The supercapacitor performance was evaluated using a three-electrode cell with 1 mol/L H_2SO_4 aqueous solution serving as the electrolyte. To the best of our knowledge, $\text{Ti}_3\text{C}_2\text{T}_x$ is stable from $-1.0 \sim 0$ V *vs.* Hg/HgSO_4

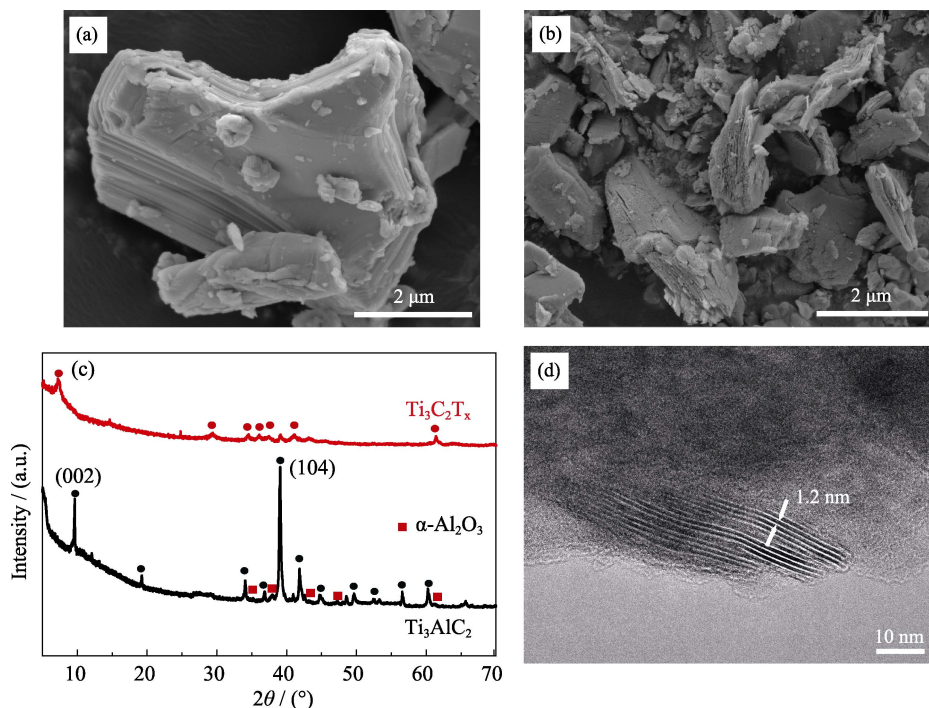


Fig. 1 Characterizations of $\text{Ti}_3\text{C}_2\text{T}_x$
 (a) SEM image of raw material Ti_3AlC_2 , (b) SEM image of Ti_3AlC_2 via NaOH etching,
 (c) XRD patterns of Ti_3AlC_2 and $\text{Ti}_3\text{C}_2\text{T}_x$, (d) TEM image of NaOH- $\text{Ti}_3\text{C}_2\text{T}_x$

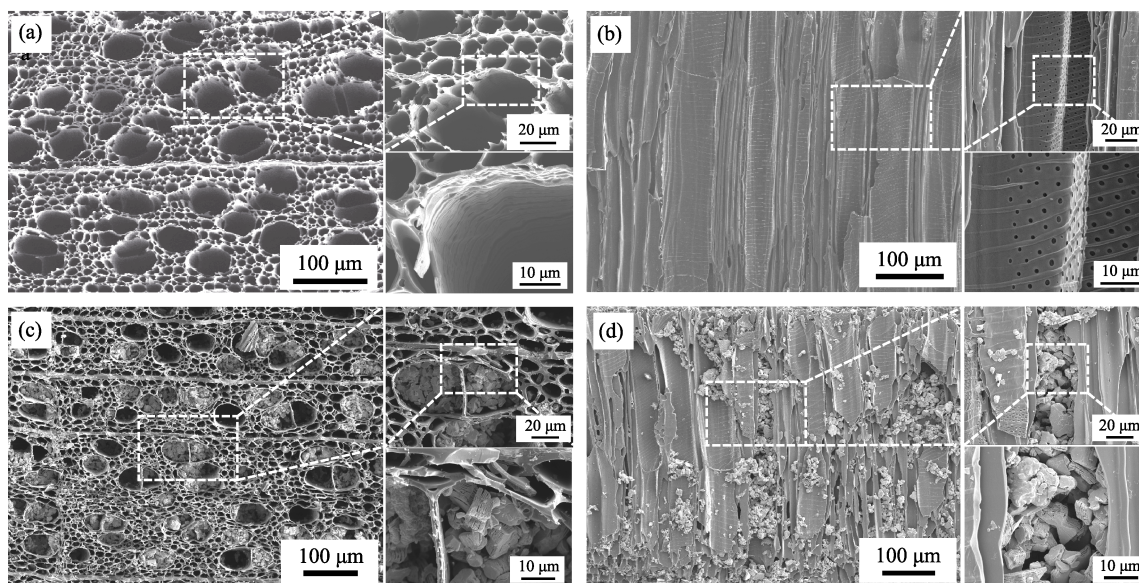


Fig. 2 SEM images of bare wood carbon ((a, b)) and $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon ((c, d))
 (a, c) Top-view; (b, d) Cross-sectional view

while the wood carbon is stable from -0.5 – 0.5 V vs. Hg/HgSO₄ according to our experiments (Fig. 3(a)). Therefore, -0.5 – 0 V was chosen as this potential range which was suitable for both of them. Fig. 3(b) showed the CV curves of $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon. Different scanning rates ranging from 1 to 20 mV/s were applied. For wood carbon, the CV curves were nearly like rectangle shape which indicated fast charge transfer in the electrode and the wood carbon exhibited high areal capacity of 1030 mF/cm² at 2 mV/s. For $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon (mass loading: 17.9 mg/cm²), the area of CV curves were

relatively larger than that of bare wood at the same potential range and the performance was 1983 mF/cm² at 2 mV/s, representing good values among $\text{Ti}_3\text{C}_2\text{T}_x$ electrodes reported before^[10]. Even at 20 mV/s, the capacity of $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon electrode still maintained 1539 mF/cm². GCD curves of $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon were showed in Fig. 3(c). The linear GCD curves with triangular shape suggested ideal capacitive behavior, while the repeated GCD curves at 10 mA/cm² for 10⁴ cycles indicated good stability (maintained ~92%) for $\text{Ti}_3\text{C}_2\text{T}_x$ /wood carbon electrode (Fig. 3(d)).

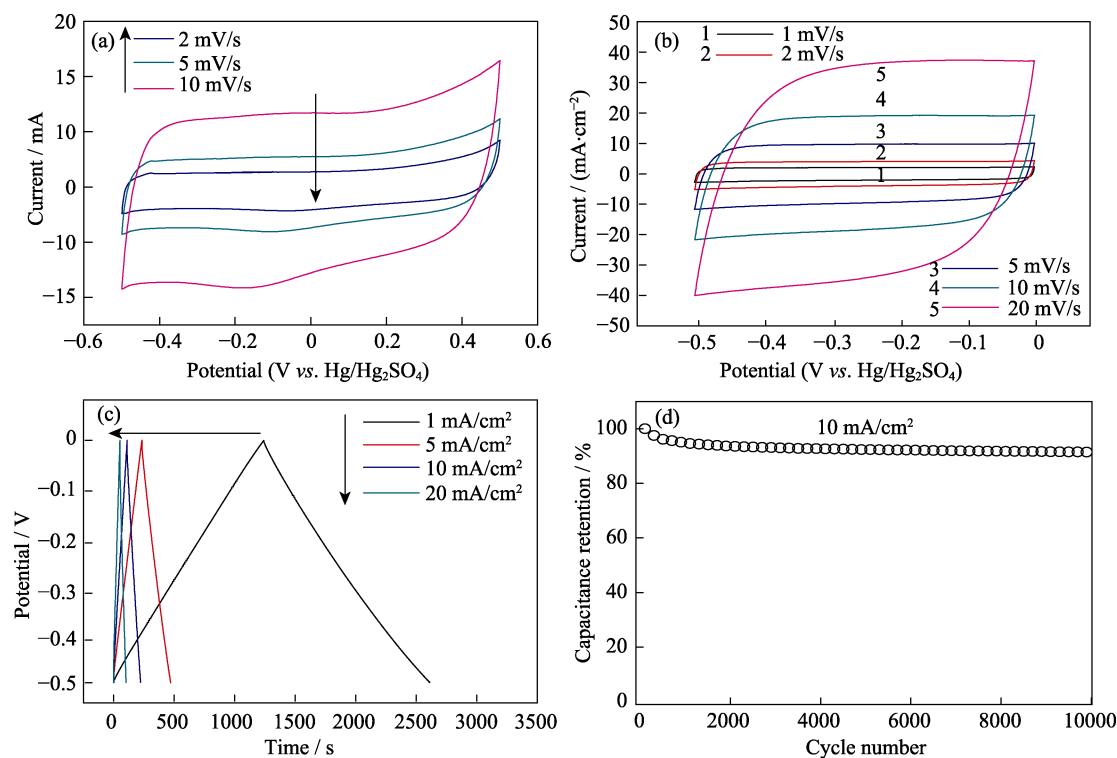


Fig. 3 CV curves of bare wood carbon (a) and $\text{Ti}_3\text{C}_2\text{T}_x/\text{wood carbon}$ (b), GCD cycling profiles (c) of the $\text{Ti}_3\text{C}_2\text{T}_x/\text{wood carbon}$ electrode collected at 1, 5, 10, and 20 mA/cm^2 , and capacitance retention test at 10 mA/cm^2 (d)

3 Conclusions

In conclusion, we have demonstrated wood-based $\text{Ti}_3\text{C}_2\text{T}_x$ electrode a unique structure for supercapacitor to achieve high areal capacity. Open channels of wood benefit fast and efficient transportation of ions. When $\text{Ti}_3\text{C}_2\text{T}_x$ was combined with wood carbon, a high areal capacity (1983 mF/cm^2 at 2 mV/s) of with high mass loading (17.9 mg/cm^2) was achieved. This work also inspires us to explore other natural structure, which may possess fascinating properties in combination with other materials.

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碳化钛/椴木多孔碳复合材料用于超级电容器性能的研究

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摘要: 碳化钛作为一种新兴的层状二维材料具有一些独特的物理化学性质, 近年来引起了科研工作者广泛的注意。它是由化学选择性刻蚀的方法获得, 在电化学如锂电池, 超级电容器等领域展现出极好的应用前景。目前研究中碳化钛的电极往往活性物质负载量较低, 导致面容量不佳, 从而限制了其在大规模生产中的应用。本工作受自然界中椴木结构的启发, 利用其多孔道、孔道弯曲度低、导电性好、低价环保等特点, 将碳化钛与椴木活性炭复合, 获得了一种具有高面电容且稳定的超级电容器, 该电容器在 2 mV/s 的扫速下具有 1983 mF/cm² 的面容量, 同时活性材料负载量可以达到 17.9 mg/cm²。本研究为后续利用自然界构型材料与功能材料的复合提供了一定的借鉴。

关键词: 碳化钛; 椴木活性炭; 超级电容器; 面电容

中图分类号: TQ174 **文献标识码:** A