

Effects of pulse width and repetition rate of pulsed laser on kinetics and production of singlet oxygen luminescence

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Pulsed and continuous-wave (CW) lasers have been widely used as the light sources for photodynamic therapy (PDT) treatment. Singlet oxygen (${}^{1}O_{2}$) is known to be a major cytotoxic agent in type-II PDT and can be directly detected by its near-infrared luminescence at 1270 nm. As compared to CW laser excitation, the effects of pulse width and repetition rate of pulsed laser on the kinetics and production of ${}^{1}O_{2}$ luminescence were quantitatively studied during photosensitization of Rose Bengal. Significant difference in kinetics of ${}^{1}O_{2}$ luminescence was found under the excitation with various pulse widths of nanosecond, microsecond and CW irradiation with power of 20 mW. The peak intensity and duration of ${}^{1}O_{2}$ production varied with the pulse widths for pulsed laser excitation, while the ${}^{1}O_{2}$ was generated continuously and its production reached a steady state with CW excitation. However, no significant difference (P > 0.05) in integral

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 $^{1}O_{2}$ production was observed. The results suggest that the PDT efficacy using pulsed laser may be identical to the CW laser with the same wavelength and the same average fluence rate below a threshold in solution.

Keywords: Photodynamic therapy; pulsed laser; continuous-wave laser; singlet oxygen luminescence.

1. Introduction

Photodynamic therapy (PDT) is an effective therapeutic modality that utilizes a light source to excite photosensitizer (PS) for the treatment of malignant and nonmalignant conditions, and singlet oxygen ($^{1}O_{2}$) is known to be the primary cytotoxic agent generated photochemically in type-II PDT.^{1,2} $^{1}O_{2}$ can be detected directly by its nearinfrared (NIR) luminescence emission at around 1270 nm,³ and $^{1}O_{2}$ NIR luminescence was investigated as a dosimetric predictor of PDT responses both *in vitro* and *in vivo*.⁴⁻⁶

PDT involves three major components: light, PS and molecular oxygen.⁷ Continuous-wave (CW) and pulsed lasers were widely used as excitation light sources for activating PS in PDT, and the effects of PS types,^{8,9} oxygen concentration^{10,11} on ¹O₂ production have been intensively evaluated, respectively. However, as one of the three essential components of PDT, the effects of laser parameters (i.e., pulse width, repetition rate) for pulsed lasers on the kinetics and production of ¹O₂ luminescence have not been fully explored.

The aim of the study was to examine the effects of pulse width and repetition rate of pulsed laser on the kinetics and production of ${}^{1}O_{2}$ luminescence during photosensitization by directly detecting its NIR luminescence. Water-soluble Rose Bengal (RB) was used as model PS, and RB mediated ${}^{1}O_{2}$ luminescence under the excitation with various pulse widths and repetition rates were quantitatively studied by using a time-resolved ${}^{1}O_{2}$ luminescence detection system.

2. Materials and Methods

2.1. Chemicals

For generation of ${}^{1}O_{2}$, RB (Sigma-Aldrich, St. Louis, MO, USA) was used as a model PS in the solutionbased measurements. A 100 μ M RB was prepared as stock solution in initially air-statured phosphate buffered saline (PBS) at pH 7.5, and stored at 4°C in the dark. $1.00 \,\mu\text{M}$ RB was freshly prepared from the stock solution in each measurement. The absorption spectra of RB in PBS were measured with a UV/Vis/NIR spectrophotometer (Lambda950, PerkinElmer, Waltham MA, USA).

2.2. Excitation light sources

A Q-switched 523 nm Nd:YLF laser (QG-523-500, Crystalaser Inc., Reno, USA) and a custom developed modulated 532 nm diode-pumped all-solidstate laser (DPSSL) (Institute of Semiconductors, Chinese Academy of Sciences, Beijing, China) were used as light sources to excite the PS, respectively. The Q-switched 523 nm laser provides high energy pulses with adjustable pulse repetition rates in range of 1–100 kHz. Typically, the pulse width of 523 nm laser was 20.7 ns at 10 kHz. The modulated $532 \,\mathrm{nm}$ DPSSL can be operating in both CW and pulse modes. For pulse mode, the DPSSL was modulated between "on" and "off" state by a Transistor–Transistor-Logic (TTL) signal generated from a two-channel function/arbitrary waveform generator (DG1022U, Beijing RIGOL Technology Co., Ltd., Beijing, China). The minimum pulse width was determined to be about $10\,\mu s$, and the pulse repetition rates can range from 1 Hz to 15 kHz. The pulse width and repetition rate were measured using a digital storage oscilloscope (TDS3054C, Tektronix Inc., Beaverton, OR, USA) and a photodiode (PDM-400, Becker & Hickl GmbH, Berlin, Germany). Figure 1 shows the normalized laser pulse of 523 nm and 532 nm (20% duty cycle) pulsed laser at the repetition rate of 10 kHz. The output laser pulse shape of 532 nm DPSSL was not squarewave pulse due to the long rise and fall time.

2.3. $^{1}O_{2}$ luminescence detection system

The ${}^{1}O_{2}$ luminescence was measured by using a custom developed time-resolved NIR luminescence detection system, which was described in detail previously.¹² Briefly, the excitation laser light was



Fig. 1. The normalized laser pulse shape of 523 nm and 532 nm pulsed laser at the repetition rate of 10 kHz.

directed into the measured 2.5 ml solution placed in a standard 10 mm pathlength quartz cuvette, and the beam diameter at the sample is about 1 mm. The cuvette was open so that the solutions were exposed to room air at the top. The excited luminescence was collected at right angle to the excitation beam through a 1000 nm long-pass filter, the collection optics and the series of three NIR narrow-band filters centered at 1230, 1270 and 1310 nm (OD3 blocking, 20 nm full width at halfmaximum (FWHM); Omega Optical, Brattleboro, USA) that were placed in front of the NIR-photomultiplier tube (NIR-PMT, H10330-45, Hamamatsu Corp., Hamamatsu, Japan) with a constant operating voltage of -900 V. The PMT output pulses were pre-amplified and finally recorded in the fast photon counter (MSA-300, Becker & Hickl GmbH, Berlin, Germany), which was triggered by synchronizing trigger signal of excitation laser. In order to achieve a sufficient signal-to-noise ratio (SNR), the laser power and irradiation time for each measurement were set to $20 \,\mathrm{mW}$ and $30 \,\mathrm{s}$, respectively. The three narrow-band filters allow sampling of the NIR spectrum across the ${}^{1}O_{2}$ luminescence peak at 1270 nm and background. The $^{1}O_{2}$ luminescence intensity at 1270 nm for RB in PBS was corrected for background by subtracting the average luminescence intensity of the 1230 nm and 1310 nm filters.

2.4. Experimental protocols

A series of pulse widths for modulated 532 nm DPSSL was obtained by adjusting the duty cycle for a square waveform from 10% to 50% in

Table 1. Pulse widths of excitation lasers.

Wavelengths (nm)	Mode	Pulse width
523 532 532	Pulse Pulse CW	$\begin{array}{c} 20.7\mathrm{ns}\\ 10.7,14.0,19.9,28.0,36.0\mu\mathrm{s} \end{array}$

increments of 10% using the waveform generator. The pulse widths of excitation lasers were shown in Table 1. For evaluating the effect of repetition rates on the kinetics and production of ${}^{1}O_{2}$ luminescence, the time-resolved ${}^{1}O_{2}$ luminescence excited by modulated 532 nm DPSSL with repetition rates of 1, 5, 10 and 15 kHz was also investigated as compared to CW excitation. In addition, the obtained time-resolved ${}^{1}O_{2}$ luminescence curve and the corresponding time-integrated ${}^{1}O_{2}$ luminescence intensity after background subtraction were comparatively studied.

2.5. Kinetic models for ${}^{1}O_{2}$ production

The kinetics of the PDT photochemical reactions can be described by the following three coupled differential equations¹³:

$$\frac{d[S_1]}{dt} = I\sigma[S_0] - \frac{[S_1]}{\tau_s},$$
(1)

$$\frac{d[T_1]}{dt} = \frac{\Phi_T}{\tau_s} [S_1] - \frac{[T_1]}{\tau_T},$$
(2)

$$\frac{d[{}^{1}O_{2}]}{dt} = \frac{\Phi_{D}}{\Phi_{T}\tau_{T}}[T_{1}] - \frac{[{}^{1}O_{2}]}{\tau_{D}}, \qquad (3)$$

where I is the excitation laser intensity in photons $(\text{cm}^{-2} \text{ s}^{-1})$ and is a function of time, σ is the molecular absorption cross-section of PS (cm^{-2}) . $[S_0], [S_1]$ and $[T_1]$ are the concentrations of PS ground-state, singlet-state and triplet-state, respectively. Φ_D is the ¹O₂ quantum yield and Φ_T is the PS triplet-state quantum yield. τ_s is the PS singlet-state lifetime, τ_D and τ_T are the lifetimes of ¹O₂ and PS triplet-state, respectively. We have assumed that the $[S_0]$ and $[{}^{3}O_2]$ are independent of time.

For a sufficiently short ($\langle \sim 20 \text{ ns} \rangle$) laser pulse excitation, the excitation laser pulse can be treated as an "instantaneous" source and the triplet-state PS molecules are created instantaneously after excitation (valid for $\tau_s \ll \tau_T$), the ¹O₂ concentration as a function of time generated after laser pulse can be described as 13 :

$$[{}^{1}O_{2}]_{t} = N\sigma[S_{0}]\Phi_{D}\frac{\tau_{D}}{\tau_{T} - \tau_{D}}\left[\exp\left(\frac{-t}{\tau_{T}}\right) - \exp\left(\frac{-t}{\tau_{D}}\right)\right]$$

$$\tag{4}$$

where N is the total number of incident photons per unit area (photons cm⁻²) at time t = 0.

For a relatively long (i.e., $10.7-36 \,\mu$ s used in the study) laser pulse excitation, Eq. (4) is no longer valid while laser is on. To gain a better understanding of ${}^{1}O_{2}$ production, we solved Eqs. (1)–(3) numerically using Matlab software (Matlab2010, MathWorks Inc., USA).

For CW excitation, the concentration of ${}^{1}O_{2}$ reaches steady-state when PS is continuously irradiated. In this case, the rate of ${}^{1}O_{2}$ production $I_{a}\Phi_{D}$ equals the rate of ${}^{1}O_{2}$ decay $k_{d}[{}^{1}O_{2}]_{ss}$, 14 and thus

$$[{}^1O_2]_{ss} = \frac{I_a \Phi_D}{k_d},\tag{5}$$

where I_a is the photon absorption rate, and k_d is the decay rate of ${}^{1}O_2$.

2.6. Data analysis

The temporal profiles of ${}^{1}O_{2}$ luminescence were simulated by Matlab software, combined with Eqs. (1)–(3) for the 10.7 μ s pulse excitation, Eq. (4) for the 20.7 ns pulse excitation and Eq. (5) for CW excitation, respectively. All the measured data were processed and analyzed using OriginPro 9.1 software (OriginLab Corp.). Data are presented as means \pm the standard deviation (SD) of at least three independent measurements. Statistical analysis was performed with two-tailed unpaired Student's *t*-test. Differences were considered statistically significant at P < 0.05.

3. Results and Discussion

3.1. Kinetics of ${}^{1}O_{2}$ production

Figures 2(a)-2(c) show the representative time-resolved NIR luminescence curves for three individual wavelengths and the corresponding ${}^{1}O_{2}$ luminescence curve after background subtraction for $1.00 \,\mu\text{M}$ RB in PBS with pulse width of 20.7 nm at 523 nm, pulse width of $10.7 \,\mu\text{s}$ at 532 nm, CW at 532 nm, respectively. Significant difference in kinetics of ${}^{1}O_{2}$ luminescence was found under the excitation with various pulse widths of nanosecond, microsecond and CW irradiation with power of 20 mW. The 1230 nm and 1310 nm filters provide the reference of background luminescence, while the 1270 nm filter contains underlying background luminescence and ${}^{1}O_{2}$ emission. Figure 2(d) shows the time-integrated luminescence spectra by summing over all the luminescence counts for the three individual wavelengths, respectively. In each case, a significant spectra peak was observed at 1270 nm, which is consistent with ${}^{1}O_{2}$ luminescence.

The kinetics of ${}^{1}O_{2}$ luminescence under 20.7 ns and 10.7 μ s pulse laser excitation were simulated using Eq. (4) and Eqs. (1)–(3), respectively. As shown in Figs. 3(a) and 3(b), the simulated data are in good agreement with the measured data, which implies that the equation models allow the prediction for the kinetics of ${}^{1}O_{2}$ luminescence with various pulse widths over a large range.

In the study, the 523 nm Q-switched pulsed laser produces short pulses (20.7 ns), which meets the requirement of sufficient short pulse laser excitation. It can be assumed that the triplet-state PS is created instantaneously after excitation,¹³ and the tripletstate PS then can react with the molecular oxygen to generate ${}^{1}O_{2}$. After the termination of the laser pulse, the kinetic of ${}^{1}O_{2}$ luminescence can be described by the difference of two exponentials, as shown in Eq. (4). The 532 nm modulated DPSSL provides a relatively longer laser pulses $(10.7 \,\mu s)$. The tripletstate PS continuously generates and populates ¹O₂ when the laser is on. The kinetics of ${}^{1}O_{2}$ luminescence changed continuously with the excitation laser pulse shape. The temporal profiles of ${}^{1}O_{2}$ luminescence and PS fluorescence were also simulated by MathCAD software using a pulsed diode laser with standard square-wave pulses (pulse widths: $1-10 \,\mu s$),^{16,17} and the intensity was found to be rapidly changed at the beginning and at the terminal of the square-wave laser pulse, respectively. However, the output laser pulse shape of 532 nm DPSSL was not square-wave pulse due to the long rise and fall time. As indicated in Fig. 3, no rapid changes in the ${}^{1}O_{2}$ luminescence were observed. For the CW excitation, the ${}^{1}O_{2}$ was generated continuously and its luminescence reached a steady state during irradiation, which was in accordance with Eq. (5).

3.2. ${}^{1}O_{2}$ production varied with pulse width

As shown in Fig. 4(a), the peak intensity of ${}^{1}O_{2}$ luminescence decreased with the increase in pulse



Fig. 2. Typical time-resolved luminescence spectra of three individual wavelengths (1230, 1270, 1310 nm) and the corresponding ${}^{1}O_{2}$ luminescence from photo-irradiation of 1.00 μ M RB in PBS with pulse widths of (a) 20.7 ns at 523 nm and (b) 10.7 μ s at 532 nm, (c) CW at 532 nm; (d) the time-integrated luminescence spectra of (a), (b) and (c).

width, while the duration of ${}^{1}O_{2}$ luminescence increased with the increase in pulse width. The ${}^{1}O_{2}$ production kept constant when CW mode was applied. Figure 4(b) indicates the time-integrated ${}^{1}O_{2}$

luminescence intensity calculated by integrating the total ${}^{1}O_{2}$ luminescence counts after background subtraction. No significant difference (P > 0.05) in ${}^{1}O_{2}$ production was found under the excitation



Fig. 3. Temporal profiles for ${}^{1}O_{2}$ luminescence with two different pulse widths excitation: (a) simulated curves using the equation models with $\tau_{D} = 4 \,\mu s$, $\tau_{T} = 2 \,\mu s$, $\Phi_{D} = 0.76$, $\Phi_{T} = 0.9^{12,15}$; and (b) the measured results.



Fig. 4. (a) The typical time-resolved ${}^{1}O_{2}$ luminescence kinetics after background subtraction from RB 1.00 μ M in PBS, and (b) the corresponding integrated ${}^{1}O_{2}$ luminescence intensity with pulse width of 20.7 ns at 523 nm, and pulse widths of 10.7, 14.0, 19.9, 28, 36 μ s and CW at 532 nm. *P < 0.05.

between the 532 nm CW laser and 532 nm pulsed laser with various pulse widths of microsecond, while a significant difference (P < 0.05) in ${}^{1}O_{2}$ production was observed between 532 nm CW laser and 523 nm pulsed laser with pulse width of 20.7 ns.

It should be noted that the absorbance of RB in PBS at 532 nm was measured to be 0.030, which was about 1.5 times than 0.020 measured at 523 nm. Our previous study showed that the ${}^{1}O_{2}$ production of RB in solution depend linearly on the absorbance up to 0.248.³ The difference in ${}^{1}O_{2}$ production between 532 nm CW laser and 523 nm pulsed laser with pulse width of 20.7 ns might be mainly attributed to the different absorbance of RB at different wavelengths in that no significant difference (P > 0.05) was found in ${}^{1}O_{2}$ production between them after the ${}^{1}O_{2}$ production of 523 nm pulsed laser multiplied by 1.5 (data not shown).

3.3. ${}^{1}O_{2}$ production varied with pulse repetition rates

As shown in Figs. 5(a) and 5(b), the time-integrated ${}^{1}O_{2}$ luminescence intensity was calculated by integrating the total ${}^{1}O_{2}$ luminescence counts after background subtraction. No significant difference (P > 0.05) was found in ${}^{1}O_{2}$ production between CW and pulsed 532 nm laser with various repetition rates.

 $^{1}\text{O}_{2}$ is known to the major cytotoxic agent in type-II PDT, and its amount is the direct indicator for PDT dosimetry.⁷ Wei *et al.* investigated the influence of the fluence rates (15, 45, 135 mW/cm² with two fluences 15 J and 75 J, respectively) with light wavelength of 635 nm on the PDT treatment outcome for mouse tumor model by using chemiluminescence, which provided a selective mean for $^{1}\text{O}_{2}$



Fig. 5. (a) The typical time-resolved ${}^{1}O_{2}$ luminescence kinetics after background subtraction from RB 1.00 μ M in PBS, and (b) the corresponding time-integrated ${}^{1}O_{2}$ luminescence intensity with repetition rates of 1, 5, 10, 15 kHz and CW at 532 nm.

detection.¹⁸ The results indicate that a good correlation between the cumulative chemiluminescence and the PDT treatment outcome.¹⁸ In addition, less ¹O₂ was produced at a higher light fluence rate.¹⁸ One possible reason is rapid depletion of oxygen due to PDT photochemical reactions that induced a local hypoxia. In our study, ¹O₂ was directly measured by its luminescence at 1270 nm generated from photosensitization of RB solution in cuvette, which was exposed to room air at the top. Therefore, no significant reduction in ¹O₂ was observed when using pulsed laser with high peak fluence.

Repetition rate and pulse width are two important parameters determining the effectiveness of PDT when pulsed lasers are utilized,¹⁹ and the peak power is inversely proportional to the repetition rate and pulse width. In principle, a very high peak power may saturate the PS and thus limit ${}^{1}O_{2}$ generation.²⁰ According to a previous study, the effectiveness of pulsed excitation in PDT is comparable to that of CW excitation for the peak power density below $4 \times 10^4 \,\mathrm{W/cm^{2.20}}$ Above the threshold, the PDT efficiency drops significantly. In this study, the maximum peak power density obtained from the Q-switch 523 nm laser with pulse width of 20.7 ns at the power of 20 mW was 1.2×10^4 W/ cm^2 , which was below this threshold. Therefore, no significant decrease in the ${}^{1}O_{2}$ production was observed, as expected.

In fact, the clinical PDT efficacy could be influenced by the light penetration in tissues, PS concentration, molecular oxygen concentration and so on, all of which maybe varied during PDT treatments.²¹ In this study, all the measurements were performed in the RB solution. Therefore, we do not have to take into account the factors such as light penetration depth, PS distribution and so on, and allow focus on discussing the effects of the laser parameters on the kinetics and production of ${}^{1}O_{2}$, which is the critical cytotoxic species that determines the PDT efficacy. For the first time, the effects of laser pulse width range from nanosecond to CW on the kinetics and production of ${}^{1}O_{2}$ luminescence were directly compared. No significant difference was observed in ¹O₂ production among various pulse widths and repetition rates with the same average irradiation power of 20 mW, which implies that the efficacy of PDT using pulsed laser may be identical to the CW laser with the same wavelength and the same average fluence rate in solution. However, a number of studies compared CW and pulsed lasers for PDT in $vitro^{22-25}$ and in vivo,^{26–29} and the pulsed laser irradiation found to be superior to, 22,26 equivalent to, $^{27-29}$ or inferior to^{23-25} CW laser in PDT efficacy by comparing the cytotoxicity ratio of cells or depth of necrosis. Possible reasons for the discrepancies are likely to be: first, various pulse widths $(10^{-1}-10^{-9} s)$ and pulse repetitions $(10^1 - 10^4 \text{ Hz})$ were used in the previous studies, and the pulsed laser with short pulse width provides much higher peak power, which may saturate ${}^{1}O_{2}$ production and thus limit photodynamic effect; second, the peak intensity and duration of $^{1}O_{2}$ production varying with the pulse width observed in this study might lead to different cytotoxic effect in PDT. In this regard, the peak power should be carefully determined if ${}^{1}O_{2}$ production is saturated when using pulsed laser.

4. Conclusions

The effects of pulse width and repetition rate of pulsed laser on the kinetics and production of ${}^{1}O_{2}$ luminescence were quantitatively investigated during photosensitization, which was also compared to the CW laser excitation. Significant difference was found for the kinetics of ${}^{1}O_{2}$ luminescence under the excitation with various pulse widths of nanosecond, microsecond and CW excitation, while no significant difference was observed in ${}^{1}O_{2}$ production among various pulse widths and repetition rates with the same average irradiation power of 20 mW. The results suggest that the PDT efficacy using pulsed laser may be identical to the CW laser with the same wavelength and the same average fluence rate below a threshold in solution.

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