

Study on low-refractive-index sol-gel SiO₂ antireflective coatings

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A study on low-refractive-index SiO₂ antireflective (AR) coatings by a sol-gel method is reported. Variations in the properties of the coatings are related to the molar ratios of ammonia to deionized water being changed in the process of preparing the sols. From the performance test results, the optimal ratio of the reactants necessary to prepare low-refractive-index SiO₂ AR coatings is determined. Of all the SiO₂ AR coatings, the lowest recorded refractive index is 1.16 at a wavelength of 700 nm. The largest water contact angle is 121.2°, and the peak transmittance is 99.95% at a wavelength of 908 nm. Furthermore, the sol used to deposit the film with the lowest refractive index is stable because of the narrow size distribution of its constituent particles.

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The sol-gel method is a widely used technique in the preparation of materials with different morphologies. Thin films, powders, fibers, and other morphologies can be prepared by altering the processing conditions. The sol-gel method has been adopted widely and has seen rigorous development in the field of thin films. Sokolova *et al.* were the first to apply sol-gel coatings in the domain of high-power laser devices in 1967^[1]. After that, sol-gel chemical coatings developed quickly and became an important feature of high-power laser devices. SiO₂ antireflective (AR) coatings prepared by the sol-gel method have good optical properties and high laser-induced damage thresholds. There are many optical elements that require SiO₂ AR coatings in large high-power laser devices around the world^[2], such as the National Ignition Facility in the USA^[3], the Laser MégaJoules in France^[4,5], and the “Shengguang” systems in China^[6–8].

Compared with films of high-refractive-index media made by physical vapor deposition, the refractive index of SiO₂ AR coatings made by the sol-gel method are not only low but also controllable^[9,10]. According to the theory of optical film system design, double-layer, wide-bandwidth AR coatings can be achieved by depositing a high-refractive-index layer onto a substrate, followed by the deposition of a low-refractive-index thin film. It can effectively improve the light transmission efficiency of optical elements. In this Letter, SiO₂ AR coatings with low refractive indices were obtained by adjusting the molar ratios of ammonia and deionized water when the sols were prepared. The particle size distributions and viscosities of the sols were measured; sols with a narrow particle size distribution are stable and have a long lifetime. The lowest refractive index recorded was 1.16 at a 700 nm wavelength; the optical properties and hydrophobic performances of this coating are excellent. Thus, the results

could be beneficial in the development of double-layer, wide-bandwidth AR coatings.

The SiO₂ sols were prepared by the Stöber method using base catalysis of ammonia^[11]. The solutions, catalyzed with ammonia (NH₃), were mixed with the reactants tetraethoxysilane (TEOS), deionized water (H₂O), and ethanol (C₂H₅OH). The molar ratios of TEOS, H₂O, NH₃, and C₂H₅OH in the solutions were $n(\text{TEOS})/n(\text{H}_2\text{O})/n(\text{NH}_3)/n(\text{C}_2\text{H}_5\text{OH})=1:x:y:34.2$, ($x=2, 3, 4, 5$; $y=0.9, 1.0, 1.1$) and the preparation process of the SiO₂ sols is shown in Fig. 1. The chemical reagents were stirred vigorously for more than 5 h at 5°C because the hydrolysis and condensation reaction associated with sol gels are slower at reduced temperatures. After stirring for 2 h at 20°C, all the solutions were placed into an oven to age for 7 days at 60°C. Finally, the solutions needed to be refluxed to remove ammonia to obtain the stable SiO₂ sols. The coatings were deposited on substrates of JGS1 silica glass (Φ32 mm × 7 mm) and silica wafers (Φ51 mm × 0.3 mm) by dip coating with a withdrawal speed of 8 cm/min and then were baked at 180°C for 24 h.

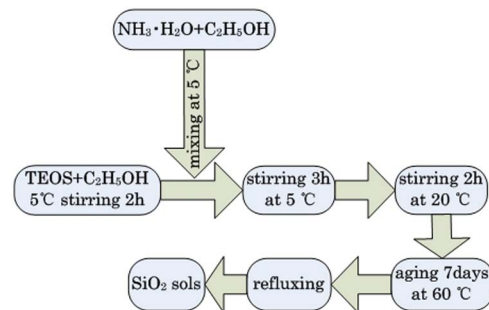


Fig. 1. SiO₂ sol preparation process.

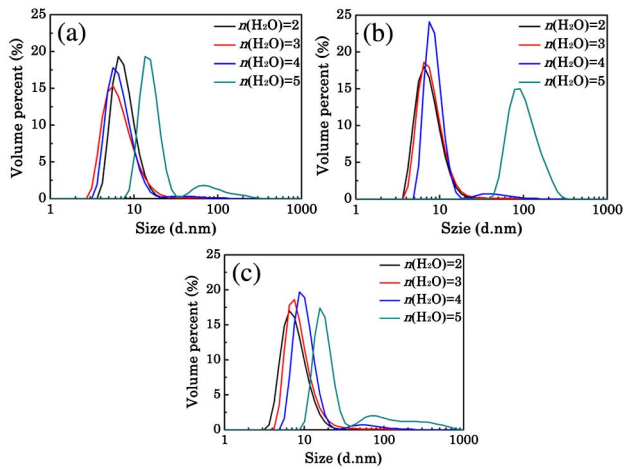


Fig. 2. Particle size distributions of sols. (a) $n(\text{NH}_3) = 0.9$; (b) $n(\text{NH}_3) = 1.0$; (c) $n(\text{NH}_3) = 1.1$.

Twelve sol samples were prepared using different molar ratios of NH_3 and H_2O , and the particle size distributions of the sols were measured by using a Malvern Nano ZS Mastersizer. It is important to judge the stability of sols from the particle size, which may influence the coating uniformity and transmittance. Figure 2 shows the particle size distributions of the sols. When the molar ratio of NH_3 is fixed, the size of the particles within the colloidal silica suspension become larger as the water content is increased. Under these conditions, the mean particle sizes within the sols are approximately 7, 8, and 10 nm for corresponding H_2O molar ratios of 2, 3, and 4, respectively, and the particle size distributions show only a single peak. However, when the molar ratio of H_2O is 5, there are particles with a much larger size in the colloidal silica suspension; therefore, the particle size distributions display double peaks. These sols with large particles are unlikely to be stable over a long duration. This phenomenon is mainly determined by the content of H_2O : when the water content is relatively low, the water can fully react with tetraethoxysilane, and homogeneous sol particles can be formed. However, with the increasing water content, in particular as the molar ratio approaches 5, there will be excess water in the solution, resulting in further reactions that form larger particles.

The refractive indices of SiO_2 AR coatings were measured using a Sopra GES-5 E Spectroscopic Ellipsometer at a wavelength of 700 nm. The film with the lowest refractive index, 1.16, originated from a sol with the following molar ratio of chemical reagents: $n(\text{TEOS})/n(\text{H}_2\text{O})/ n(\text{NH}_3)/n(\text{C}_2\text{H}_5\text{OH}) = 1:2:0.9:34.2$. The results in Fig. 3 show that when the molar ratios of NH_3 are identical, the refractive indices of coatings first increase and then decrease as the molar ratios of H_2O rise from 2 to 5. If Figs. 2 and 3 are analyzed together, the change in refractive indices is consistent with the change in particle sizes. This is because the refractive index of a coating is related to its porosity, and the size of the particles within the sol affects the porosity of the coating. For the sols with a mean

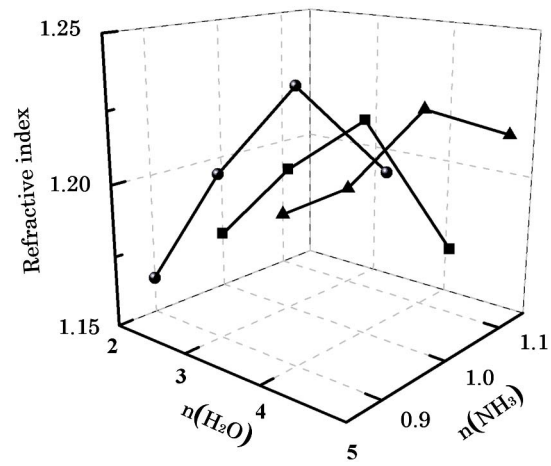


Fig. 3. Refractive indices of SiO_2 AR coatings.

distribution of particles, relatively small particles will occupy the pores that were shaped from the large-particle deposition during the film dip-coating process. The porosity of the coating will be reduced, and the refractive index will rise^[12]. Conversely, for the sol with an H_2O molar ratio of 5, most of the particles are large, so the small particles cannot completely occupy the pores, resulting in a decrease in the film refractive index.

The SiO_2 AR coatings prepared by the sol-gel method possess a porous structure; many polar OH groups exist on the surface of the films. The water contact angles of the SiO_2 AR coatings were measured using a DataPhysics OCA40 instrument in order to assess the hydrophobic properties of the films. This information can be used to determine the environment for which the films could be suitable. Figure 4 shows that films deposited from sols prepared using an H_2O molar ratio of 2 exhibited contact angles about 120° , regardless of NH_3 content, and therefore had excellent hydrophobic properties compared to the films of other H_2O molar ratios. As the H_2O molar ratio increased, the contact angles of the films diminished

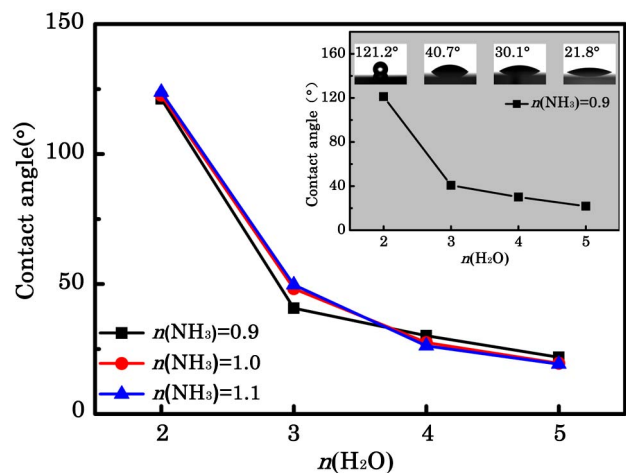


Fig. 4. Contact angles of the SiO_2 AR coatings deposited on JGS1 substrates.

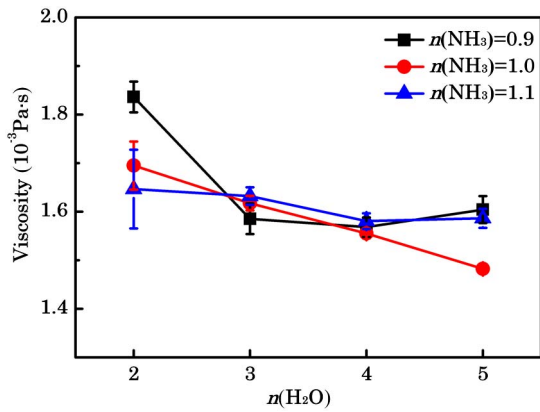


Fig. 5. Viscosities of the different sols.

Table 1. Transmittance Peaks of Different SiO₂ AR Coatings

$n(\text{H}_2\text{O})$	$n(\text{NH}_3) = 0.9$		$n(\text{NH}_3) = 1.0$		$n(\text{NH}_3) = 1.1$	
	T_{\max}	λ_{\max}	T_{\max}	λ_{\max}	T_{\max}	λ_{\max}
2	99.95	908	99.98	900	99.97	890
3	99.93	506	99.95	526	99.97	502
4	99.75	410	99.89	414	99.90	416
5	99.79	450	97.35	322	99.56	434

significantly down to a minimum value about 20°. This could have a deleterious effect on the quality of films when it is used in a humid environment. It can be explained that non-polar Si-O-Si groups were more prevalent than polar OH groups on the surface of films deposited using sols with an H₂O molar ratio of 2 after heating at 180°C. By contrast, sols with H₂O molar ratios of 3, 4, or 5 resulted in films with many polar OH groups in existence, even after the films were heated.

The optical spectra of SiO₂ AR coatings were acquired using a PerkinElmer Lambda 900 spectrophotometer, and the data are presented in Table 1. The transmittances of all the SiO₂ AR coatings are higher than 99.5% at the peak wavelengths (λ_{\max}) except for the film deposited using a sol prepared by the molar ratios $n(\text{TEOS})/n(\text{H}_2\text{O})/n(\text{NH}_3)/n(\text{C}_2\text{H}_5\text{OH}) = 1:5:1:34.2$, due to some absorption in the ultraviolet range. Whereas the peak wavelengths of coatings are different with the changing of the molar ratios of water, it can show the physical thickness of coatings estimated by $1/4\lambda_{\max}$ optical thickness. It

is for this reason that the physical thickness of films deposited by sol-gel dip coating can be predicted by the Landau-Levich-Derjaguin equation^[13]. The viscosity of the sol becomes the main factor determining the thickness of the coating when other parameters, including withdrawal speed and density, are similar. The viscosities of sols measured by a TA Ares-G2 rheometer are listed in Fig. 5 and are consistent with the physical thickness of film equation; the thickness of the film is proportional to the viscosity.

In conclusion, this study aims to prepare sol-gel SiO₂ AR coatings with low refractive indices, which can be used in high-power laser devices as well as to make double-layer, wide-bandwidth AR coatings. The experimental results clearly show that the ammonia catalyst content does not affect the properties of the sols and the corresponding coatings. Conversely, the water content has an important influence on the reaction processes and the properties of the resultant sols and coatings. The sol prepared using the molar ratios $n(\text{TEOS})/n(\text{H}_2\text{O})/n(\text{NH}_3)/n(\text{C}_2\text{H}_5\text{OH}) = 1:2:0.9:34.2$ result in films of the highest quality, in terms of the parameters tested (i.e., particle size, viscosity, refractive index, water contact angle, and transmittance), and thus, excellent low-refractive-index coatings can be realized.

References

1. R. S. Sokolova, V. A. Serebryakov, N. A. Razumovskaya, and V. E. Yashin, *Soviet J. Opt. Technol.* **44**, 562 (1977).
2. Q. Zhang, L. Zhou, W. Yang, H. Hui, J. Wang, and Q. Xu, *Chin. Opt. Lett.* **12**, 071601 (2014).
3. P. K. Whitman, S. C. Frieders, J. Fair, I. M. Thomas, R. Aboud, C. B. Thorsness, and A. K. Burnham, *Lawrence Livermore Nat. Lab.* **105821**, 163 (1999).
4. E. Lavastre, S. Fontaine, R. Bergez, P. Wender, P. Cormont, and C. Pellegrini, *Proc. SPIE* **7102**, 1068 (2008).
5. P. Prené, J. J. Priotton, L. Beaurain, and P. Belleville, *J. Sol-Gel Sci. Technol.* **19**, 533 (2000).
6. R. Liu, R. Zhan, Y. Tang, and J. Zhu, *Chin. Opt. Lett.* **4**, 119 (2006).
7. D. Zhao, L. Wan, Z. Lin, P. Shao, and J. Zhu, *High Power Laser Sci. Eng.* **3**, 1 (2015).
8. D. Hu, J. Dong, D. Xu, X. Huang, W. Zhou, X. Tian, D. Zhou, H. Guo, W. Zhong, X. Deng, Q. Zhu, and W. Zheng, *Chin. Opt. Lett.* **13**, 041406 (2015).
9. I. M. Thomas, *Appl. Opt.* **31**, 6145 (1992).
10. Z. Yang, D. Zhu, D. Lu, A. Zhang, M. Zhao, N. Ning, and Y. Liu, *Acta Opt. Sin.* **23**, 1366 (2003).
11. W. Stöber, A. Fink, and E. Bohn, *J. Colloid Interface Sci.* **26**, 62 (1968).
12. B. Shen, H. Li, H. Xiong, X. Zhang, and Y. Tang, *Chin. J. Lasers* **41**, 0906002 (2014).
13. C. Jing, X. Zhao, and H. Tao, *Surf. Coat. Technol.* **201**, 2655 (2006).