

First-principles study on the conventional superconductivity of N-doped $fcc\text{-LuH}_3$

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ABSTRACT

Recently, room-temperature superconductivity has been reported in a nitrogen-doped lutetium hydride at near-ambient pressure [Dasenbrock-Gammon *et al.*, Nature **615**, 244 (2023)]. The superconducting properties might arise from $Fm\bar{3}m\text{-LuH}_{3-\delta}\text{N}_\epsilon$. Here, we systematically study the phase diagram of Lu–N–H at 1 GPa using first-principles calculations, and we do not find any thermodynamically stable ternary compounds. In addition, we calculate the dynamic stability and superconducting properties of N-doped $Fm\bar{3}m\text{-LuH}_3$ using the virtual crystal approximation (VCA) and the supercell method. The $R3m\text{-Lu}_2\text{H}_5\text{N}$ predicted using the supercell method could be dynamically stable at 50 GPa, with a T_c of 27 K. According to the VCA method, the highest T_c is 22 K, obtained with 1% N-doping at 30 GPa. Moreover, the doping of nitrogen atoms into $Fm\bar{3}m\text{-LuH}_3$ slightly enhances T_c , but raises the dynamically stable pressure. Our theoretical results show that the T_c values of N-doped LuH_3 estimated using the Allen–Dynes–modified McMillan equation are much lower than room temperature.

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I. INTRODUCTION

Hydrides have received much attention because of their excellent superconductivities under pressure.^{1–4} In 2014, H_3S was predicted to be a high-temperature superconductor with a T_c of 191–204 K,⁵ which was confirmed by the experimentally measured T_c of 203 K at 155 GPa.^{6,7} Following this success, several superhydrides with high T_c values above 200 K, such as LaH_{10} (260 K, 170 GPa), YH_6 (224 K, 166 GPa), YH_9 (260 K, 201 GPa), and CaH_6 (215 K, 172 GPa), were predicted and synthesized.^{8–18} Recently, Song *et al.*¹⁹ comprehensively studied the Lu–H system and predicted that $Im\bar{3}m\text{-LuH}_6$ should have a T_c of 273 K at 100 GPa. Extensive experimental research has been conducted on the superconducting properties of lutetium hydrides under pressure. $Fm\bar{3}m\text{-LuH}_3$ has been successfully synthesized and found to have a T_c of 12 K at 122 GPa.²⁰ Moreover, $Pm\bar{3}n\text{-Lu}_4\text{H}_{23}$ has been obtained, with a T_c of 71 K at 218 GPa.²¹ In particular, a very recent experimental study has reported superconductivity in the Lu–N–H

system at near-ambient pressure (~ 1 GPa), with the highest T_c to date of 294 K,²² thus achieving room-temperature superconductivity near ambient pressure. The authors of this study attributed the room-temperature superconductivity in the Lu–N–H system to $Fm\bar{3}m\text{-LuH}_{3-\delta}\text{N}_\epsilon$, which can be regarded as $Fm\bar{3}m\text{-LuH}_3$ doped with nitrogen atoms.

This report could represent a landmark event for the scientific community, but many open questions surrounding this important discovery remain unanswered, for example, the exact stoichiometry and the positions of the hydrogen and nitrogen atoms. A subsequent experimental observation indicates that the pressure-induced color change of LuH_2 is similar to that of N-doped lutetium hydride.²³ In addition, a similar compound ($\text{LuH}_{2\pm x}\text{N}_y$) has been synthesized, but no evidence for superconductivity at pressures ranging from 1 to 6 GPa has been found.²⁴ Density functional theory (DFT) calculations have been used to investigate the optical properties of lutetium hydrides²⁵ and the phase diagram of the Lu–N–H system at 0, 5, and 10 GPa,²⁶ but no thermodynamically stable

ternary compounds have been found. Therefore, an investigation of the superconducting properties of N-doped lutetium hydrides is necessary.

In the work described in this paper, we performed a comprehensive first-principles study on the Lu–N–H system at 1 GPa. No thermodynamically stable Lu–N–H ternary compounds were found using the structure search method. We then analyzed the dynamic stability and superconducting properties of N-doped $Fm\bar{3}m$ -LuH₃ using the virtual crystal approximation (VCA) and the supercell method. Using the Allen–Dynes–modified McMillan (A–D–M) equation, we found that the estimated highest T_c of N-doped LuH₃ did not exceed 30 K, which is much lower than room temperature. In addition, we found that the doping of nitrogen atoms may slightly enhance T_c but also increase the dynamically stable pressure of LuH₃.

II. COMPUTATIONAL DETAILS

At 1 GPa, we performed a variable-composition crystal structure search in the Lu–N–H system with $\sim 10\,000$ structures using the *Ab Initio* Random Structure Searching (AIRSS)²⁷ code. We then re-optimized the structures using the *ab initio* calculation from the Cambridge Serial Total Energy Package (CASTEP).²⁸ An on-the-fly ultrasoft pseudopotential with valence electrons $1s^1$ for H, $2s^2 2p^3$ for N, and $4f^{14} 5s^2 5p^6 5d^1 6s^2$ for Lu was used, with a kinetic cutoff energy of 800 eV. The Brillouin zone was sampled using a k -point mesh of $2\pi \times 0.03 \text{ \AA}^{-1}$ to make the enthalpy calculations converge well to less than 1 meV/atom. Structural relaxations were performed using projector-augmented wave (PAW)^{29,30} potentials, as implemented in the Vienna *Ab Initio* Simulation Package (VASP),³¹ with a cutoff energy of 600 eV. The exchange–correlation functional was described using the Perdew–Burke–Ernzerhof generalized gradient approximation.³²

We then prepared a conventional cell of $Fm\bar{3}m$ -LuH₃, including four formula units (f.u.). The calculations were performed for a range of pseudostoichiometric Lu₄H_{*m*}N_{12–*m*}, including Lu₄H₁₁N, Lu₂H₅N, Lu₄H₉N₃, and LuH₂N, which correspond to nitrogen-to-hydrogen atomic ratios of 0.09, 0.2, 0.33, and 0.5, respectively. Next, we calculated the formation enthalpies and superconducting properties of these structures at 1, 10, and 50 GPa. The formation enthalpy of Lu₄H_{*m*}N_{12–*m*} was calculated using the following equation:

$$H^f = H(\text{Lu}_4\text{H}_m\text{N}_{12-m}) - 4H(\text{LuH}_3) - mH(\text{N}) + mH(\text{H}). \quad (1)$$

We investigated the pressure dependence of the superconductivity of $Fm\bar{3}m$ -LuH₃ at 0.5%–2% doping with N using the VCA at pressures below 100 GPa. The VCA was performed by generating a virtual pseudopotential V_{VCA} of H_{1–*x*}N_{*x*}, where $V_{\text{VCA}} = (1-x)V_{\text{H}} + xV_{\text{N}}$. Furthermore, we calculated the equation of state for LuH_{2.97}N_{0.03}, which is close to the elemental analysis data from previous experiments,²² using DFT and DFT + U (with $U = 5.5$ eV, which has been used for metal mononitrides²²). The result is shown in Fig. S1 (supplementary material). The DFT result without the Hubbard U effect fits better with the experimental data. Therefore, the DFT level calculation for N-doped LuH₃ is acceptable.

Electronic structures, phonon spectra, and electron–phonon coupling (EPC) were calculated using the QUANTUM ESPRESSO (QE)³³ package. The PAW pseudopotentials with valence electrons $1s^1$ for H, $2s^2 2p^3$ for N, and $5s^2 5p^6 5d^1 6s^2$ for Lu were used in the QE package. The self-consistent electron density was evaluated using a k -mesh of $20 \times 20 \times 20$. The phonon spectra and EPC were calculated using a q -mesh of $5 \times 5 \times 5$. The conventional superconducting transition temperature was estimated using the A–D–M equation³⁴

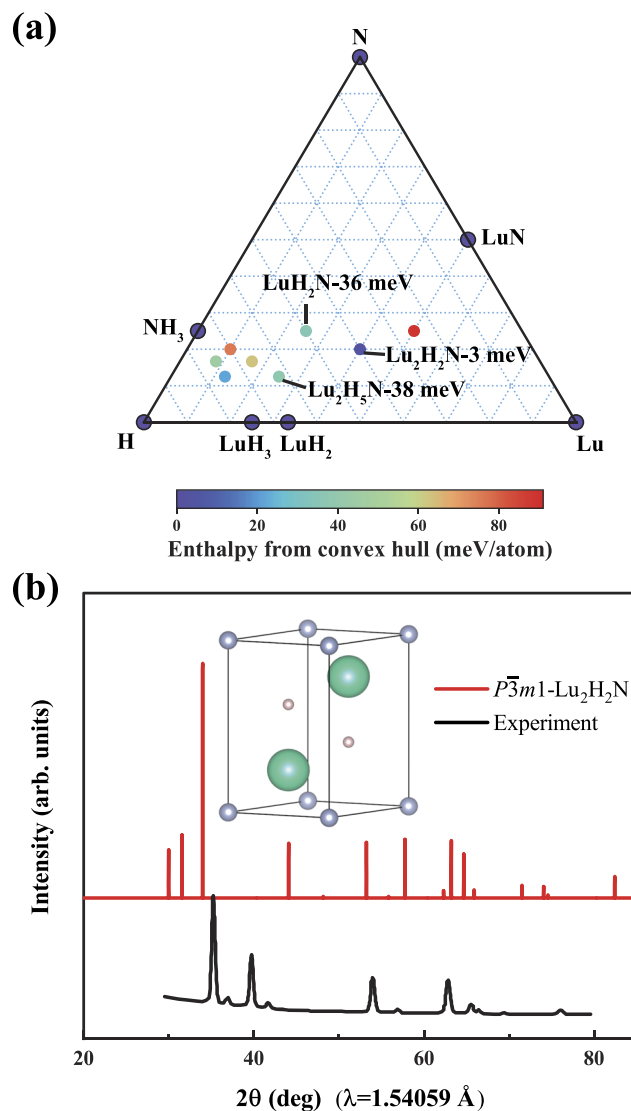


FIG. 1. (a) Ternary phase diagram (convex hull) of the Lu–N–H system at 1 GPa. The values of the enthalpy from the convex hull are shown next to the colored circles. The circles with black solid boundaries denote the stable phases. (b) Comparison of the XRD pattern from reported experimental data²² (lower black curve) and the pattern calculated for $P\bar{3}m1$ -Lu₂H₂N at ambient pressure (upper red curve). The inset shows the crystal structure of $P\bar{3}m1$ -Lu₂H₂N, with green, gray, and pink spheres depicting Lu, N, and H atoms, respectively.

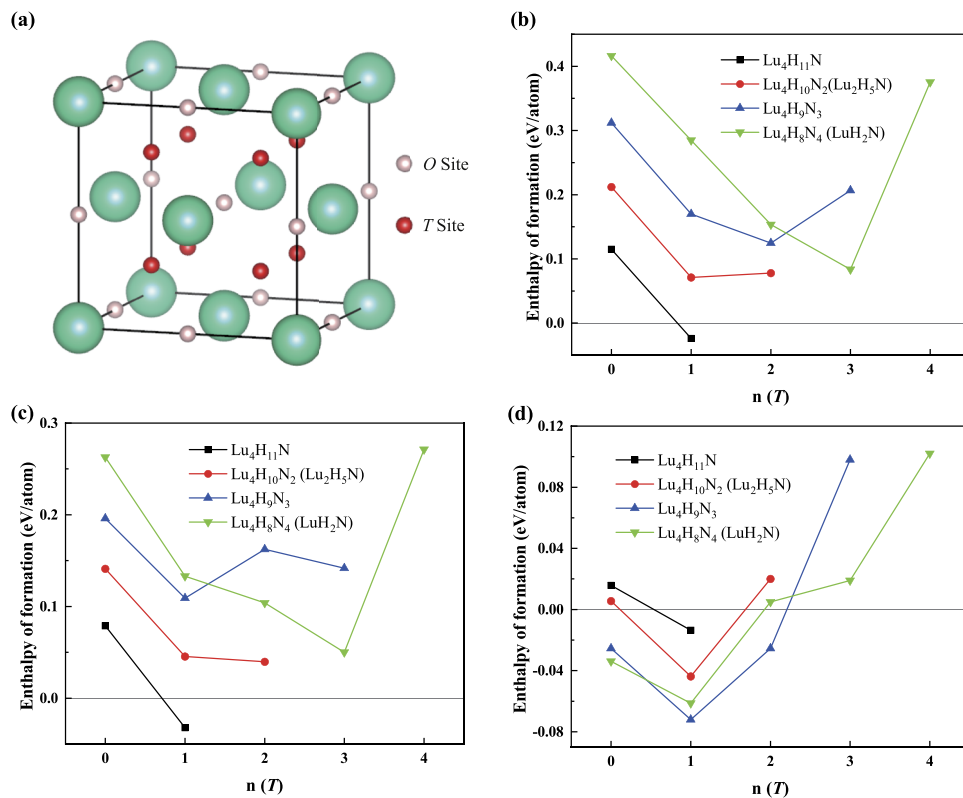


FIG. 2. (a) Crystal structure of $Fm\bar{3}m$ - LuH_3 with different types of H occupancy sites. Green spheres depict Lu atoms; pink and red spheres depict H occupancy of O and T sites, respectively. (b)–(d) Calculated enthalpy of formation of $\text{Lu}_4\text{H}_m\text{N}_{12-m}$ as a function of T-site occupancy at 1, 10, and 50 GPa, respectively. $n(T)$ is the number of nitrogen atoms in each stoichiometry occupying T sites.

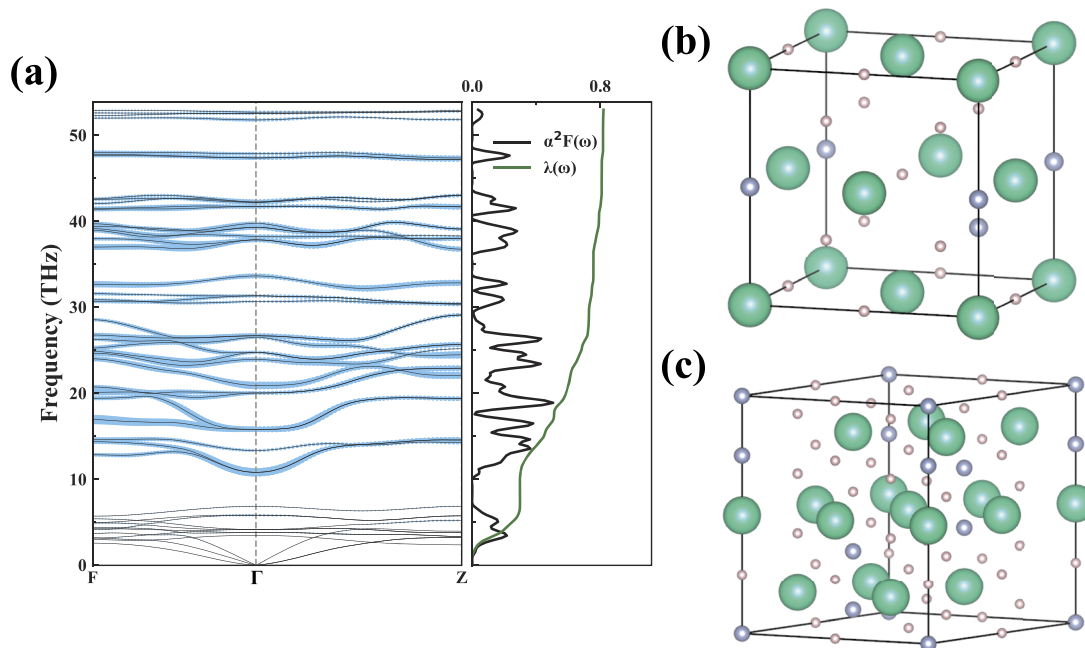


FIG. 3. (a) Phonon spectra, Eliashberg phonon spectral function $\alpha^2F(\omega)$, and electron–phonon integral λ for $R3m$ - $\text{Lu}_2\text{H}_5\text{N}$ at 50 GPa. The magnitudes of the phonon linewidths are represented by the sizes of the blue circles. (b) and (c) Crystal structures of $R3m$ - $\text{Lu}_2\text{H}_5\text{N}$ in the framework of $Fm\bar{3}m$ - LuH_3 and the conventional cell, respectively. Green, gray, and pink spheres depict Lu, N, and H atoms, respectively.

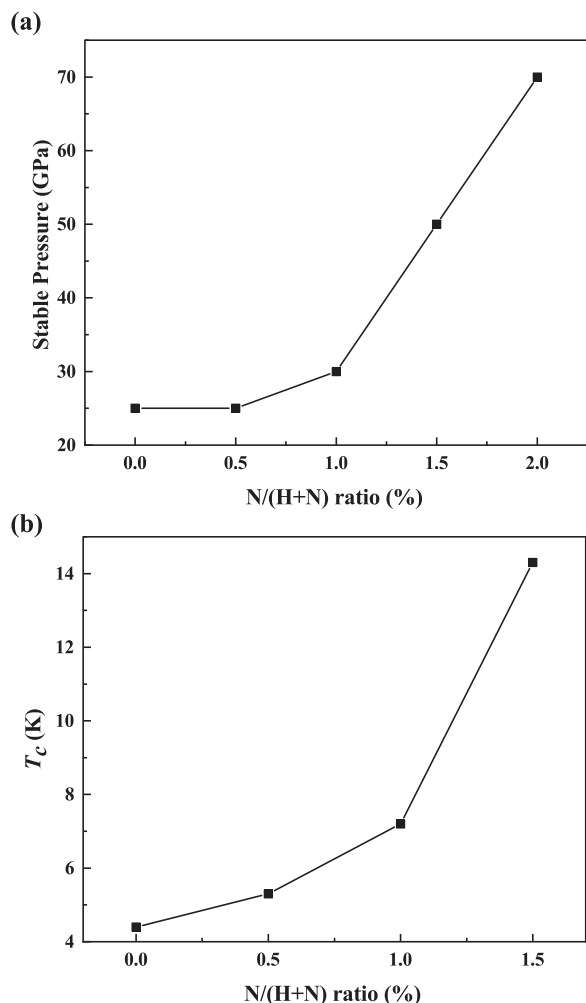


FIG. 4. (a) Dependence of minimum dynamically stable pressure on N-doping concentration. (b) Dependence of T_c on N-doping concentration at 50 GPa. The Coulomb pseudopotential uses $\mu^* = 0.10$.

with correction factors and a Coulomb pseudopotential³⁵ with $\mu^* = 0.10$ or 0.13.

III. RESULTS AND DISCUSSION

We performed a random structure search in the Lu–N–H system and constructed the ternary phase diagram (convex hull) at 1 GPa [Fig. 1(a)]. Some of the binary compounds were adopted from previous papers.^{36–39} Notably, all predicted potential ternary compounds lie above the convex hull at 1 GPa. Thus, no ternary Lu–N–H compounds can remain thermodynamically stable at this pressure, which is consistent with the main results of Xie *et al.*²⁶ In addition, $P\bar{3}m1$ -Lu₂H₂N was found with an enthalpy of ~ 3 meV/atom above the convex hull. Detailed information about the structural parameters is listed in Table S1 (supplementary material). According to the inorganic crystal structure database, 20% of experimentally

synthesized materials are metastable.^{40,41} Therefore, we calculated the x-ray diffraction (XRD) pattern for this metastable compound. Figure 1(b) shows a comparison of experimental and calculated XRD patterns. The calculated XRD pattern of $P\bar{3}m1$ -Lu₂H₂N deviates clearly from the experimental one,²² indicating that this compound does not occur in high-pressure experiments on N-doped lutetium hydride superconductors.

Next, we investigated the doping effect on N-doped LuH₃ using the supercell method. The cubic cell of Lu₄H_{*m*}N_{12–*m*} was constructed by replacing hydrogen atoms with nitrogen atoms, for $m = 8$ –11. For each concentration, we performed a geometry optimization on the configurations of various octahedral (*O*) and tetrahedral (*T*) sites [see Fig. 2(a)] and then calculated the total energy. Figures 2(b)–2(d) show the results for the formation enthalpy of Lu₄H_{*m*}N_{12–*m*} at different pressures. At 1 GPa, only the formation enthalpy of Lu₄H₁₁N is negative (~ -24 meV/atom), indicating that this compound may be more stable than the others. A similar situation also occurs at 10 GPa (where the formation enthalpy is ~ -32 meV/atom for Lu₄H₁₁N). Thus, only Lu₄H₁₁N is thermodynamically encouraged to be formed below 10 GPa. At 50 GPa, Lu₄H₁₁N, Lu₂H₅N, Lu₄H₉N₃, and LuH₂N are thermodynamically encouraged to be formed. Thus, Lu₄H_{*m*}N_{12–*m*} with low nitrogen doping is enthalpically favored at low pressure. The doping concentration of nitrogen increases with increasing pressure. In addition, when one nitrogen atom occupies a *T* site, the formation enthalpy of Lu₄H_{*m*}N_{12–*m*} at 50 GPa is the lowest. Therefore, one nitrogen atom occupying a *T* site in Lu₄H_{*m*}N_{12–*m*} is enthalpically preferred.

We then calculated the phonon spectra and superconducting properties of Lu₄H_{*m*}N_{12–*m*}. T_c values were estimated using the A–D–M equation with correction factors (Table S2, supplementary material). At 1 and 10 GPa, Lu₄H_{*m*}N_{12–*m*} cannot dynamically stabilize, which may be due to the thermodynamical⁴² and dynamical instability of the parent $Fm\bar{3}m$ -LuH₃ [see Fig. S2(b), supplementary material]. At 50 GPa, only $R\bar{3}m$ -Lu₂H₅N can dynamically stabilize. The crystal structure, phonon spectrum, and EPC of $R\bar{3}m$ -Lu₂H₅N are illustrated in Fig. 3. $R\bar{3}m$ -Lu₂H₅N (Lu₄H₁₀N₂) can be regarded as $Fm\bar{3}m$ -LuH₃ with two nitrogen atoms substituting its *T* and *O* sites, respectively. The calculated phonon spectrum and EPC show that the contributions of medium- and low-frequency phonons (13–25 THz) to the EPC are the highest (about 46% to total λ), whereas high-frequency phonons (45–55 THz) make hardly any contribution. Thus, the EPC of $R\bar{3}m$ -Lu₂H₅N is primarily contributed by medium- and low-frequency phonons. The calculated EPC parameter λ for $R\bar{3}m$ -Lu₂H₅N at 50 GPa in the harmonic approximation is 0.82. Using the calculated logarithmic average frequency ω_{\log} , along with a Coulomb pseudopotential μ^* value of 0.1, the resultant T_c value is 27 K.

In the case of low nitrogen doping concentrations, we used the VCA method to investigate the pressure dependence of T_c in N-doped $Fm\bar{3}m$ -LuH₃. Figure 4(a) shows the dependence of the minimum dynamically stable pressure on the N-doping concentration. We find that the minimum dynamically stable pressure of N-doped $Fm\bar{3}m$ -LuH₃ increases from 25 to 70 GPa when the doping concentration increases from 0% to 2%. Thus, doping with N atoms will raise the dynamically stable pressure of LuH₃. Through a softening mechanism,⁴³ pressure can affect T_c by altering the electron–phonon constant λ . Therefore, we calculated the T_c of

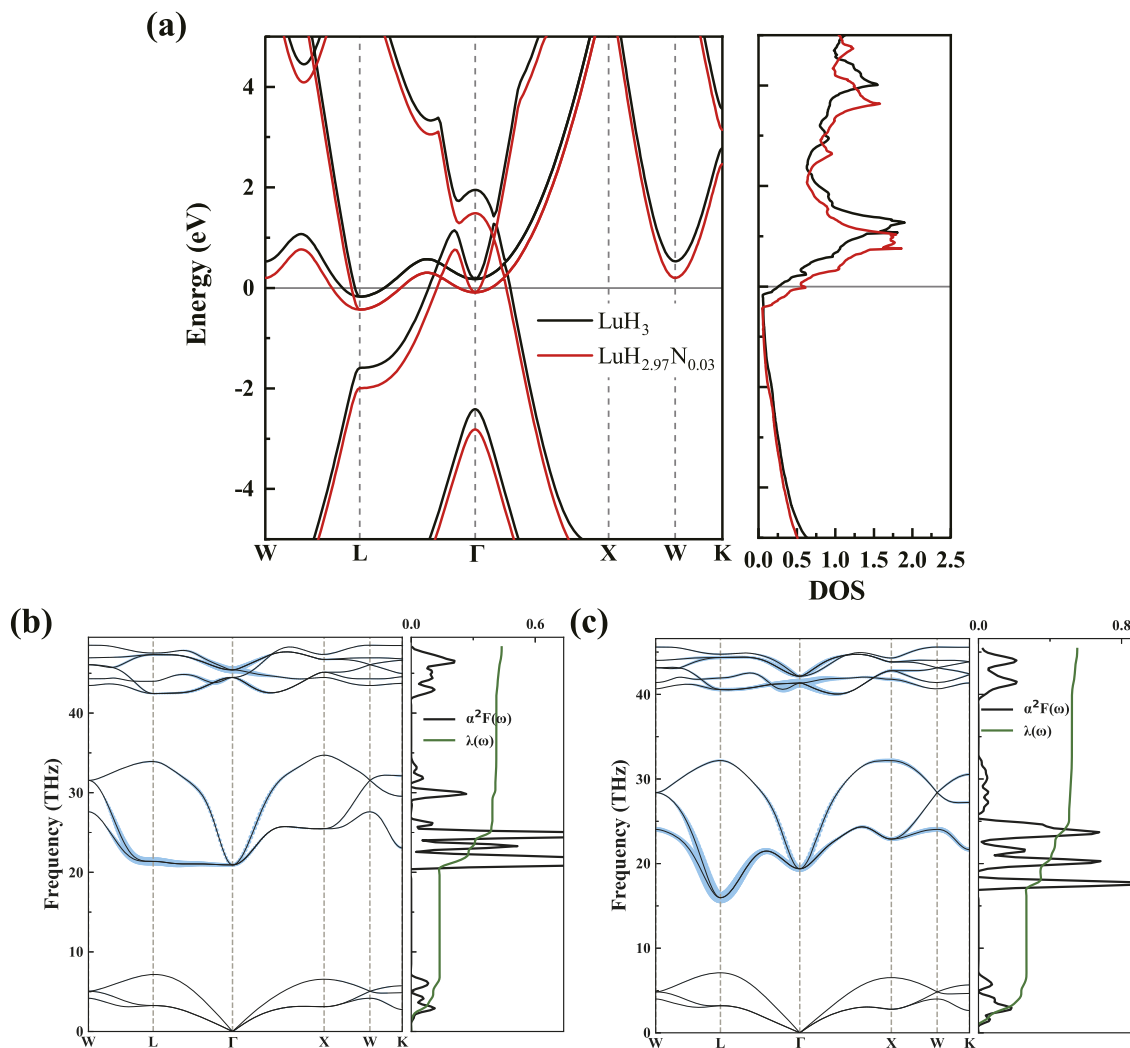


FIG. 5. (a) Electronic band structure and DOS (states $\text{eV}^{-1} \text{f.u.}^{-1}$) of LuH_3 (solid black curves) and $\text{LuH}_{2.97}\text{N}_{0.03}$ (solid red curves) at 50 GPa. The grey solid horizontal line indicates the Fermi energy. (b) and (c) Phonon spectra, Eliashberg phonon spectral function $\alpha^2F(\omega)$, and electron-phonon integral λ for LuH_3 and $\text{LuH}_{2.97}\text{N}_{0.03}$, respectively, at 50 GPa. The magnitudes of the phonon linewidths are represented by the sizes of the blue circles.

N-doped LuH_3 with doping concentrations ranging from 0% to 1.5% at 50 GPa [see Fig. 4(b)] to investigate the effects of doping with nitrogen atoms on superconductivity. Our simulations show that the lowest T_c is 4 K for LuH_3 without doping. In addition, T_c increases with increasing N-doping concentration; thus, doping N atoms into LuH_3 will increase T_c . However, the highest T_c in this VCA calculation is 22 K, obtained with 1% N-doping at 30 GPa (see Table S2, supplementary material), which is much lower than room temperature. Notably, the anharmonicity correction to the atomic motions imposed by the large ionic quantum fluctuation will renormalize the phonon frequency in hydrogen-based superconductors. Therefore, the anharmonic effect may potentially decrease the dynamically stable pressure of N-doped LuH_3 and affect T_c , which requires further theoretical investigation in this system.

We calculated the band structure and density of states (DOS) of LuH_3 and $\text{LuH}_{2.97}\text{N}_{0.03}$ (1% N-doped) at 50 GPa [Fig. 5(a)]. The finite DOS at the Fermi level indicates the metallic nature of these structures. LuH_3 has a DOS $N(\epsilon_F)$ that reaches 0.269 states $\text{eV}^{-1} \text{f.u.}^{-1}$ at the Fermi level. Doping N atoms into LuH_3 will raise the Fermi level and cause the bands near the Γ point to fall on top of the Fermi level. Consequently, the Fermi level can be moved closer to the DOS peak, and $N(\epsilon_F)$ will then increase to 0.6 states $\text{eV}^{-1} \text{f.u.}^{-1}$. Therefore, doping by N atoms can significantly enhance the metallic characteristic of LuH_3 by moving the Fermi level closer to the DOS peak.

We then examined the phonon spectra and EPC of LuH_3 and $\text{LuH}_{2.97}\text{N}_{0.03}$ at 50 GPa [Figs. 5(b) and 5(c)]. In LuH_3 , the electron-phonon constant λ is primarily contributed by the optical

branch (λ_{opt}), which accounts for ~69% of the total λ . Conversely, the acoustic branch (λ_{ac}) makes a much smaller contribution to λ , accounting for ~31% of the total λ . Doping by nitrogen atoms at 1% concentration increases λ_{ac} from 0.137 to 0.268, while λ_{opt} remains almost unchanged. Therefore, the increased T_c of N-doped LuH₃ can be attributed to the enhancement of λ_{ac} . Interestingly, the optical branch with a frequency of ~21 THz along the $W-L-T$ direction in LuH₃ varies slightly with the wave vector \mathbf{q} in the Brillouin zone. However, a significant softening of this phonon mode is observed after doping by nitrogen atoms [Fig. 5(c)]. When the pressure decreases to 25 GPa (Fig. S4, supplementary material), the imaginary phonon frequency occurs near the L point, which indicates that the optical branch softening induced by N doping is responsible for the reduced stability of LuH₃.

IV. CONCLUSIONS

We performed a first-principles study on the Lu–N–H system and found no stable ternary compounds at 1 GPa. We then analyzed N-doped $Fm\bar{3}m$ -LuH₃ using supercell and VCA methods. The result of the supercell method indicates that $R3m$ -Lu₂H₅N can be dynamically stable at 50 GPa, with a T_c of 27 K. The VCA results indicate that the highest T_c is 22 K, obtained with 1% N-doping at 30 GPa. In addition, doping with nitrogen atoms slightly increases the T_c of $Fm\bar{3}m$ -LuH₃ by enhancing the EPC of acoustic phonons. However, this doping effect also leads to significant phonon softening and increases the dynamically stable pressure of LuH₃. Finally, within the pressure range investigated in our study, the highest T_c of N-doped $Fm\bar{3}m$ -LuH₃ does not exceed 30 K, which is much lower than room temperature. Our theoretical calculations were performed using standard DFT parameters and assuming conventional superconductivity, and calculations considering more corrections, including strong electron correlations, spin–orbit coupling, the anharmonicity effect, and unconventional mechanisms of superconductivity are required in the future.

SUPPLEMENTARY MATERIAL

See the supplementary material for supplementary figures and tables.

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AUTHOR DECLARATIONS

Conflict of Interest

The authors have no conflicts to disclose.

Author Contributions

Zihao Huo: Conceptualization (equal); Data curation (equal); Formal analysis (equal); Writing – original draft (equal); Writing – review & editing (equal). **Defang Duan:** Conceptualization (lead); Formal analysis (lead); Funding acquisition (lead); Investigation (lead); Supervision (lead); Writing – review & editing (lead). **Tiancheng Ma:** Data curation (equal); Methodology (equal). **Zihan Zhang:** Conceptualization (supporting); Writing – review & editing (supporting). **Qiwen Jiang:** Writing – review & editing (supporting). **Decheng An:** Data curation (supporting). **Hao Song:** Conceptualization (supporting). **Fubo Tian:** Methodology (supporting). **Tian Cui:** Conceptualization (lead); Funding acquisition (lead); Methodology (equal); Supervision (equal); Writing – review & editing (equal).

DATA AVAILABILITY

The data that support the findings of this study are available within the article and its supplemental material and from the corresponding authors upon reasonable request.

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