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# 快速光脉冲藻类光合作用测量方法的激发条件研究

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**摘要:** 激发光强和激发持续时间是快速光脉冲藻类光合作用测量方法的关键实验条件. 通过光脉冲激发实验, 定量分析了不同平均激发光强下还原态初级电子受体的比例和发生再氧化的初级电子受体的比例. 结果表明: 快速光脉冲激发的最佳平均光强为  $30\,000\ \mu\text{mol quanta} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ , 最佳激发持续时间为  $70\ \mu\text{s}$ ;  $30\,000\ \mu\text{mol quanta} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$  平均激发光强能够在  $70\ \mu\text{s}$  内还原  $96.08\%$  的初级电子受体, 且仅  $9.81\%$  的初级电子受体发生了再氧化.

**关键词:** 藻类; 光合作用; 荧光; 电子传递; 光学测量

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## Determining the Optimal Excitation Condition of High-frequency Flash Method for Algae Photosynthetic Parameters Measurement

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**Abstract:** The excitation intensity and excitation duration are the essential experimental conditions of the high-frequency flash method for the measurement of algae photosynthetic parameters. The fraction of reduced primary electron acceptors and the reoxidation extent of these reduced primary electron acceptors were analyzed under different excitation intensities and excitation durations. The analysis results indicate that  $30\,000\ \mu\text{mol quanta} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$  is the optimal excitation intensity and  $70\ \mu\text{s}$  is the optimal excitation duration. Under this optimal excitation condition,  $96.08\%$  of the primary electron acceptors are reduced and only  $9.81\%$  primary electron acceptors reoxidize.

**Key words:** Algae; Photosynthesis; Fluorescence; Electron transfer; Optical measurement

**OCIS Codes:** 170.1420; 280.4788; 120.0120; 300.6280

## 0 Introduction

The chlorophyll *a* fluorescence yield can be used to access the algae photosynthesis activity. Many methods have been developed based on this principle<sup>[1-5]</sup>, including Pump and Probe (P&P) method<sup>[6]</sup> and Pulse

Amplitude Modulation (PAM) method<sup>[7]</sup>. While a high-frequency flash method put forward by Kolber in the 1990's<sup>[8-9]</sup> is much more promising because it eliminates the influence of the reoxidation of reduced primary electron acceptors, improves the retrieval accuracy and realizes the measurement of  $\sigma_{\text{PSII}}$ . This

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method uses a sequence of high-frequency excitation flashes at microsecond intervals to reduce all the primary electron acceptors, inducing a microsecond-level fluorescence yield curve. The curve is then fitted by fluorescence yield model to calculate the PSII functional absorption cross section  $\sigma_{\text{PSII}}$  and PSII maximum quantum yield of photochemistry  $F_v/F_m^{[10]}$ .

The accuracy of the high-frequency flash method depends on whether the flashes can reduce all the primary electron acceptors with negligible reoxidation of reduced primary electron acceptors. Therefore, it's critical to choose appropriate excitation intensity and duration. This paper analyzes the fraction of reduced primary electron acceptors and the reoxidation extent of these reduced primary electron acceptors under different average excitation intensities, and determines the optimal excitation intensity and duration.

## 1 Experimental system

As shown in Fig. 1, the experimental system is composed of LED array and the driver unit, sample cell, detection unit for fluorescence signal and LED excitation light (using as reference signal), digital oscilloscope and computer. The excitation light from the LED array is modulated to flashes of 2  $\mu\text{s}$  duration at 1  $\mu\text{s}$  intervals using modulated signal that is generated by MCU ADuC841. The excitation intensity is controlled by changing the LED current. The induced fluorescence is collected by a PMT (photomultiplier) in 90° direction to the excitation light, and the excitation light is detected by a PIN photodiode. The two signals are sampled by a digital oscilloscope DLM2054 with a sample rate of 25 MS/s and transferred to a computer. The fluorescence curve is calculated, and consequently the photosynthetic parameters are retrieved on the computer.

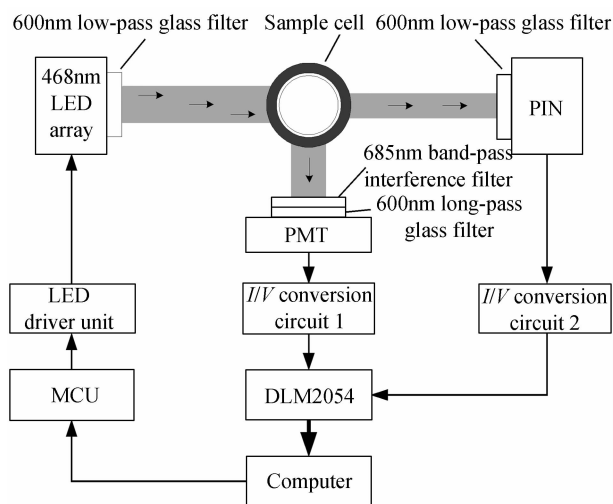


Fig. 1 Experimental system based on high-frequency flash method for photosynthetic parameter measurement

## 2 Method and analysis

The energy transfer process of algae photosynthesis is shown in Fig. 2. Following the flash excitation, the primary electron acceptor  $Q_A$  gets electron and is reduced to  $Q_A^-$ , leading to an increase in the fluorescence yield. However, in the flash intervals,  $Q_A^-$  transfers electron to  $Q_B$  and reoxidizes to

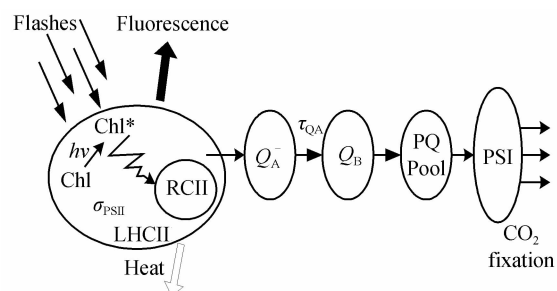


Fig. 2 Energy transfer diagram of algae photosynthesis  $Q_A$ , leading to a decrease of the fluorescence yield. The time constant of  $Q_A^-$  reoxidation  $\tau_{Q_A}$  is several hundreds of microseconds ( $100 \sim 300 \mu\text{s}$ )<sup>[11]</sup>, which is called single turnover period of  $Q_A$ . The induced fluorescence yield curve in the high-frequency flash method is described by Eq. 1, from which  $\sigma_{\text{PSII}}$  and  $F_v/F_m$  can be calculated.

$$F_i = F_o + (F_m - F_o)A_i = F_o + F_v \left[ 1 - \exp\left(-\sigma_{\text{PSII}} \sum_{j=1}^{i-1} l_j\right) \right] \quad (1)$$

where  $F_o$  is the minimal fluorescence yield measured when all PSII reaction centers are open,  $F_m$  is the maximum fluorescence yield measured when all PSII reaction centers are closed,  $F_v$  is variable fluorescence (i. e. the difference between the maximum and minimal fluorescence yield),  $A_i$  ( $0 \leq A_i \leq 1$ ) is the fraction of reduced primary electron acceptors at a given state of the excitation, and  $l_j$  is the energy of the j-th flash.

Eq. 1 is a fluorescence yield model that requires the  $Q_A^-$  reoxidation is negligible, and the fluorescence yield approaches  $F_m^{[8-9]}$ . The requirements of Eq. 1 can be met by reducing all the primary electron acceptors within the single turnover period, because there's few  $Q_A^-$  reoxidation in the single turnover period of  $Q_A$ . In this way, the fluorescence yield can approach  $F_m$  by reducing all the primary electron acceptors while the  $Q_A^-$  reoxidation is negligible.

The high-frequency method with appropriate excitation intensity and duration can reduce all the primary electron acceptors within the single turnover period of  $Q_A$ . The determination of excitation intensity and duration can be achieved by analyzing  $A_i$  and  $e_i$ , which are respectively the fraction of reduced primary electron acceptors and the reoxidation extent of these reduced primary electron acceptors.  $A_i$  and  $e_i$  can be

calculated from Eq. 2 and Eq. 3<sup>[10]</sup>. These two equations indicate that the values of  $l_i$ ,  $\sigma_{\text{PSII}}$  and  $\tau_{\text{QA}}$  should be determined before calculating  $A_i$  and  $e_i$ .  $l_i$  can be determined by the reference signal during the measurement, but extra experiments are required to determine  $\sigma_{\text{PSII}}$  and  $\tau_{\text{QA}}$ .

$$A_i = A_{i-1} \exp(-\Delta t / \tau_{\text{QA}}) + [1 - \exp(-\sigma_{\text{PSII}} l_i)] \cdot [1 - A_{i-1} \exp(-\Delta t / \tau_{\text{QA}})] = 1 - \exp(-\sigma_{\text{PSII}} l_i) \cdot [1 - A_{i-1} \exp(-\Delta t / \tau_{\text{QA}})] \quad (2)$$

$$e_i = \sum_{i=1}^i A_i \exp(-\Delta t / \tau_{\text{QA}}) \quad (3)$$

### 3 Experiments

The single turnover period of  $Q_A$  varies between 100 ~ 300  $\mu\text{s}$  according to the algae class and photosynthetic activity. To ensure all the primary electron acceptors are reduced within the single turnover period in any condition, we analyzed the changing process of  $A_i$  and  $e_i$  in 100  $\mu\text{s}$ . Based on the analysis results, the optimal excitation intensity and duration were determined. According to Eq. 2 and Eq. 3, the values of  $\sigma_{\text{PSII}}$  and  $\tau_{\text{QA}}$  are needed for calculations of  $A_i$  and  $e_i$ . Thus, we first designed experiments to determine  $\sigma_{\text{PSII}}$  and  $\tau_{\text{QA}}$ . Then excitation experiments were implemented for the determination of the optimal excitation intensity and duration. The algae sample used in all the experiments in this paper was *chlorella pyrenoidosa*, with a concentration of 80  $\mu\text{g/L}$ . The excitation wavelength was 468nm. The concentration of DCMU solution was 100  $\mu\text{mol/L}$ .

#### 3.1 $\sigma_{\text{PSII}}$ measurement

$\sigma_{\text{PSII}}$  can be calculated from Eq. 1 when the  $Q_A^-$  reoxidation is negligible. In the  $\sigma_{\text{PSII}}$  measurement experiment, we employed DCMU to inhibit the  $Q_A^-$  reoxidation<sup>[12-14]</sup> for the measurement of  $\sigma_{\text{PSII}}$ . 1 mL DCMU solution was added to 100 mL *chlorella pyrenoidosa* solution, and all the primary electron acceptors were reduced by excitation flashes with an average excitation intensity of 10 000  $\mu\text{mol quanta} \cdot$

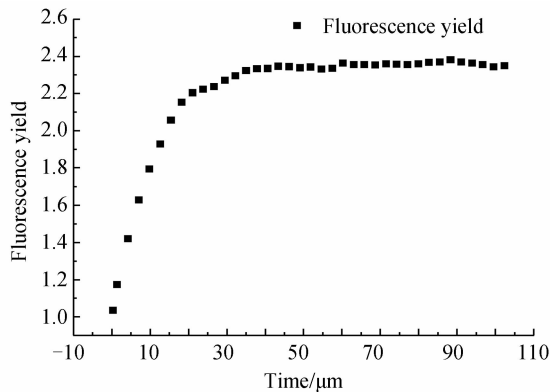


Fig. 3 Fluorescence yield curve measured with the presence of DCMU

$\text{m}^{-2} \cdot \text{s}^{-1}$ . As the  $Q_A^-$  reoxidation was inhibited by DCMU, there was no special requirement for the excitation intensity<sup>[15]</sup>. The induced fluorescence yield curve was shown in Fig. 3, and  $\sigma_{\text{PSII}}$  was calculated to be 310  $\text{\AA}^2$  by fitting the curve to Eq. 1.

#### 3.2 $\tau_{\text{QA}}$ measurement

A part of primary electron acceptors are reduced when healthy algae are excited by intense flashes, and consequently the fluorescence yield increases to  $F_i$ . When the excitation stops, the fluorescence yield begins to decrease because of the reoxidation of the reduced primary electron acceptors. The decay curve is described by Eq. 4<sup>[16-17]</sup>, and  $\tau_{\text{QA}}$  can be calculated from this curve.

$$F_i = F_o + (F - F_o) \exp(-t / \tau_{\text{QA}}) \quad (4)$$

In the  $\tau_{\text{QA}}$  measurement experiment, the algae sample was excited by high-frequency flashes of 2  $\mu\text{s}$  duration at 1  $\mu\text{s}$  intervals for 40  $\mu\text{s}$ , and the average intensity was 30 000  $\mu\text{mol quanta} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ . Then weak modulated measuring light was employed to record the decay curve of the fluorescence yield, as shown in Fig. 4. The measuring light was modulated pulses of 0.3  $\mu\text{s}$  duration at 60  $\mu\text{s}$  intervals. The average intensity of the measuring light was 3  $\mu\text{mol quanta} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ , which was so weak that the actinic effect was negligible<sup>[18]</sup>. Finally,  $\tau_{\text{QA}}$  was calculated to be 223  $\mu\text{s}$  by fitting the recorded decay curve to Eq. 4.

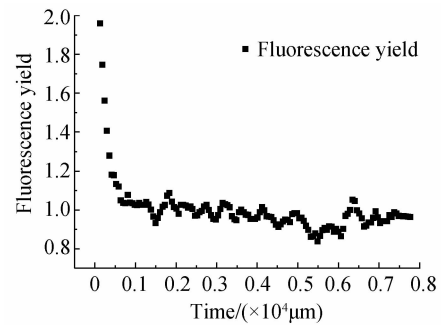


Fig. 4 Decay curve of fluorescence yield following intense excitation flashes

#### 3.3 Excitation experiment

We implemented excitation experiments using a series of excitation intensities to determine the optimal excitation intensity and duration. The average excitation intensities we employed were 5 000, 10 000, 15 000, 20 000 and 30 000  $\mu\text{mol quanta} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$ . The corresponding fluorescence yield curves are shown in Fig. 5.  $A_i$  curves calculated using Eq. 2 are shown in Fig. 6.

As shown in Fig. 5 and Fig. 6, the fluorescence yield curve and  $A_i$  curve increased more rapidly with higher excitation intensity. After 100  $\mu\text{s}$  of excitation, the fluorescence yield curves and  $A_i$  curves under average excitation intensities of 5 000 ~ 20 000  $\mu\text{mol}$

quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup> still had rising trend, and did not achieve the maximum. But when the average excitation intensity was 30 000  $\mu$ mol quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup>, both the fluorescence yield curve and  $A_i$  curve achieved the maximum at 100  $\mu$ s, and  $A_i$  reached up to 96.46%, which meant that almost all of the primary electron acceptors were reduced.

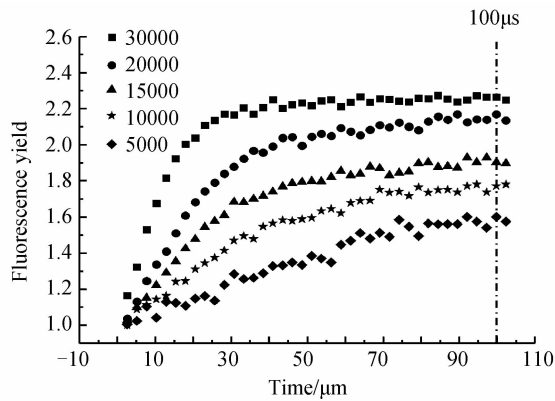


Fig. 5 Fluorescence yield curves under different average excitation intensities

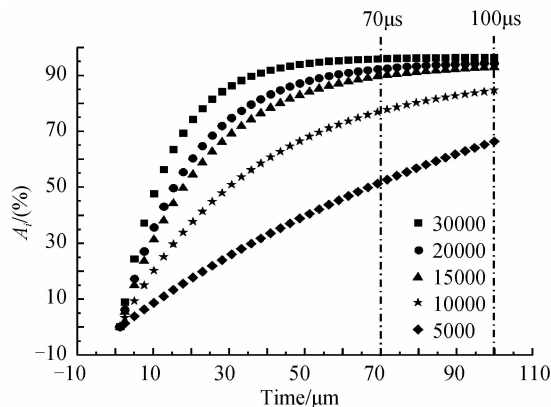


Fig. 6 Fractions of the reduced primary electron acceptors under different average excitation intensities

In order to determine the optimal excitation duration, we further analyzed the change trend of  $A_i$  and  $e_i$  under 30 000  $\mu$ mol quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup> average excitation intensity excitation, as shown in Fig. 7.  $A_i$  reached up to 96.08% at 70  $\mu$ s and increased to

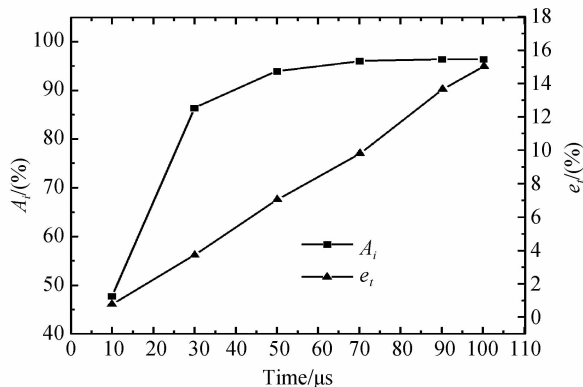


Fig. 7  $A_i$  and  $e_i$  versus time under an average excitation intensity of 30 000  $\mu$ mol quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup>

96.46% at 100  $\mu$ s, and the increment per microsecond was only 0.013%.  $e_i$  were 9.67% and 15.04% respectively at 70  $\mu$ s and 100  $\mu$ s, and the increment per microsecond was 0.17%, which was 13 times as high as that of  $A_i$ . The analysis showed that  $A_i$  nearly reached the maximum at 70  $\mu$ s and kept stable, but  $e_i$  increased rapidly from 70  $\mu$ s to 100  $\mu$ s. Thus the excitation should be stopped as soon as  $A_i$  nearly reached to the maximum at 70  $\mu$ s to avoid more  $Q_A^-$  reoxidation and reduction.

The above analysis of  $A_i$  and  $e_i$  indicated that when the average excitation intensity was 30 000  $\mu$ mol quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup>, almost all of the primary electron acceptors were reduced ( $A_i = 96.08\%$ ) within 70  $\mu$ s, while  $e_i$  was just 9.67%, which was negligible<sup>[10]</sup>. Therefore, we chose 30 000  $\mu$ mol quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup> as the optimal average excitation intensity, and 70  $\mu$ s as the optimal excitation duration. It is necessary to point out that all the primary electron acceptors can be reduced within a shorter time with higher excitation intensity, but higher excitation intensity would lead to non-photochemical fluorescence quenching caused by either carotenoid triplets formation<sup>[19]</sup> or P680<sup>+</sup> accumulation<sup>[20]</sup>, which would influence the retrieval accuracy.

## 4 Conclusion

In conclusion, we present a method to determine the optimal excitation intensity and duration of the high-frequency flash method for the measurement of algae photosynthetic parameters. The requirements of excitation intensity and duration are specified by analyzing the fluorescence yield model used in the high-frequency flash method. Then the optimal average excitation intensity and excitation duration are respectively determined to be 30 000  $\mu$ mol quanta  $\cdot$  m<sup>-2</sup>  $\cdot$  s<sup>-1</sup> and 70  $\mu$ s based on the analysis results of the fraction of reduced primary electron acceptors and the reoxidation extent of these reduced primary electron acceptors. Under this optimal excitation condition, 96.08% of the primary electron acceptors are reduced and only 9.81% primary electron acceptors reoxidize. For different optical structures and excitation protocols, the requirements of excitation intensity and duration may be different, but the analysis method in this paper is universal and can still be employed for the determination of excitation intensity and duration.

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