

# Vacuum Packaging Sensor Based on Time-Resolved Phosphorescence Spectroscopy

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**Abstract:** Intelligent food packaging with the multisensory analysis is promising as the next generation technology of food packaging. The oxygen content in food packaging is one of the crucial parameters affecting the food quality and shelf life. Caviar is among the most nutritious and costly food sources. Here, a photonic oxygen-sensing system, based on the time-resolved phosphorescence spectroscopy of a platinum complex, is developed for non-contact, non-intrusive, and real-time vacuum packaging quality control, and implemented for caviar packaging. The sensor is embedded in protective polyethylene layers and excited with a short-pulsed light emitting diode (LED) source. Integration of a blue pulsed light source, a fast and amplified silicon photodiode controlled by the Spartan-6 field programmable gate array (FPGA), and a long lifetime platinum complex results in a photonics-based oxygen sensor with a fast response and high sensitivity to the vacuum packaging damage, which is suitable for caviar. It is revealed that applying the polyethylene layers protects the caviar from the platinum complex, leaching while not interfering with the sensor functionality. Characterizing the photonic system based on its sensitivity, repeatability, stability, and long-term operation demonstrates its capability for this application.

**Keywords:** Caviar; photoluminescence lifetime; oxygen sensor; platinum porphyrin complex; vacuum packaging

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## 1. Introduction

Nowadays, there is tremendous interest in technologies that make food packaging smart, because the expiration date on the package does not guarantee or provide information on the quality of packaging, transport, and storage of the product [1]. Moreover, even if all the standard regulations are followed, the lack of real-time information on the food status may leave consumers at risk due to foodborne illnesses [2]. Traditionally, a food package, as a passive barrier, primarily helps to protect food from environmental parameters, such as

moisture, light, oxygen (O<sub>2</sub>), microbes, chemical and mechanical stresses, and dust [3].

In recent decades, due to the growing interest of the final consumers in the freshness, quality, and safety assurance, two main strategies have been pursued to enhance the current packaging systems, namely, active and intelligent packaging [4]. Active packaging is realized by applying additives, such as gas scavengers, temperature controllers, moisture controllers, and antimicrobial and antioxidant-releasing components to increase the food quality and prolong its shelf-life. Vacuum and modified atmosphere packaging (MAP) technologies are the

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most commonly used concepts in active packaging [5]. Intelligent packaging is a packaging system that continuously reports the parameters of a product to the customer throughout the supply chain based on the ability of sensing, detecting or recording the external/internal changes in the product [6]. The main concepts in intelligent packaging are indicators or sensors. Indicators often exhibit qualitative information about packaged food based on a visible color change [7]. For instance, the UV-activated O<sub>2</sub> indicator inks usually contain redox dyes, electron donors, UV-absorbing photocatalysts, and polymers [8], which in the absence of O<sub>2</sub> lose their color, and in the presence of O<sub>2</sub> gain color [9]. Unlike the indicator systems, photonics-based O<sub>2</sub> sensors provide a prompt, reversible, and quantitative response to the non-destructive headspace analysis of the packaged food. Among them, photoluminescence (PL)-based O<sub>2</sub> sensors are the most advanced [10, 11]. The easy access of the photonic sensors to the packaging atmosphere, availability of the measurement tools, and protection of the sensor from the atmosphere outside the package are important requirements for integrating the sensors [12]. The membrane of these sensors consists of a phosphorescent oxygen-sensitive probe (POSP) embedded in a polymeric or sol-gel matrix [13, 14]. Commonly used the POSP includes a metal-organic complex of platinum (II), palladium (II) porphyrins, and ruthenium (II) since they exhibit a long lifetime, high O<sub>2</sub> sensitivity, a high molar extinction coefficient, excellent quantum yields, and long photostability [15, 16]. They have several significant advantages over other systems [17, 18], such as no O<sub>2</sub> consumption during measurement and non-invasive and contactless measurement in a sealed container. In addition, these sensors are affordable, disposable, easily miniaturized, and calibration-free. As a result, they can be used on a large scale. These sensors provide accurate and fast responses in real time [19, 20].

The most important criteria for selecting O<sub>2</sub>

sensors used for packaging applications are the planar shape of the sensing platform for the ease of integration, suitable solvent for low toxicity and fast evaporation, preventing leaching of the complex, insensitivity to the moisture, photostability, and long shelf-life stability. The application of an O<sub>2</sub>-permeable film laminated on top of the sensor, between the sensor and the food product, prevents the migration of the sensor's components into the food, which is vital from the food safety perspective. The POSP usually relies on the collisional quenching of the excited metal-organic complex by O<sub>2</sub> molecules, which decreases the emission intensity and corresponding lifetime depending on the O<sub>2</sub> concentration. The POSP molecules have a long emission lifetime than fluorescent dye molecules. As a result, they are more likely to interact with the O<sub>2</sub> molecules, thus reducing their emission intensity and lifetime accordingly [21]. In a homogeneous microenvironment, the O<sub>2</sub> quenching process usually follows the Stern-Volmer equation:

$$I_0/I = \tau_0/\tau = 1 + K_{sv}[O_2] \quad (1)$$

where  $I_0$  and  $\tau_0$  are the unquenched luminescence intensity and lifetime at the O<sub>2</sub> concentration of zero,  $I$  and  $\tau$  are the intensity and lifetime at the measured state,  $K_{sv}$  is the Stern-Volmer quenching constant, and  $[O_2]$  is the O<sub>2</sub> partial pressure [22]. The initial characterization for food packaging applications involves plotting the sensors' lifetime or intensity ratio based on the O<sub>2</sub> concentration to obtain a reproducible calibration function. Also, the sensitivity of these sensors varies based on the environmental temperature; therefore, calibration functions must be corrected based on the food packaging's temperature. Also, the sensor performance must be investigated through a comprehensive evaluation of the stability, reversibility, and repeatability to determine the POSP characteristics [23, 24]. The PL intensity is influenced by the factors, such as the fluctuation in the excitation light source, detector drift, optical

path, degradation or leaching of the metal-organic complex, thickness of the POSP, and concentration of the dye in the POSP. Instead, the PL lifetime is immune to the mentioned factors because it is an intrinsic property of the POSP. The time-domain and frequency-domain modes are commonly used to measure the PL lifetime. In the frequency-domain mode, the optical source is a sinusoidally modulated light at a fixed frequency, producing a modulated PL emission signal. The PL signal exhibits a phase shift corresponding to the PL lifetime that can be measured. In the time-domain mode, the POSP is excited by short light pulses. Then, the PL intensity decays exponentially, and the decay curve is fitted with an exponential function to calculate the PL lifetime, which exhibits better performance than frequency-domain sensors [25–27]. In recent years, O<sub>2</sub> sensors using the platinum octaethyl porphyrin ketone (PtOEPK) complex have been adopted to track O<sub>2</sub> levels in the cheddar cheese [28], bread [29], and chicken [30] packages utilizing the frequency-domain technology and in salad packages using the time-domain technology [31]. In another study, using the time-domain technology, an O<sub>2</sub> sensor comprising of the platinum benzoporphyrin (PtBP) complex, was used to track O<sub>2</sub> levels in meat packages [32].

Eggs separated from the ovaries of various sturgeons are commonly called caviar. They are used as energetic and nutritious food in many parts of the world [33]. The most valuable type of caviar is obtained from wild sturgeons, which for instance, are found in the Caspian Sea region, including beluga (*huso huso*), *asetra* (*acipenser persicus*), and *sevruga* (*acipenser stellatus*) caviars [34, 35]. The caviar contains high levels of lipids, phospholipids, long-chain polyunsaturated fatty acids (PUFA) [36], proteins [37], vitamin A, and B-complex vitamins [38]. Since the caviar contains a high percentage of proteins (22%–28%) and lipids (15%–78%), in addition, it does not go under pasteurization during processing, and the potential for microbial and chemical spoilage is high [39].

The caviar is packed under the vacuum condition, which effectively removes the possible biological and chemical contaminants from the surrounding environment [40]. However, factors, such as the poor air evacuation, poor sealing, exchange between O<sub>2</sub> trapped within the caviar eggs and the package headspace, and damage to the packaging during the vacuum packaging (VP) process or during transport and storage, can lead to O<sub>2</sub> leaking into the package [41] and cause the oxidation of lipids, and decomposition of proteins and discoloration. This in turn, creates an optimal condition for the growth and activity of all aerobic microorganisms that spoil the caviar [42]. Oxidation also reduces the caviar's nutritional value and changes the sensory properties such as the taste, texture, and aroma [43]. Thus, an accurate evaluation of the O<sub>2</sub> levels within the VP caviar packages makes it possible to identify the damaged packages and take the necessary steps [44].

In this paper, we develop a photonics-based O<sub>2</sub> sensor system for the caviar vacuum packaging quality control by integrating a composite of platinum meso-tetra (pentafluorophenyl) porphyrin (PtTFPP) in a polystyrene (PS) matrix embedded in two O<sub>2</sub>-permeable polyethylene layers with the high- and low-density, and a nanosecond pulsed light emitting diode (LED) with an amplified photodiode controlled by a field programmable gate array (FPGA) microcontroller. This sensor is implemented for real-time, non-intrusive, and contactless O<sub>2</sub> detection in caviar packages based on the time-resolved phosphorescence spectroscopy. The probe functionally is investigated through a comprehensive evaluation of the sensitivity, stability, reversibility, and repeatability. Finally, by embedding this sensor in vacuumed caviar packages, we demonstrate that it is possible to distinguish the damaged packages precisely based on the O<sub>2</sub> concentration.

## 2. Materials and methods

Fabrication of the POSP film: the PtTFPP, polystyrene, toluene solvent (99.9%), and

polyethylenes were purchased from Frontier Scientific, Sigma-Aldrich, Merck, and A.K.P.C, respectively. A solution was prepared by dissolving 2 mg of platinum meso-tetra (pentafluorophenyl) porphyrin powder, as an oxygen-sensitive complex, in 500  $\mu\text{L}$  of toluene as a solvent. Then, 0.2 g of polystyrene was dissolved, as the oxygen permeable matrix, in 1 mL of toluene. The solutions were thoroughly mixed using a magnetic stirrer at room temperature for 1 h to make a homogeneous solution. Subsequently, 50  $\mu\text{L}$  of each solution was poured into a vial and placed on a magnetic stirrer for 30 min. Finally, to make a circular film with a diameter of 8 mm, the solution was poured onto a glass slide, annealed overnight in an oven at 75  $^{\circ}\text{C}$ , and punched with a biopsy punch.

**Instrumentation:** the absorption spectrum was measured using Analytik Jena Specord 210 Plus. A Thorlabs CCS100 spectrometer was used for PL spectral analysis. A nanosecond pulsed LED with a wavelength of 450 nm was used as a source of excitation. The OSRAM SFH2704 silicon photodiode, with the dimensions of  $1 \times 1 \text{ mm}^2$ , accompanied by OPA354 and AD8042 rail-to-rail amplifiers, were used to detect the PL lifetime. The Spartan-6 FPGA from XILINX controlled all the electronic components. The Lutron YK-22DOA  $\text{O}_2$  meter was used to measure the  $\text{O}_2$  concentration in the air. DHT11 temperature and humidity sensors in combination with Arduino Uno were used to measure the temperature and humidity. The Spectro ARCOS system and Varian ICP-OES 730-ES ICP-OES were used for analyzing the PtTFPP leaching.

**Calibration setup:** the POSP was placed inside a temperature and gas-controlled plexiglass chamber to investigate its work function. The chamber had an inlet to supply  $\text{N}_2$  gas and a Lutron reference  $\text{O}_2$  sensor to monitor the  $\text{O}_2$  percentage. The  $\text{N}_2$  gas was then injected into the measurement chamber until the  $\text{O}_2$  concentration reached 0%, monitored by the reference oxygen meter. Then, the  $\text{N}_2$  injection stopped, and the PL lifetime values were recorded

for the oxygen concentration in a wide range from 0% to 20.9%.

### 3. Results and discussion

Figure 1 illustrates the schematic demonstrations of the photonics-based  $\text{O}_2$  sensor developed for the caviar vacuum packaging quality control and the corresponding detection mechanism. The POSP is optically excited with a nanosecond pulsed LED light source. Subsequently, excited POSP molecules are transitioned from the singlet to triplet states via intersystem crossing (ISC). The collision of the triplet-state molecular  $\text{O}_2$  with the excited POSP molecules results in non-radiative energy transferring to the  $\text{O}_2$  molecules. Therefore, it leads to PL quenching of the excited POSP, which appears as a decrease in the PL lifetime. In addition, the higher the molecular  $\text{O}_2$  concentration is, the greater the likelihood that the platinum complex film interacts with the  $\text{O}_2$  is, resulting in a higher level of lifetime quenching. The absorption spectrum of the POSP film, comprising the PtTFPP complex doped in a PS matrix, was measured by a spectrometer to determine the excitation wavelength. This spectrum enables us to select a suitable LED light source to excite the POSP optically. The absorption spectrum of the POSP is presented in Fig. 2(a) with a blue-green color showing three peaks at 392 nm, 510 nm, and 540 nm. Therefore, a 450-nm blue LED light source with the power of 4 W was selected for the POSP excitation. The PL spectrum was measured using a CSS100 spectrometer, and the measurement time was set to 1 000 ms. The PL spectrum is illustrated in Fig. 2(a) in red, with a maximum at 652 nm. Figure 2(b) shows the time-resolved PL decay of the POSP when it was excited with the LED pulses for two different arbitrary  $\text{O}_2$  concentrations at the temperature of  $(22.0 \pm 0.1) ^{\circ}\text{C}$ . According to the results, the PL intensity decays exponentially. The decay curve was fitted with a one-term exponential function using the Levenberg Marquardt algorithm to extract the lifetime of the POSP.

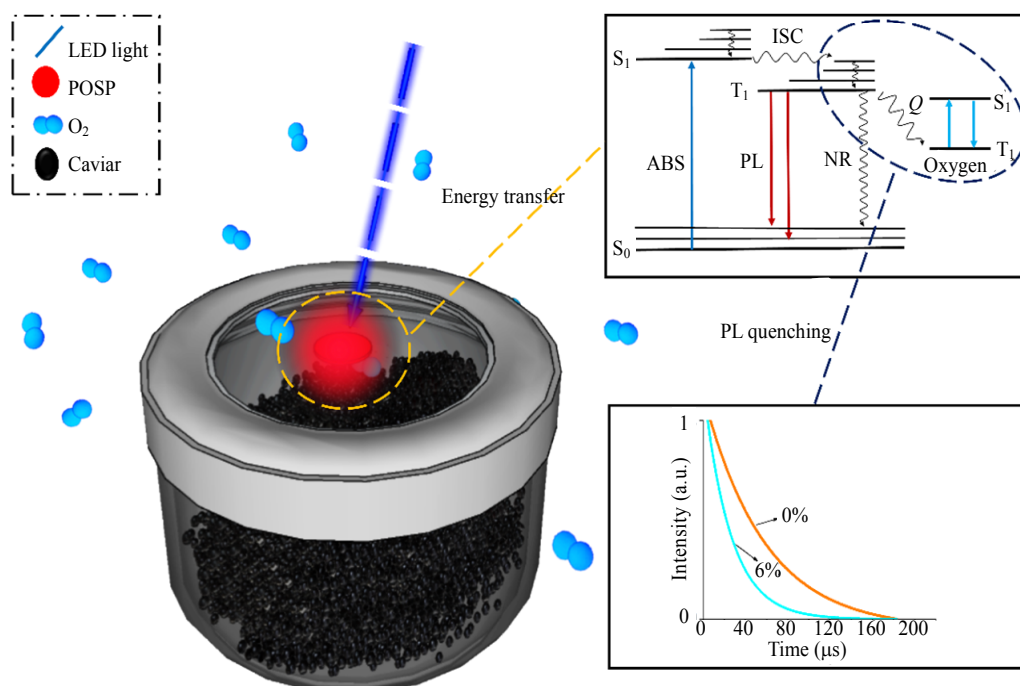


Fig. 1 Concept of the photonic sensor for caviar vacuum packaging quality control: energy diagram for energy transfer between the POSP and  $O_2$  molecules and the corresponding changes in the photoluminescence lifetime.

For instance, in Fig. 2(b), the lifetime decreases from  $53.5 \mu\text{s}$  to  $35.8 \mu\text{s}$  as the  $O_2$  concentration increases from 0% to 3.0% because as the  $O_2$  concentration is increased, the likelihood that the excited POSP interacts with the  $O_2$  molecules and non-radiative energy transferring to the  $O_2$  takes place is higher, resulting in a higher level of lifetime quenching. The time-resolved PL spectroscopy setup for  $O_2$  measurement is demonstrated in Fig. 2(c). The POSP was excited with multiple 50 ns pulses of the blue LED. The LED light was focused after passing a convex lens with the focal length of 1 cm to excite the POSP efficiently. The same lens collected the PL emission and guided it through a red filter to prevent the LED light from reaching the SFH2704 silicon photodetector (PD). Then, the PD signal was amplified by combining the OPA354 and AD8042 amplifiers. Afterward, the amplified analog signal was converted to digital data. Spartan-6 FPGA was employed to control all electronic components. All data were sent to a computer to be fitted with the one-term exponential function in order to calculate the lifetime. A

LabVIEW program was developed for the data acquisition and analysis. Figure 2(d) shows the concept that a multipulse LED source for the excitation of the POSP, with multiple 50-ns pulses and a total width of  $45 \mu\text{s}$ , is adjusted for efficient excitation.

The critical parameter of the POSP in food packaging applications is the sensitivity diagram which involves plotting the sensor's Stern-Volmer lifetime ratio ( $\tau_0/\tau$ ) based on the  $O_2$  concentrations in the working range of 0.0% to 20.9%, resulting in a reproducible calibration function. Figure 3(a) shows the lifetime ratio based on the  $O_2$  concentrations, where each point represents an average value of three times of consequent measurement, while the temperature was maintained at  $(24.0 \pm 0.1) ^\circ\text{C}$ . The results demonstrate that the observed behavior of the Stern-Volmer diagram is linear in the  $O_2$  concentration range of 0.0% to 20.9%. Therefore, the acquired data were fitted with a linear function in which the slope value was the Stern-Volmer constant. The sensitivity slope was calculated as 0.15. The graph's linearity indicates that the

dispersion of the PtTFPP molecules inside the matrix is almost homogeneous. In addition, the higher the molecular O<sub>2</sub> concentration is, the lower the PL lifetime is. The stability evaluation of the

PL lifetime was performed per every 1 min in a chamber at the O<sub>2</sub> concentration, temperature, and duration of 20.9%, 24.1 °C – 24.5 °C, and 90 min, respectively.

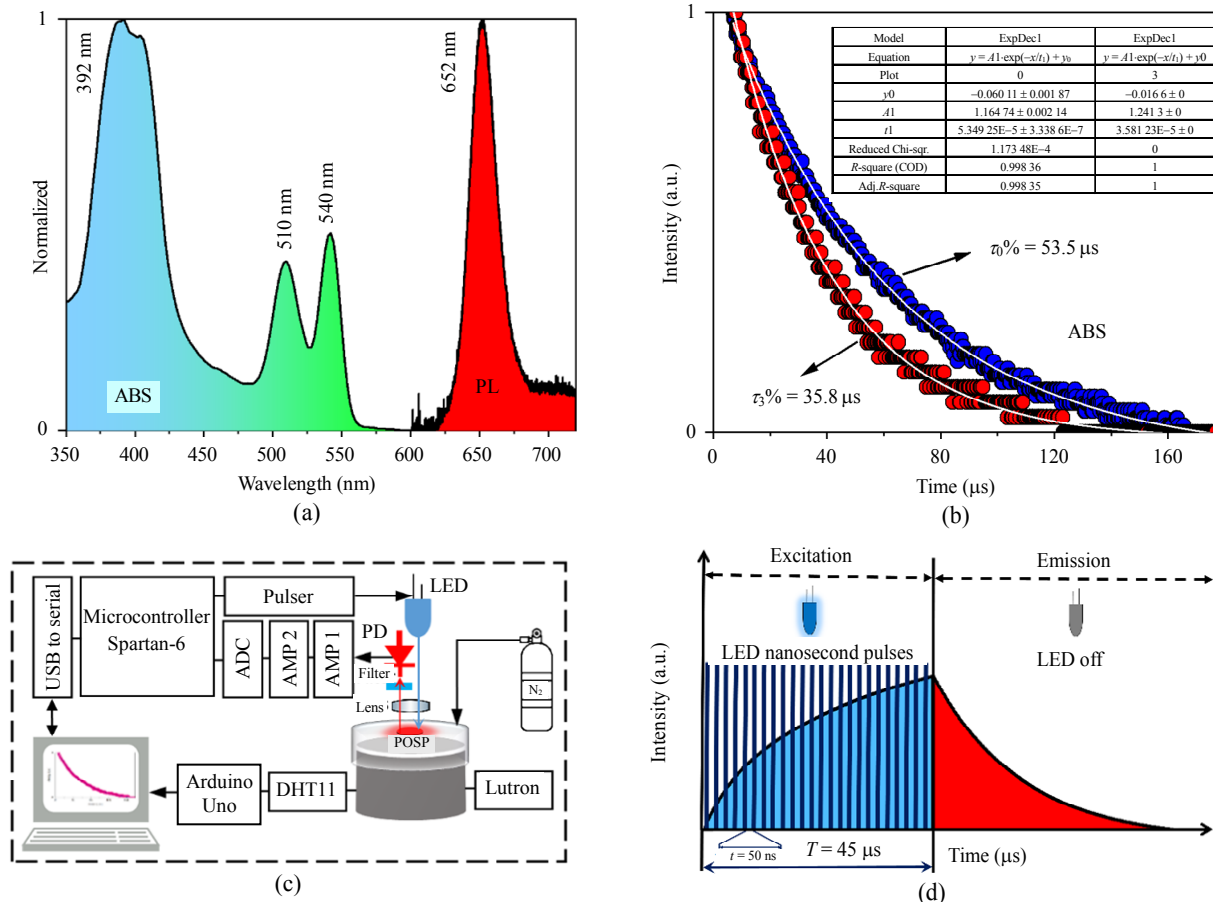


Fig. 2 Time-resolved phosphorescence spectroscopy for oxygen sensing: (a) absorption and PL spectra of the POSP, the absorption spectrum peaks are located at 392 nm, 510 nm, and 540 nm. The emission peak is at 652 nm, (b) time-resolved PL spectroscopy of the excited POSP at two different O<sub>2</sub> concentrations. Red represents the 3.0% and blue color the 0.0% O<sub>2</sub> concentration at (22.0 ± 0.1) °C, (c) measurement setup of O<sub>2</sub> concentration using the time-resolved PL spectroscopy, and (d) concept of the multipulse LED source for excitation of the POSP (ADC: analog to digital converter and AMP: amplifier).

In Fig. 3(b), the PL lifetime is plotted against time to evaluate the stability of the POSP. A PL stability of 99% was obtained, which shows that the POSP exhibited excellent lifetime stability for the caviar vacuum packaging quality control. The POSP's limit of detection was 0.2%. Next, the sensor reversibility, which is the ability to perform measurement multiple times, was obtained by calculating the relative standard deviation (RSD). Figure 3(c) shows the result of the reversibility test where the O<sub>2</sub> concentration varies several times between

0.0% and 5.3%. Each point represents an average value of the three consequent measurement times, and the temperature was maintained between 24.3 °C and 24.5 °C. The RSD values were 0.1% and 0.6% at the O<sub>2</sub> concentrations of 0% and 5.3%, respectively. The RSD value for 0% O<sub>2</sub> is lower than that for the 5.3% since at lower concentrations of O<sub>2</sub>, the emission intensity is higher, and therefore, the signal-to-noise ratio is higher. Thus, POSP exhibits the excellent reversibility. The high level of the reversibility is indicative of the reliability of the measurement. In addition, the PL lifetime was

measured for the temperature range of 10.0 °C – 35.0 °C to obtain its effect on the sensor’s work function. Figure 3(d) depicts the temperature-dependent lifetime variation, where each point represents an average value of three times

of consequent measurement. According to the figure, the lifetime decreases by increasing the temperature due to a higher level of collisional relaxation at higher temperatures, and the POSP exhibits a linear response with the slope of 0.2.

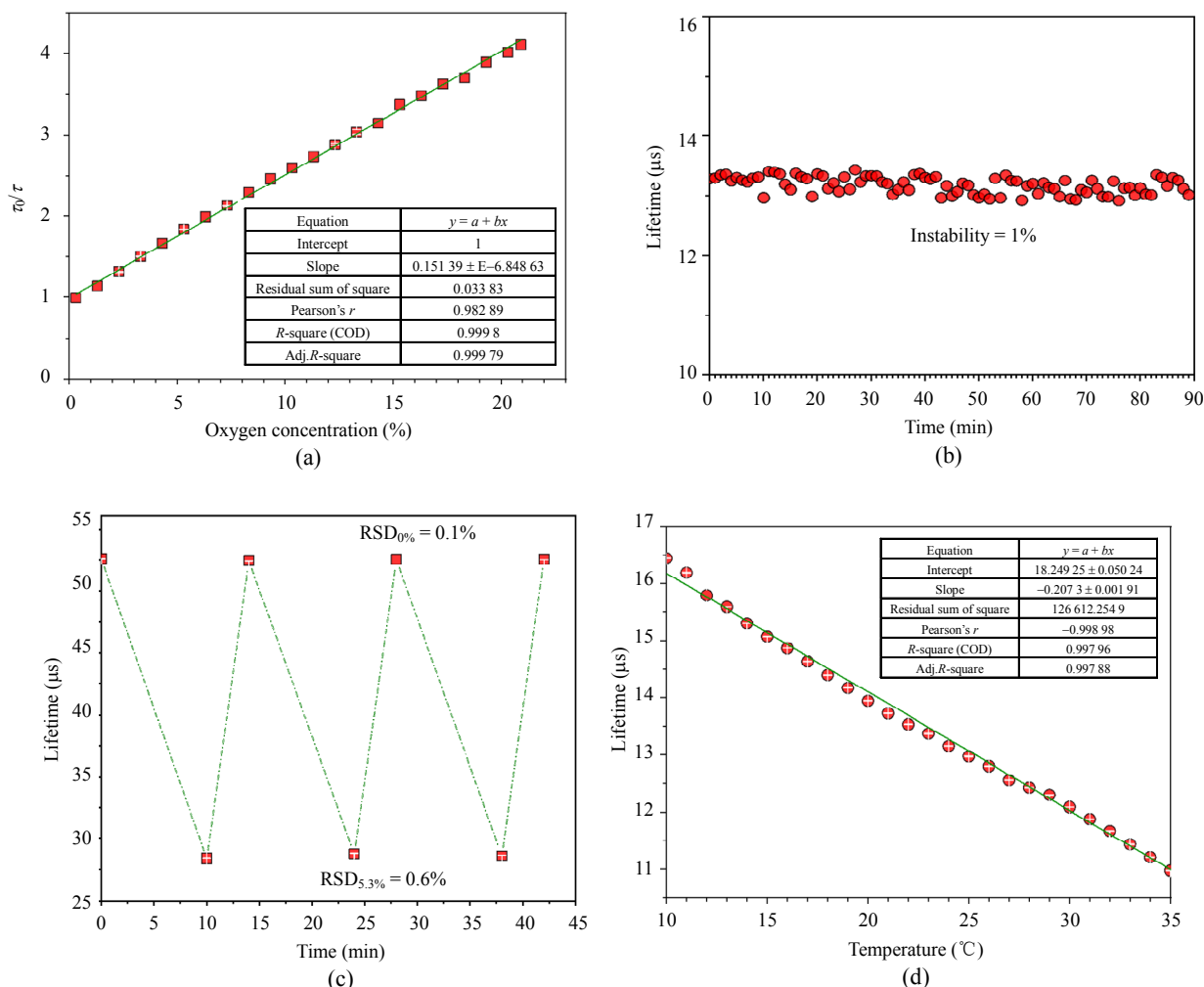


Fig. 3 Calibration and characterization of the POSP: (a) Stern-Volmer lifetime ratio versus O<sub>2</sub> concentration at (24.0 ± 0.1) °C, (b) stability of the POSP at 20.9% O<sub>2</sub> concentration, (c) repeatability of the POSP for modulation between 0.0% and 5.3% O<sub>2</sub> concentration, and (d) temperature-dependent lifetime variation.

Three identical POSPs were prepared to test the sensor-to-sensor repeatability. The O<sub>2</sub> concentration was varied from 1.0% to 20.9% to compare their responses. Figure 4(a) shows their responses where the red circles, green triangles, and blue squares correspond to samples 1, 2, and 3, respectively. According to the results, the overlap of these three POSPs represents their excellent repeatability. Figure 4(a) inset presents the three samples.

After obtaining the optical characterization of

the POSP, it was used for the quality control of caviar vacuum packaging. Thus, the POSP was laminated between a high-density polyethylene (HDPE) layer and a low-density polyethylene (LDPE) layer with a thickness of 100 μm to prevent PtTFPP from leaching into the caviar. The HDPE and LDPE were inside and outside layers correspondingly. LDPE is transparent, so it allows for the excitation of the POSP with an LED. HDPE exhibits humidity resistance, and hence, it was



placed with the direct contact with the caviar [45, 46]. Figure 4(b) depicts the setup drawn in Fig. 2(c) used for the Beluga caviar vacuum packaging control, the sandwiched POSP applied to the caviar packaging, and its corresponding structure.

Next, the effect of applying the polyethylene layers was investigated on the optical function of the POSP. First, the bare POSP was placed in direct contact with the caviar packaging atmosphere, and the O<sub>2</sub> level was modified from 20.9% to 0%. Next,

the POSP layer was embedded between the two layers of HDPE and LDPE, and a similar experiment in the caviar packaging was repeated. Figure 4(c) shows that there is approximately a 215-s time delay between the sandwiched POSP and the bare POSP to reach from 20.9% to 0% O<sub>2</sub> concentrations in the same condition. This implies that the polyethylene layers introduce a negligible delay in reducing the penetration rate of the O<sub>2</sub> molecules while protecting the caviar and POSP itself.

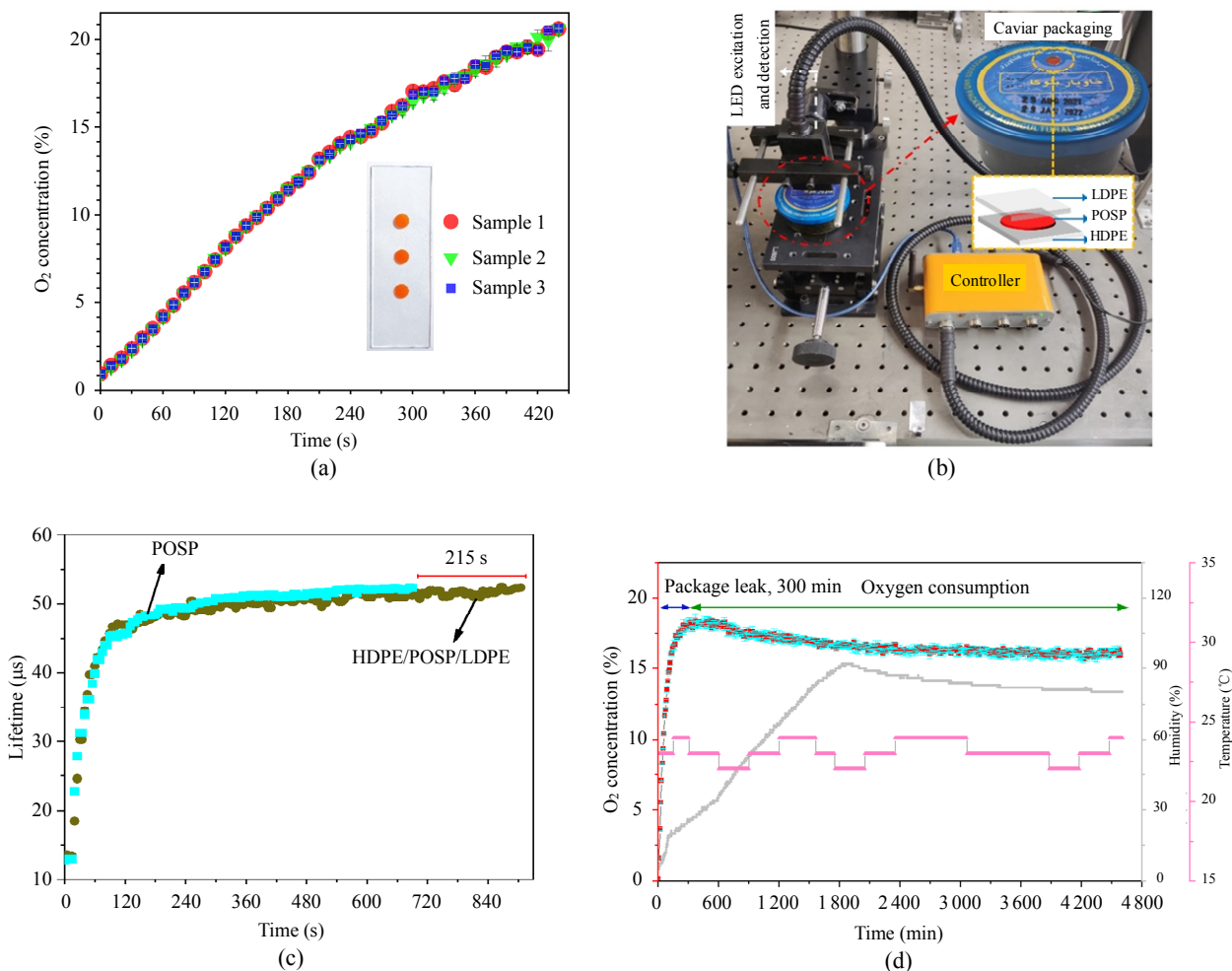


Fig. 4 Application of the POSP for contactless, non-intrusive, and real-time measuring the oxygen concentration in the caviar packaging: (a) sensor to sensor repeatability, (b) real measurement setup and the POSP structure includes LDPE, POSP, and HDPE layers, (c) effect of protective layers on the POSP in the caviar packaging: cyan squares correspond to the POSP alone, and the dark-green circles to HDPE/POSP/LDPE, and (d) real-time measurement of the oxygen concentration, humidity, and temperature in the caviar packaging: red squares correspond to the O<sub>2</sub> concentration, gray squares correspond to the humidity, the pink squares correspond to the temperature, and cyan lines correspond to error bars of measurement.

The long-term capability of the sensor was investigated by attaching the POSP into the gas- and temperature-controlled chamber containing 5 g of

caviar and a DHT11 temperature and humidity sensor for long-term monitoring, when the packaging suffered from leakage. The temperature



varied between 22.0 °C and 24.0 °C during the test. Initially, the O<sub>2</sub> concentration of the chamber reached 0% by injecting N<sub>2</sub> gas. Afterwards, the O<sub>2</sub> level inside the chamber gradually increased from 0% to 18.5% within 512 min, according to Fig. 4(d). The chamber was then sealed at 18.5% to demonstrate the capability of the POSP to detect minor changes in the O<sub>2</sub> level. The O<sub>2</sub> level was measured for the next 68 h with an 8-min time interval. Figure 4(d) shows the O<sub>2</sub> levels start to decrease due to the adsorption of O<sub>2</sub> by the caviar until it reaches approximately 15.9%. The suitable temperature for keeping caviar is between -3 °C and 3 °C, and heat accelerates the oxidation process of lipids and proteins [47]. Lopez *et al.* [48] reported that when the caviar was exposed to O<sub>2</sub>, the PUFAs content oxidized, and this process led to a taste bitter. Molds, yeasts, and aerobic bacteria need O<sub>2</sub> and a temperature of 35 °C, to survive, grow, and reproduce. The spoilage process begins at the top layer, which is most exposed to O<sub>2</sub>. The molds and yeasts grow exponentially, increasing the O<sub>2</sub> uptake into the atmosphere over time. The amount of active water and moisture content in the caviar is high; therefore, the molds and yeasts absorb the oxygen-containing moisture of the caviar during the growth and reproduction process. At the end of the spoilage process, the caviar grains almost dry out, and its texture changes, and the humidity inside the chamber reaches approximately 80% ± 5%. Thus, the O<sub>2</sub> POSPs can be used to evaluate the efficiency of large-scale vacuum packaging and quantify the quality control parameters for the O<sub>2</sub> content in these packages. Due to the nature of this thin layer, it can be easily included in food packaging.

It is essential to be confident that there is no leaching from the POSP to the package content, and therefore, inductively coupled plasma optical emission spectroscopy (ICP-OES) was employed [49, 50]. Thus, three samples were prepared and kept in a 3 mL aqueous medium for 120 h. The first sample contained control water, the second sample

contained sandwiched POSP soaked in water, and the third sample contained 2 mg PtTFPP powder mixed in water. The Spectro ARCOS system detected no trace of the PtTFPP in the first two samples with a limit of detection (LOD) of 0.02 ppm. A similar test was carried out after 24 h incubation of samples using Varian ICP-OES 730-ES with 0.01 ppm LOD, and again no trace of PtTFPP was detected for the first two samples. As a result, the polyethylene layers are reliable barriers in protecting the caviar from leaching the platinum complex.

#### 4. Conclusions

In this paper, a photonics-based oxygen sensor was developed for caviar vacuum packaging quality control by integration of a nanosecond LED light source, fast and amplified photodetector, FPGA microcontroller, and PtTFPP-PS POSP embedded in the oxygen-permeable HDPE and LDPE layers for real-time, precise, and non-contact detection. The time-resolved phosphorescence spectroscopy was used for lifetime measurement, which is more accurate than the conventional phase-detection approach. The pulsed LED light sources are preferred over laser light sources due to their low cost, accessibility, small size, and durability. Two layers of HDPE and LDPE were employed to avoid possible leaching of the PtTFPP, and through a series of experiments, no trace of PtTFPP was found. It was shown that these layers had a negligible impact on the sensor's response time while protecting the caviar and the sensor itself. It was demonstrated that the sensor system exhibited excellent sensitivity, stability, reversibility, and repeatability through systematic tests.

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## Declarations

**Conflict of Interest** The authors declare that they have no competing interests.

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